



UNIVERSIDAD NACIONAL
AUTÓNOMA DE MÉXICO
PROGRAMA DE DOCTORADO EN INGENIERÍA
FACULTAD DE QUÍMICA



*Reversible Addition-Fragmentation chain
Transfer (RAFT) polymerization in
supercritical Carbon Dioxide (scCO₂)*

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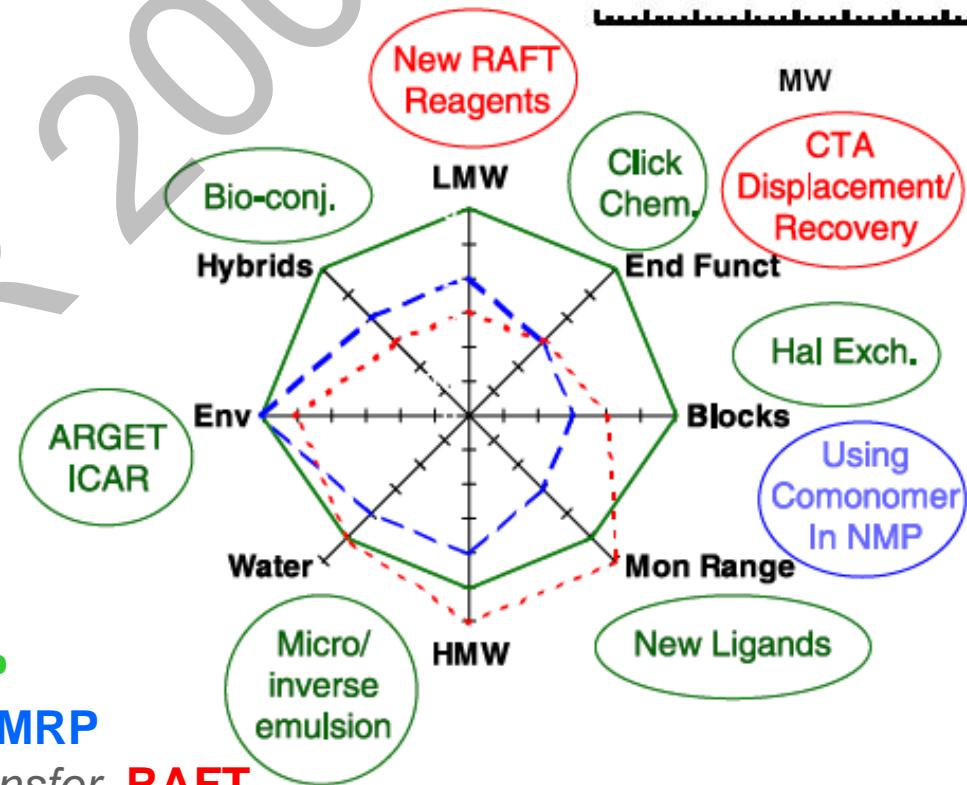
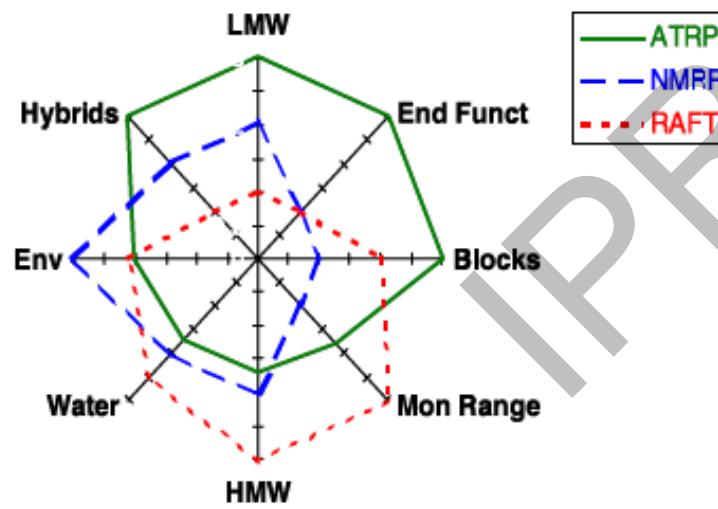
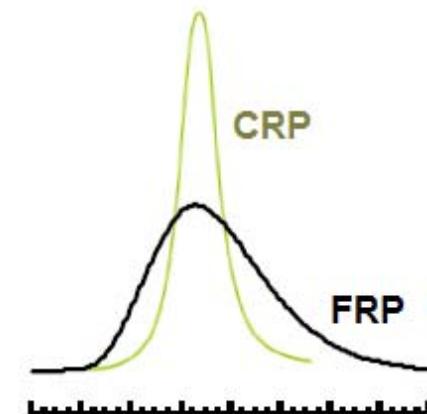
INSTITUTE FOR POLYMER RESEARCH
University of Waterloo, Waterloo, Ontario. Friday, May 1, 2009

OUTLINE

- Controlled Radical Polymerization
- RAFT Polymerization
- Why polymerization in scCO₂?
- State of the Art
- Model and Simulation
- Experimental Results
- Conclusions

CRP

- Living polymer chains with active end groups
- Predetermined MW and narrow MWD.
- Applications include adhesives, coatings, electronics, nano-technology, and biomaterials.



