MATERIAL SAFETY DATA SHEET

Section 1 - Chemical Product and Company Identification

MSDS Name: Oxalic acid dihydrate

Catalog AC129600000, AC129600010, AC388100000, AC423150000, AC423150010,

Numbers: AC423150050 AC423150050, S80113, A218-3, A218-500, A219-250, A219-3, A219-50,

A219-500, A219J500, NC9667432

Synonyms: Ethanedioic acid dihydrate.

Company Identification: Fisher Scientific

One Reagent Lane Fair Lawn, NJ 07410

For information in the US, call: 201-796-7100
Emergency Number US: 201-796-7100
CHEMTREC Phone Number. US: 800-424-9300

Section 2 - Composition, Information on Ingredients

CAS#: 6153-56-6

Chemical Name: Oxalic acid dihydrate

%: >99
EINECS#: unlisted

Hazard Symbols: Risk Phrases:

Text for R-phrases: see Section 16

Hazard Symbols: C



Risk Phrases: 20/21/22 35 63

Section 3 - Hazards Identification

EMERGENCY OVERVIEW

Danger! May cause kidney damage. Harmful if swallowed, inhaled, or absorbed through the skin. Causes burns by all exposure routes. Possible risk of harm to the unborn child. Target Organs: Kidneys, heart, eyes, skin, brain, nerves, mucous membranes.

Potential Health Effects

Eye: Causes eye burns. May result in corneal injury. Causes redness and pain.

Skin: Harmful if absorbed through the skin. Causes severe skin irritation and possible burns. Rare

chemical burns may occur from oxalic acid and may cause hypocalcemia. Gangrene has occurred in the hands of people working with oxalic acid solutions without rubber gloves. The skin lesions are characterized by cracking of the skin and the development of slow-healing ulcers. The skin may be bluish in color, and the nails brittle and yellow.

Ingestion: Causes gastrointestinal tract burns. Oxalic acid is toxic because of its acidic and chelating

properties. It is especially toxic when ingested. As little as 5 grams (71 mg/kg) may be fatal. Ulcerations of the mouth, vomiting of blood, and rapid appearance of shock, convulsions, twitching, tetany, and cardiovascular collapse may occur following ingestion of oxalic acid or its soluble salts. Oxalic acid can bind calcium to form calcium oxalate which is insoluble at physiological pH. Calcium oxalate thus formed might precipitate in the kidney tubules and the brain. Hypocalcemia secondary to calcium oxalate formation might disturb the function of

the heart and nerves.

Inhalation: Causes chemical burns to the respiratory tract. Inhalation of oxalic acid dust or vapor

produces irritation of the respiratory tract, protein in the urine, nosebleed, ulceration of the mucous membranes, headache, nervousness, cough, vomiting, emaciation, back pain (due

to kidney injury), and weakness.

Chronic: Inhalation of oxalic acid dust or mist over a long period of time might result in weight loss

and respiratory tract inflammation. Rats administered oxalic acid at 2.5 and 5% in the diet for 70 days developed depressed thyroid function and weight loss. A study of railroad car cleaners in Norway who were heavily exposed to oxalic acid solutions and vapors revealed a 53% prevalence of urolithiasis (the formation of urinary stones), compared to a rate of 12%

among unexposed workers from the same company.

Section 4 - First Aid Measures

Eyes: In case of contact, immediately flush eyes with plenty of water for at least 15 minutes.

Get medical aid immediately.

Skin: In case of contact, immediately flush skin with plenty of water for at least 15 minutes

while removing contaminated clothing and shoes. Get medical aid immediately. Wash

clothing before reuse.

Ingestion: If swallowed, do NOT induce vomiting. Get medical aid immediately. If victim is fully

conscious, give a cupful of water. Never give anything by mouth to an unconscious

erson.

Inhalation: If inhaled, remove to fresh air. If not breathing, give artificial respiration. If breathing is

difficult, give oxygen. Get medical aid.

Notes to Treat s Physician:

Treat symptomatically and supportively.

Antidote:

Intravenous administration of calcium gluconate or calcium chloride may be required if

hypocalcemia or hypocalcemic tetany occur.

Section 5 - Fire Fighting Measures

General As in any fire, wear a self-contained breathing apparatus in pressure-demand, **Information:** MSHA/NIOSH (approved or equivalent), and full protective gear. During a fire,

irritating and highly toxic gases may be generated by thermal decomposition or

combustion. Use water spray to keep fire-exposed containers cool.

Extinguishing

Use water spray, dry chemical, carbon dioxide, or alcohol-resistant foam.