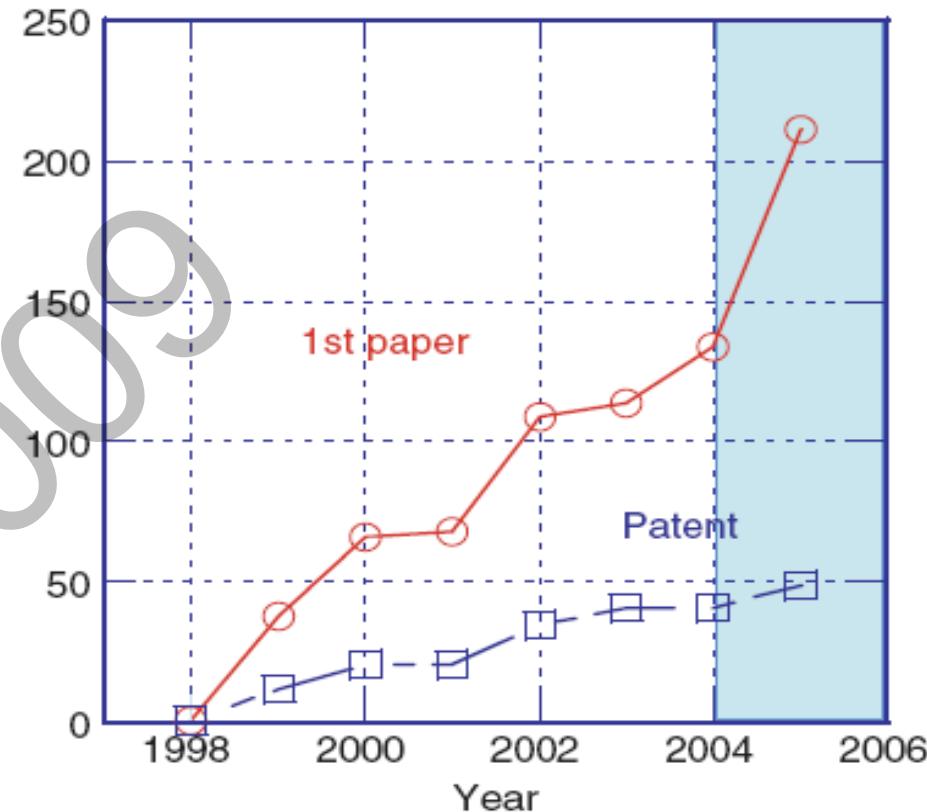
Atom Transfer Radical Polymerization, **ATRP**

Nitroxide Mediated Radical Polymerization, **NMRP**

Reversible Addition Fragmentation chain Transfer, **RAFT**

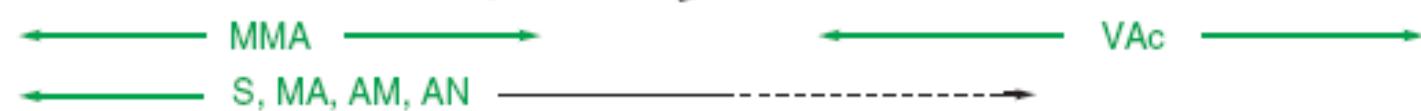
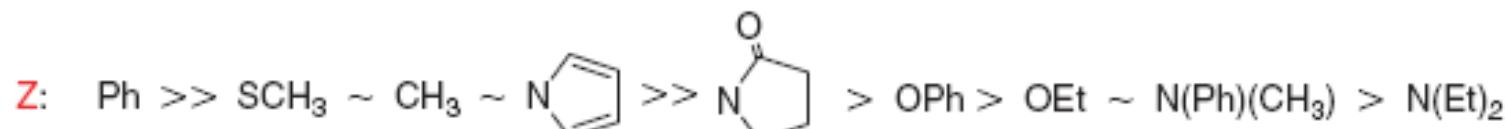
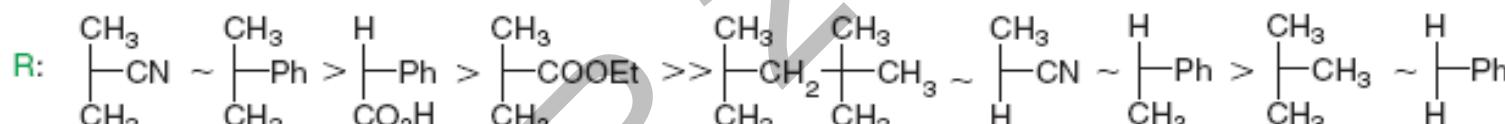
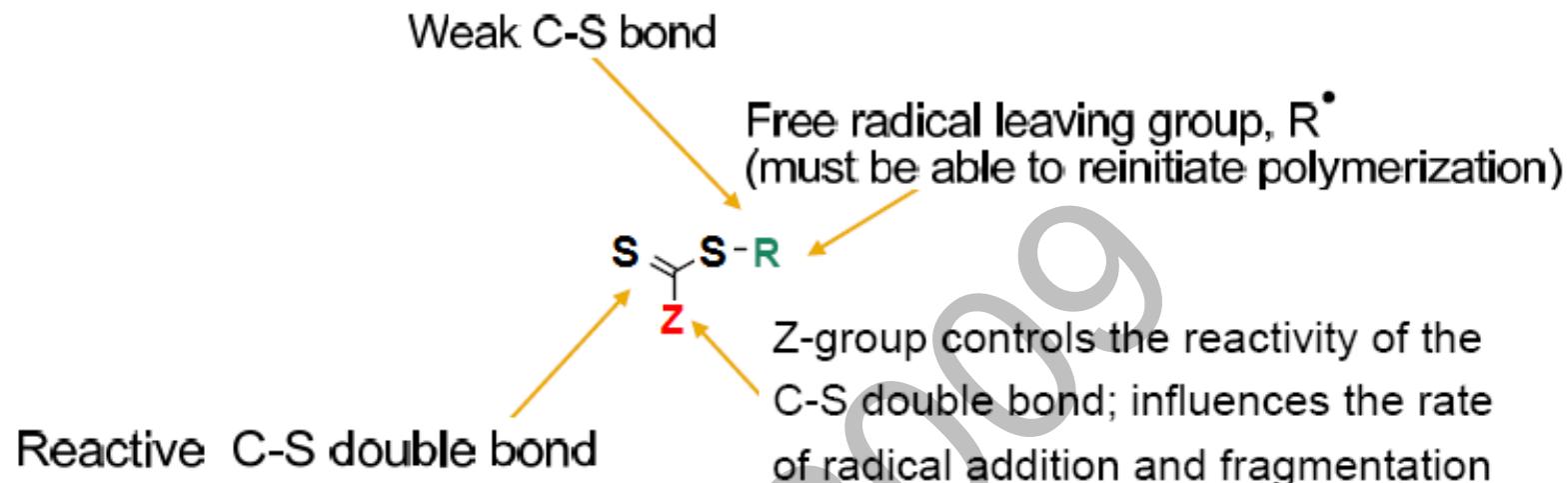
RAFT

- First communication: *J. Chiefari, Y. K. Chong, F. Ercole, J. Krstina, J. Jeffery, T. P.T. Le, R. T. A. Mayadunne, G. F. Meijls, C. L. Moad, G. Moad, E. Rizzardo, S. H. Thang, Macromolecules **1998**, 31, 5559.*
- First patent: *T. P. Le, G.Moad,E. Rizzardo, S. H.Thang, Int.Pat.WO9801478 **1998** [Chem. Abs. **1998**, 128, 115390f].*



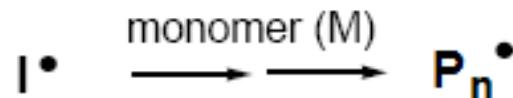
- ✓ Applicable to a large range of monomers
- ✓ Polymeric materials with controlled structure
- ✓ Success under a wide range of reaction conditions
- ✓ Wide variety of RAFT agents structures

RAFT

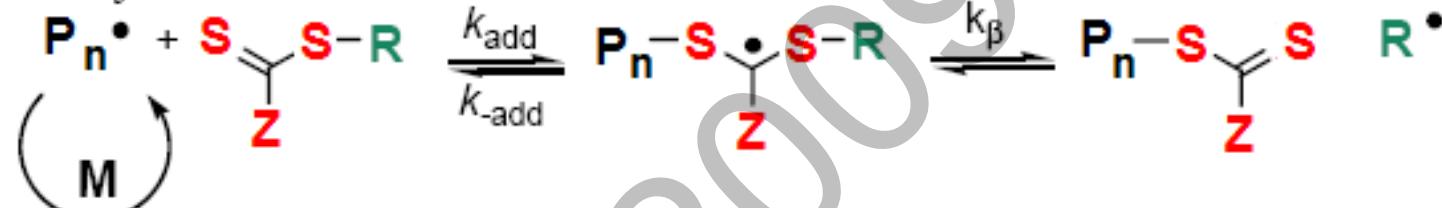


RAFT

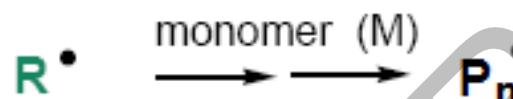
initiation



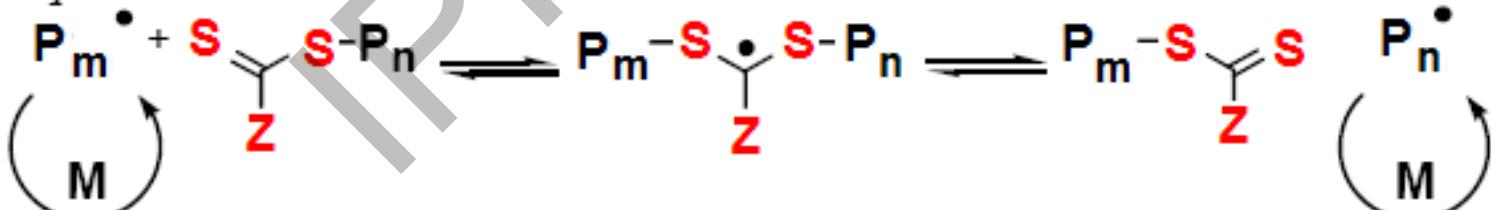
chain transfer



reinitiation



chain equilibration

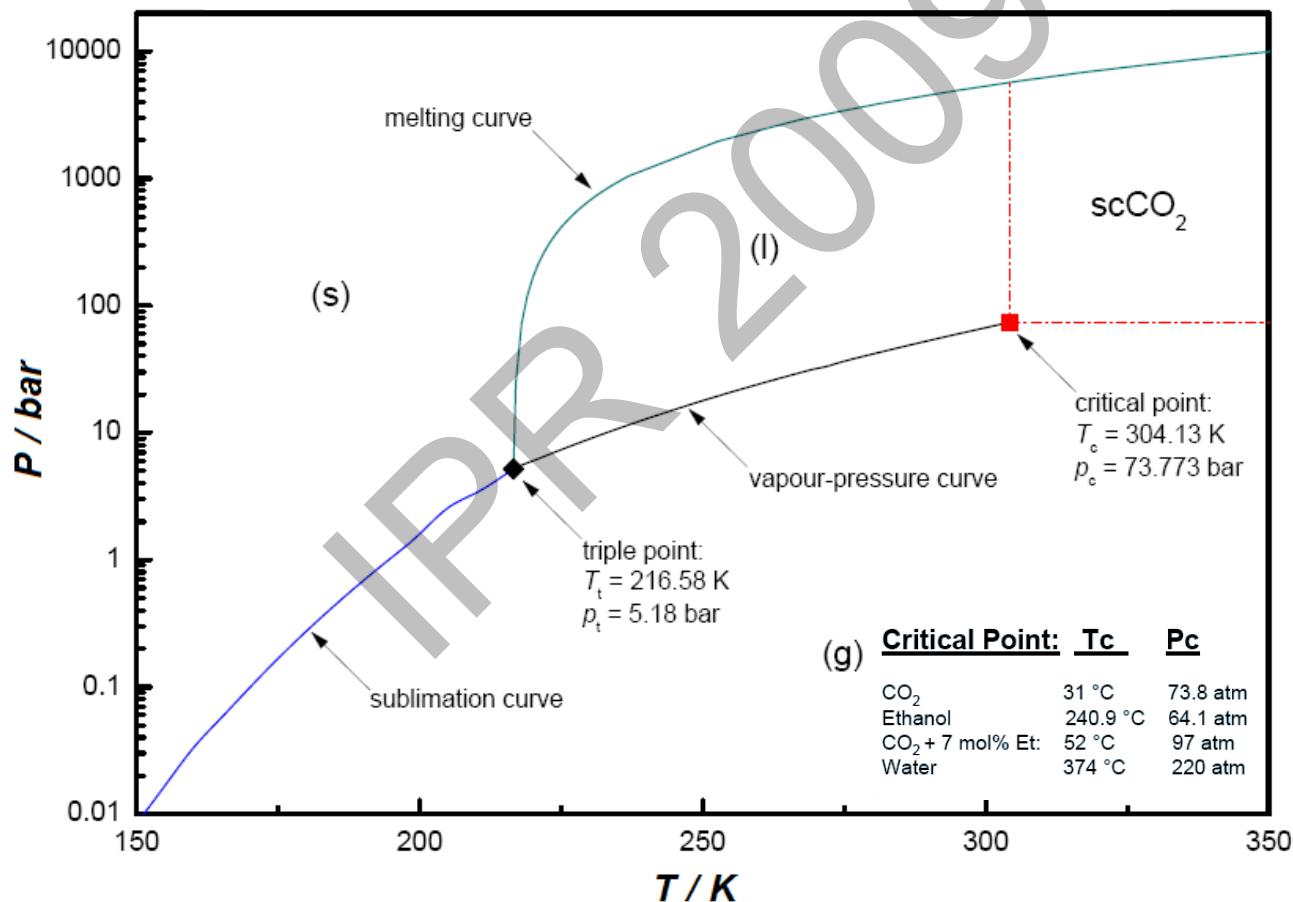


termination



scCO₂

	Liquid	Supercritical	Gas
ρ [kg m ⁻³]	1000	100-800	1*
η [Pa S]	10^{-3}	10^{-5} - 10^{-4}	10^{-5}
D [m ² s ⁻¹]	10^{-9}	10^{-8}	10^{-5}