Media:

Autoignition Not applicable.

Temperature:

Flash Point: Not applicable. Explosion Not available

Limits: Lower:

Explosion Not available

Limits: Upper:

NFPA Rating: health: 3; flammability: 1; instability: 0;

Section 6 - Accidental Release Measures

General Information:

Use proper personal protective equipment as indicated in Section 8.

Spills/Leaks:

Vacuum or sweep up material and place into a suitable disposal container. Avoid runoff into storm sewers and ditches which lead to waterways. Clean up spills immediately.

observing precautions in the Protective Equipment section. Avoid generating dusty

conditions. Provide ventilation.

Section 7 - Handling and Storage

Handling: Wash thoroughly after handling. Minimize dust generation and accumulation. Do not get in eyes, on skin, or on clothing. Do not ingest or inhale. Discard contaminated shoes. Use only with adequate ventilation.

Storage: Keep container closed when not in use. Store in a cool, dry, well-ventilated area away from incompatible substances.

Section 8 - Exposure Controls, Personal Protection

Engineering Controls:

Facilities storing or utilizing this material should be equipped with an eyewash facility and a safety shower. Use adequate general or local exhaust ventilation to keep airborne concentrations below the permissible exposure limits.

+			
Chemical Name Final PELs	·	NIOSH	OSHA -
			-
Oxalic acid, anhydr 1 TWA	. mg/m3; 2 mg/m3	1 mg/m3 TWA 500	1 mg/m3
ous	STEL	mg/m3 IDLH	I
			-
Oxalic acid dihydra n listed	none listed	none listed	none
te		I	l
++		+	-

OSHA Vacated PELs: Oxalic acid, anhydrous: 1 mg/m3 TWA Oxalic acid dihydrate: None listed

Personal Protective Equipment

Eyes: Wear appropriate protective eyeglasses or chemical safety goggles as described by

OSHA's eye and face protection regulations in 29 CFR 1910.133 or European Standard

EN166.

Skin: Wear appropriate protective gloves to prevent skin exposure. **Clothing:** Wear appropriate protective clothing to prevent skin exposure.

Respirators: Follow the OSHA respirator regulations found in 29 CFR 1910.134 or European Standard

EN 149. Use a NIOSH/MSHA or European Standard EN 149 approved respirator if exposure limits are exceeded or if irritation or other symptoms are experienced.

Section 9 - Physical and Chemical Properties

Physical State: Powder
Color: white
Odor: odorless
pH: 1.3 (0.1M soln)

Vapor Pressure: .92 mm Hg @ 60 deg C

Vapor Density: 4.62

Evaporation Rate: Not available
Viscosity: Not available
Boiling Point: Not applicable.
Freezing/Melting Point: 101 deg C sublimes

Decomposition Temperature: Not available

Solubility in water: Moderately Soluble. 1g/7ml

Specific Gravity/Density: 1.653 @ 18.5°C Molecular Formula: C2H2O4.2H2O

Molecular Weight: 126.04

Section 10 - Stability and Reactivity

Chemical Stability: Stable under normal temperatures and pressures.

Conditions to Avoid: Dust generation, excess heat.

Incompatibilities with Other Strong oxidizing agents, mercury, hypochlorite, silver, strong alkalies,

chlorites, furfuryl alcohol.

Hazardous Decomposition Carbon monoxide, carbon dioxide, formic acid.

Products

Materials

Hazardous Polymerization Has not been reported.

Section 11 - Toxicological Information

RTECS#: CAS# 144-62-7: RO2450000

CAS# 6153-56-6: None listed

LD50/LC50: RTECS:

CAS# 144-62-7: Draize test, rabbit, eye: 250 ug/24H Severe;

Draize test, rabbit, skin: 500 mg/24H Mild;

Oral, rat: LD50 = 7500 mg/kg;

RTECS:

CAS# 6153-56-6:.

Other:

Carcinogenicity: Oxalic acid, anhydrous - Not listed as a carcinogen by ACGIH, IARC, NTP, or CA

Prop 65.

Oxalic acid dihydrate - Not listed as a carcinogen by ACGIH, IARC, NTP, or CA Prop

65.

Epidemiology: A study of railroad car cleaners in Norway who were heavily exposed to oxalic acid

solutions and vapors revealed a 53% prevalence of urolithiasis (the formation of urinary stones), compared to a rate of 12% among unexposed workers from the same

company.

Teratogenicity: No information found

Reproductive: Oxalic acid caused kidney damage in fetal sheep and rats and disturbed the estrus

cycle in rats. Increased sperm abnormalities were seen in the second generation of

mice administered 0.2% oxalic acid in the drinking water.