*NIST Chemistry Webbook, NIST Standard

Reference Database 69, National Institute of Standards and Technology, Gaithersburg MD,
<http://webbook.nist.gov>

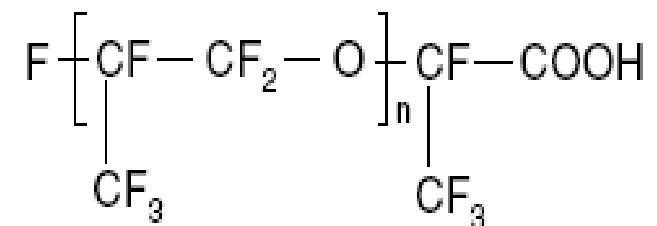
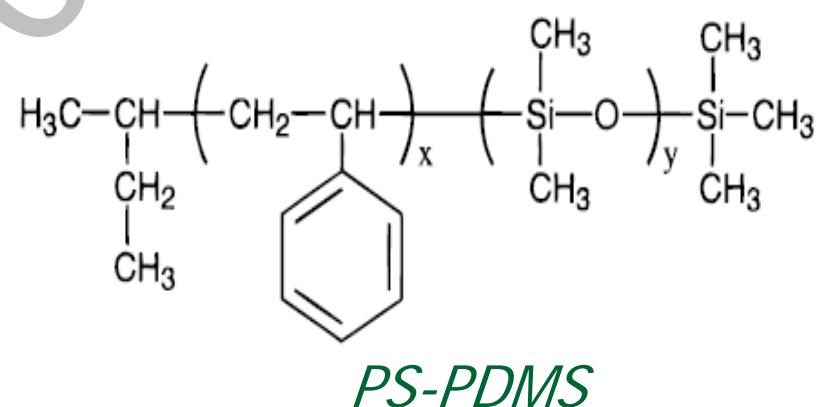
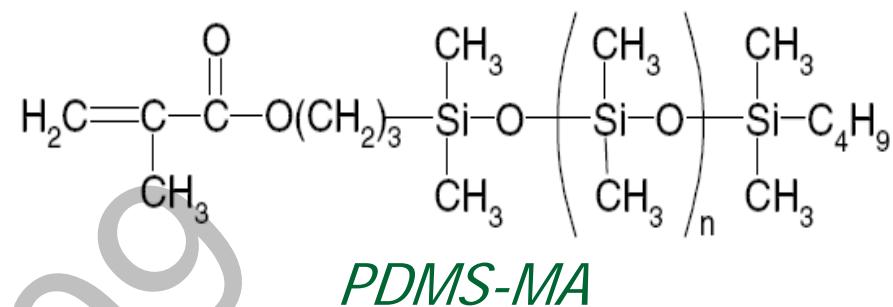
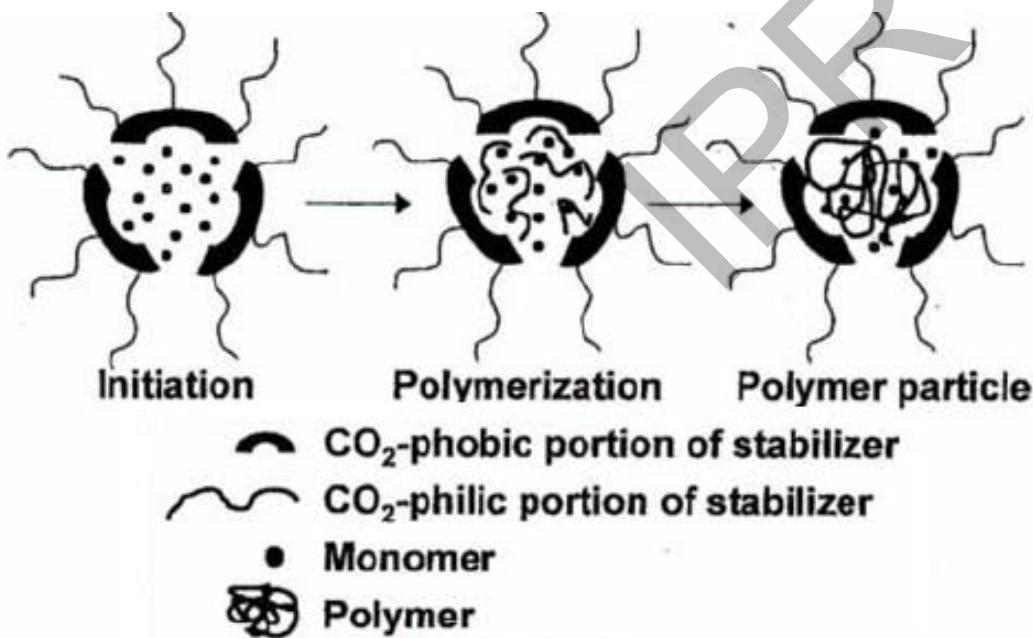
scCO₂

- ✓ High purity, low toxicity, low cost.
 - ✓ Controllable dissolving power.
 - ✓ Inert solvent.
 - ✓ Low viscosity favors mass transportation.
 - ✓ Advantages in the operation.
-
- Elevated pressures required.
 - Compression costs.
 - High capital equipment investment.

RAFT POLYMERIZATION IN scCO₂

- Complexity of the kinetics and mechanisms presents
 - Solubility of the RAFT agent, Z and R \cdot groups
 - Interaction between controller and surfactant in

dispersion polymerization



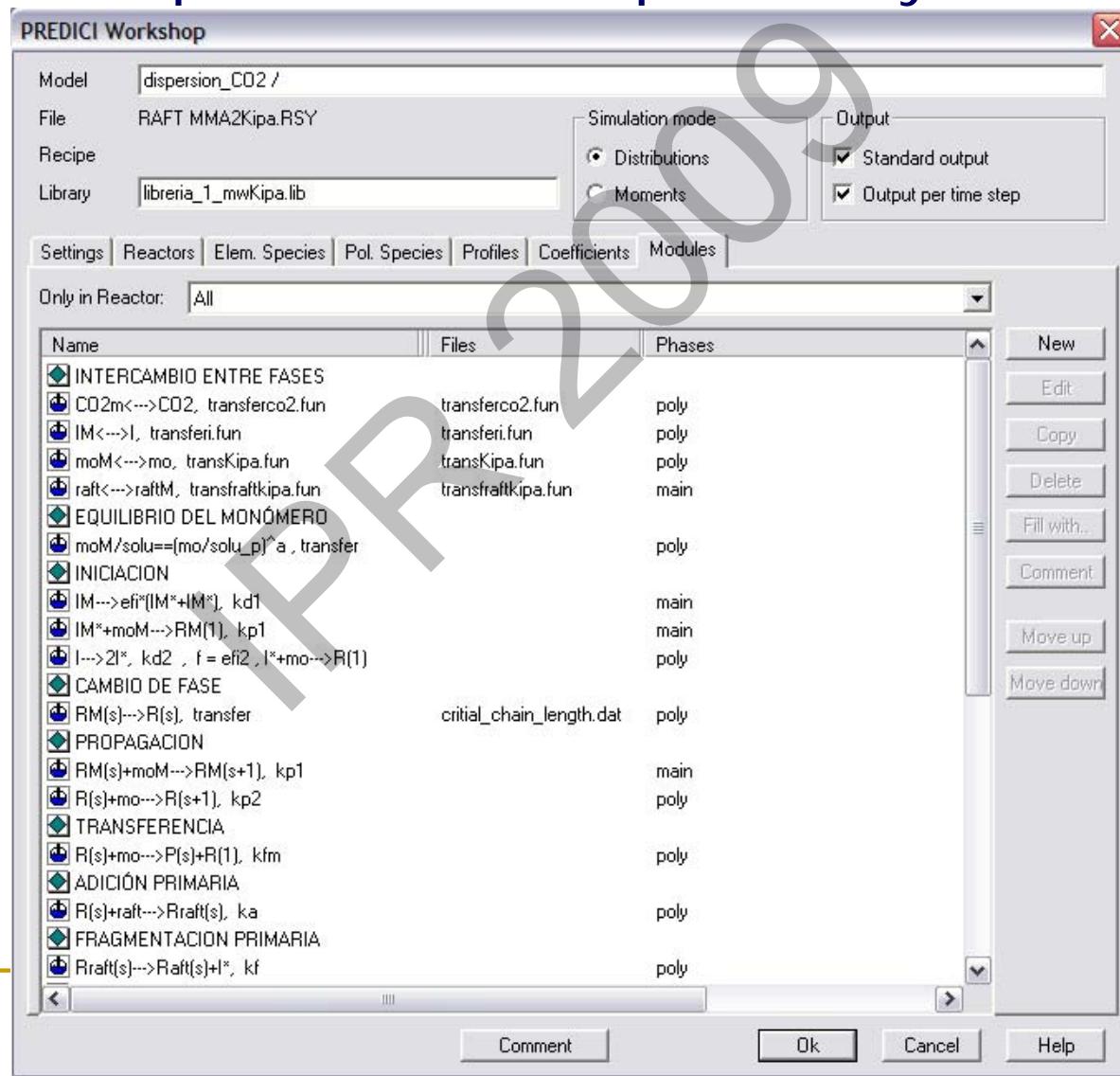
Krytox 157 FSL

STATE OF THE ART

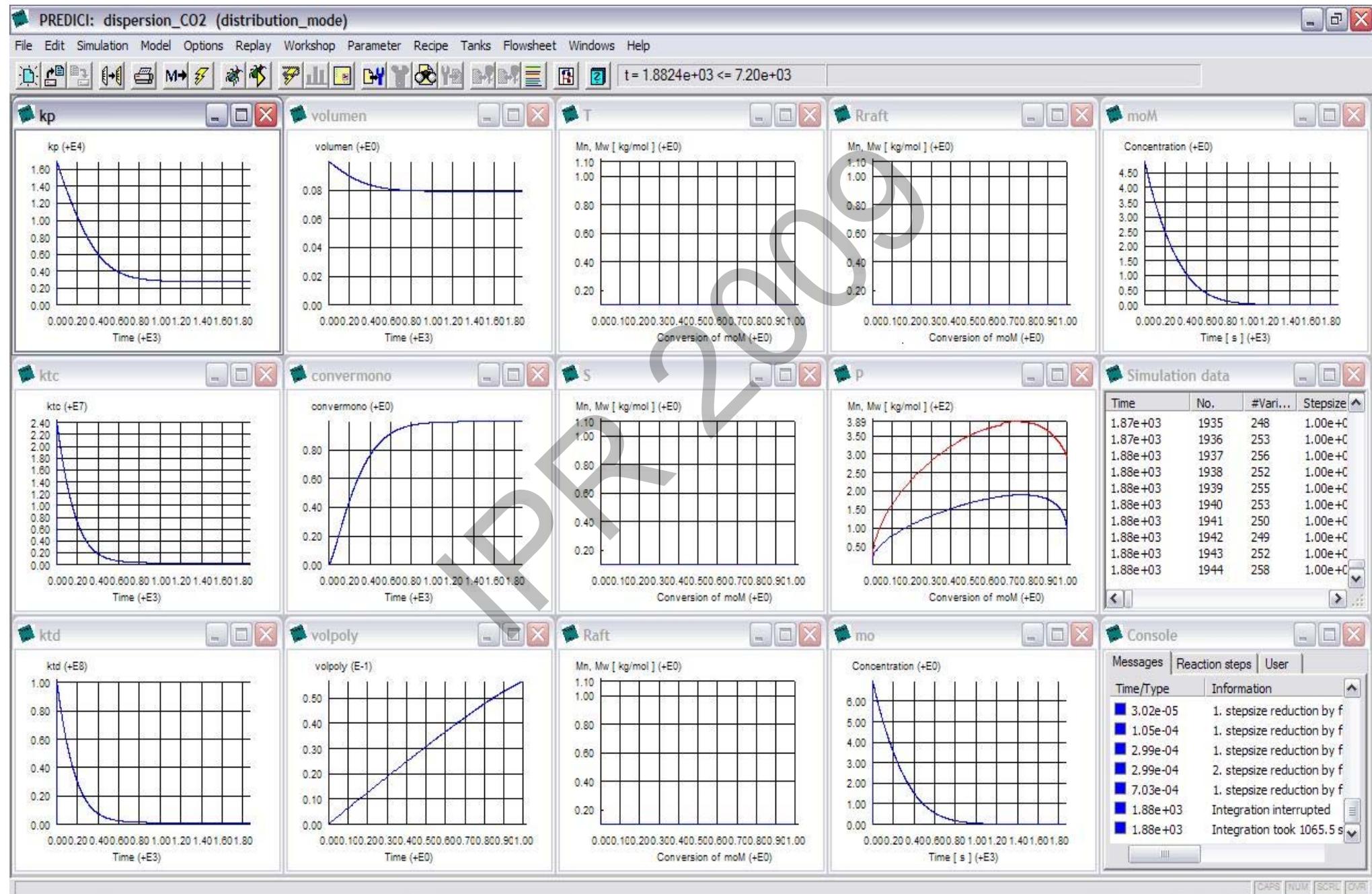
- ❖ Arita T., Beuermann, S., Vana, P. *Reversible addition fragmentation chain transfer (RAFT) polymerization of styrene in fluid CO₂*, e-Polymers **2004**, no. 003. <http://www.e-polymers.org>
- ❖ Arita, T., Beuermann, S. & Vana Philipp *RAFT Polymerization of Methyl Acrylate in Carbon Dioxide* Macromol. Mater. Eng. 2005, 290, 283-293
- ❖ Kristofer J. Thurecht, Andrew M. Gregory, Wenxin Wang, and Steven M. Howdle, "Living" *Polymer Beads in Supercritical CO₂* Communications to the Editor Macromolecules, Vol. 40, No. 9, 2007
- ❖ Andrew M. Gregory, Kristofer J. Thurecht and Steven M. Howdle *Controlled Dispersion Polymerization of Methyl Methacrylate in Supercritical Carbon Dioxide via RAFT* American Chemical Society. Macromolecules **2008**, 41, 1215-1222