Neurotoxicity: No information found Mutagenicity: No information found

Other: Not available

Section 12 - Ecological Information

Ecotoxicity: Fish: Bluegill/Sunfish: LC50 = 4000 mg/L; 24 Hr.; Static Conditions

Fish: Mosquito Fish: LC50 = 1350 mg/L; 24 Hr.; Static Conditions

Section 13 - Disposal Considerations

Chemical waste generators must determine whether a discarded chemical is classified as a hazardous waste. US EPA guidelines for the classification determination are listed in 40 CFR Parts 261.3. Additionally, waste generators must consult state and local hazardous waste regulations to ensure complete and accurate classification. RCRA P-Series: None listed. RCRA U-Series: None listed.

Section 14 - Transport Information

US DOT

Shipping Name: CORROSIVE SOLIDS, TOXIC, N.O.S.

Hazard Class: 8 UN Number: UN2923 Packing Group: III Canada TDG

Shipping Name: CORROSIVE SOLIDS, TOXIC, N.O.S.

Hazard Class: 8 UN Number: UN2923 Packing Group: III

Section 15 - Regulatory Information

US Federal

TSCA

CAS# 144-62-7 is listed on the TSCA Inventory.

CAS# 6153-56-6 is not on the TSCA Inventory because it is a hydrate. It is considered to be listed if the CAS number for the anhydrous form in on the Inventory (40CFR720.3(u)(2)).

Health & Safety

None of the chemicals are on the Health & Safety Reporting List.

Reporting List **Chemical Test Rules**

None of the chemicals in this product are under a Chemical Test Rule.

Section 12b

None of the chemicals are listed under TSCA Section 12b.

TSCA Significant New

None of the chemicals in this material have a SNUR under TSCA.

Use Rule

None of the chemicals in this material have an RQ.

CERCLA Hazardous Substances and corresponding RQs

SARA Section 302 **Extremely Hazardous**

None of the chemicals in this product have a TPQ.

Substances **SARA Codes**

CAS # 144-62-7: acute, chronic. CAS # 6153-56-6: acute, chronic.

Section 313

No chemicals are reportable under Section 313.

Clean Air Act:

This material does not contain any hazardous air pollutants. This material

does not contain any Class 1 Ozone depletors. This material does not

contain any Class 2 Ozone depletors.

Clean Water Act:

None of the chemicals in this product are listed as Hazardous Substances under the CWA. None of the chemicals in this product are listed as Priority Pollutants under the CWA. None of the chemicals in this product are listed

as Toxic Pollutants under the CWA.

OSHA:

STATE Oxalic acid, anhydrous can be found on the following state right to know

lists: California, New Jersey, Pennsylvania, Minnesota, Massachusetts. Oxalic acid dihydrate can be found on the following state right to know lists:

Pennsylvania.

California Prop 65

California No Significant None of the chemicals in this product are listed.

Risk Level:

European/International Regulations

European Labeling in Accordance with EC Directives

Hazard Symbols: C

Risk Phrases:

R 20/21/22 Harmful by inhalation, in contact with skin and if swallowed.

R 35 Causes severe burns.

R 63 Possible risk of harm to the unborn child.

Safety Phrases:

S 36/37/39 Wear suitable protective clothing, gloves and eye/face protection.

S 45 In case of accident or if you feel unwell, seek medical advice immediately (show the label where possible).

WGK (Water Danger/Protection)

CAS# 144-62-7: 1 CAS# 6153-56-6: 1

Canada

CAS# 144-62-7 is listed on Canada's DSL List Canadian WHMIS Classifications: E, D1B

This product has been classified in accordance with the hazard criteria of the Controlled Products Regulations and the MSDS contains all of the information required by those regulations.

CAS# 144-62-7 is listed on Canada's Ingredient Disclosure List CAS# 6153-56-6 is not listed on Canada's Ingredient Disclosure List.

Section 16 - Other Information

MSDS Creation Date: 12/12/1997 **Revision #13 Date** 9/29/2005

The information above is believed to be accurate and represents the best information currently available to us. However, we make no warranty of merchantibility or any other warranty, express or implied, with respect to such information, and we assume no liability resulting from its use. Users should make their own investigations to determine the suitability of the information for their particular purposes. In no event shall the company be liable for any claims, losses, or damages of any third party or for lost profits or any special, indirect, incidental, consequential, or exemplary damages howsoever arising, even if the company has been advised of the possibility of such damages.