MODEL AND SIMULATION

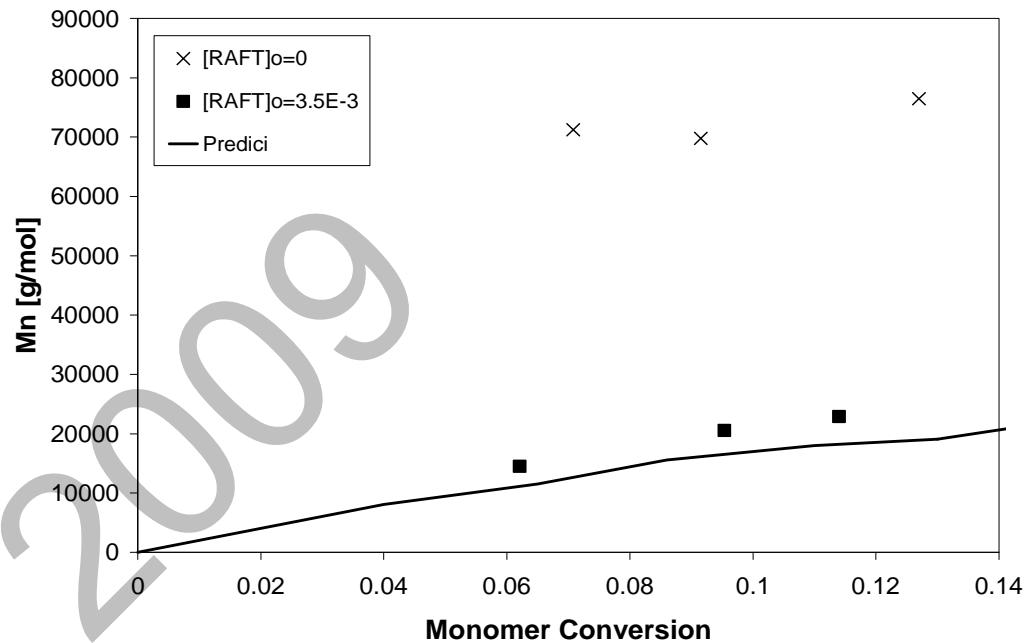
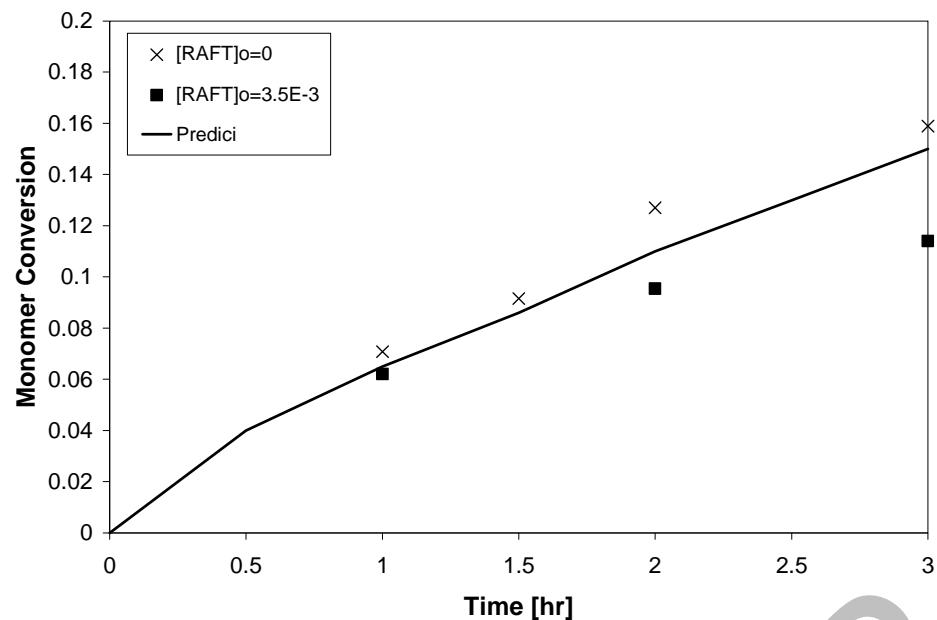
→ To develop a model, using Predici, for RAFT polymerization processes in dispersion systems in scCO₂.



SIMULATION RESULTS



SIMULATION RESULTS



T= 80°C

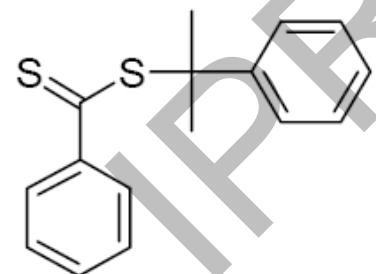
P= 300 bar

scCO₂= 20% v/w

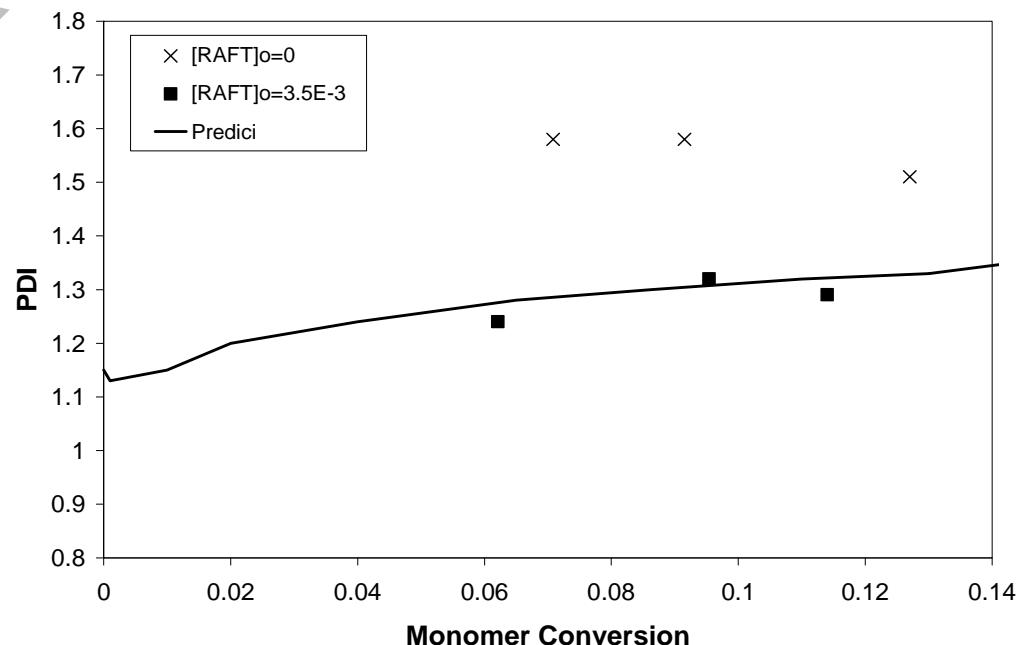
[AIBN]= 2.6X10⁻³M

[STY]= 6.49 M

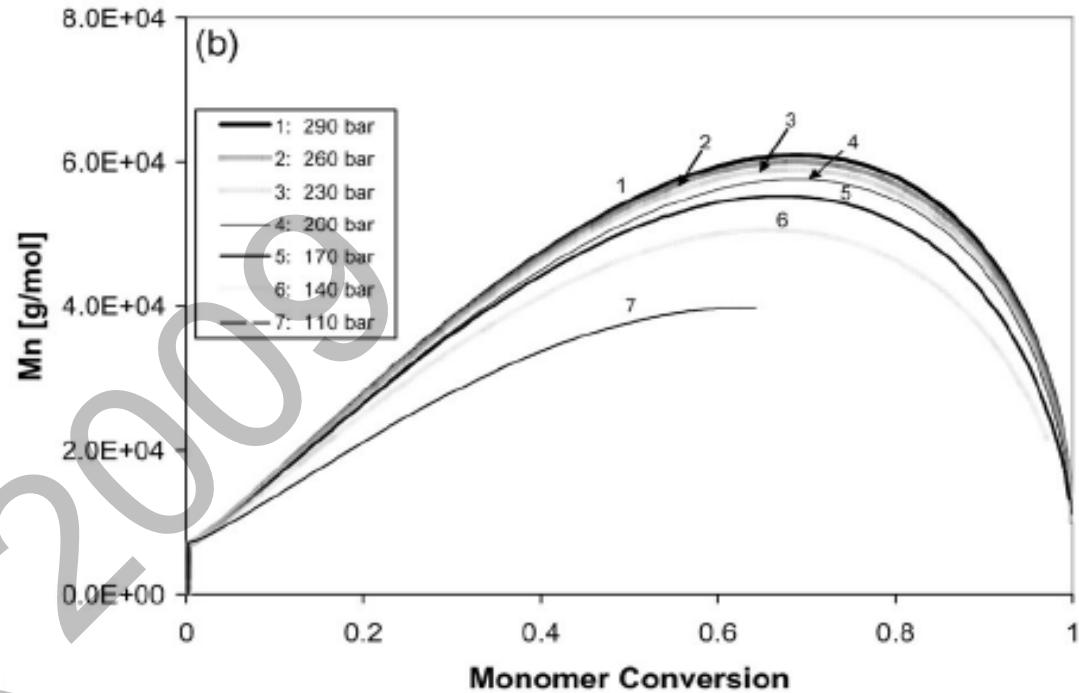
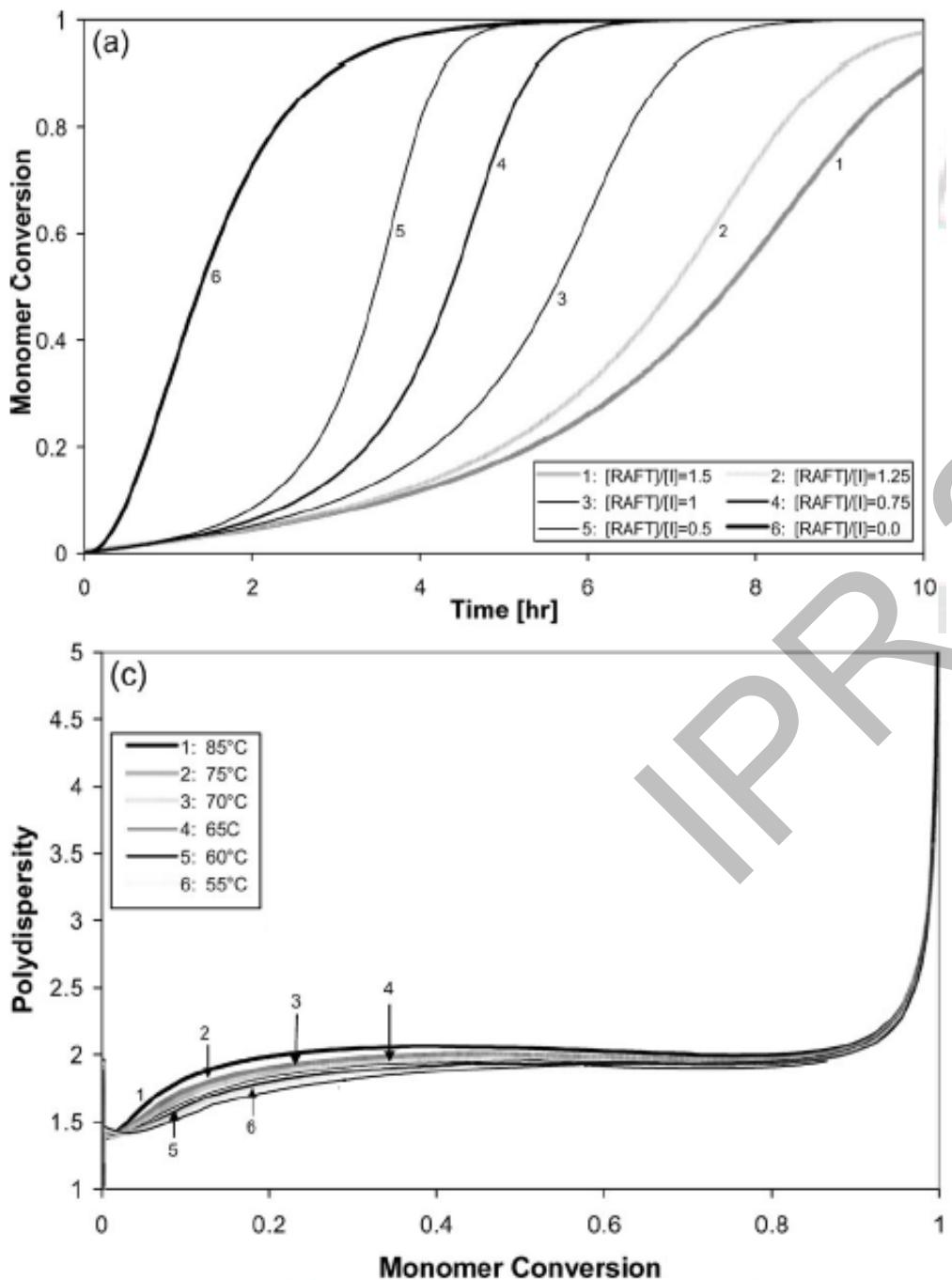
[RAFT]= 7.0X10⁻³M



*Cumil
Ditiobenzoate
(CDB)*



SIMULATION RESULTS



$T = 65^\circ\text{C}$

$P = 200 \text{ bar}$

$\text{scCO}_2 = 20\% \text{ v/w}$

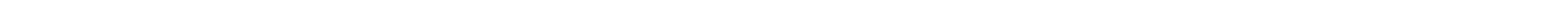
$[\text{AIBN}] = 1.19 \times 10^{-2} \text{ M}$

$[\text{MMA}] = 1.96 \text{ M}$

$[\text{RAFT}] = 1.19 \times 10^{-2} \text{ M}$

SIMULATION CONCLUSIONS

- The model captured the expected effects of RAFT agent to initiator ratio, pressure and temperature on polymerization rate and molecular weight development.
- More detailed and systematic experimental studies on this topic are needed for model validation purposes.



EXPERIMENTAL RESULTS

- To evaluate RAFT agents, commercially available or synthesized in dispersion polymerization in scCO₂
- Evaluate the effect of operating conditions, P, T on monomer conversion and MWD
- Designing and improved recipes and operating conditions for the RAFT dispersion polymerization in scCO₂

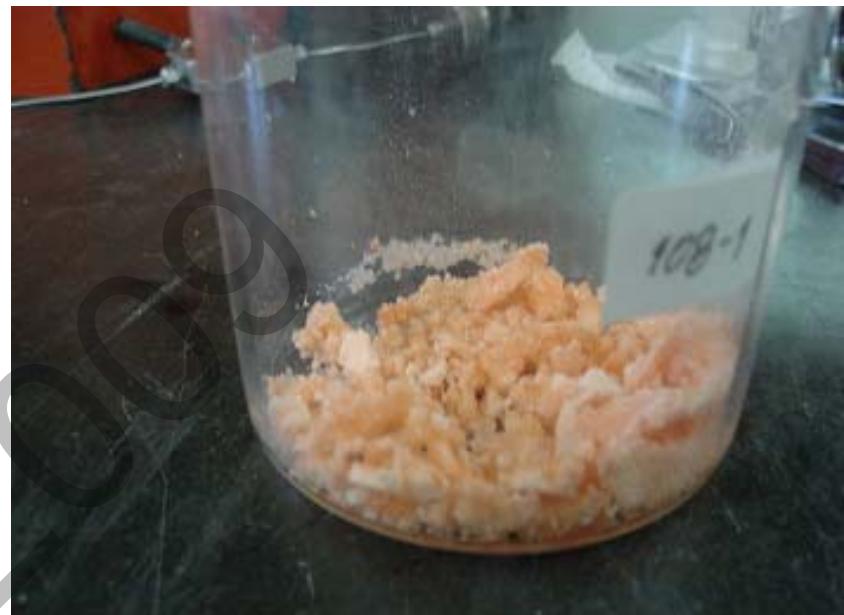
EXPERIMENTAL RESULTS



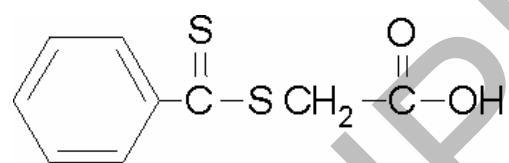
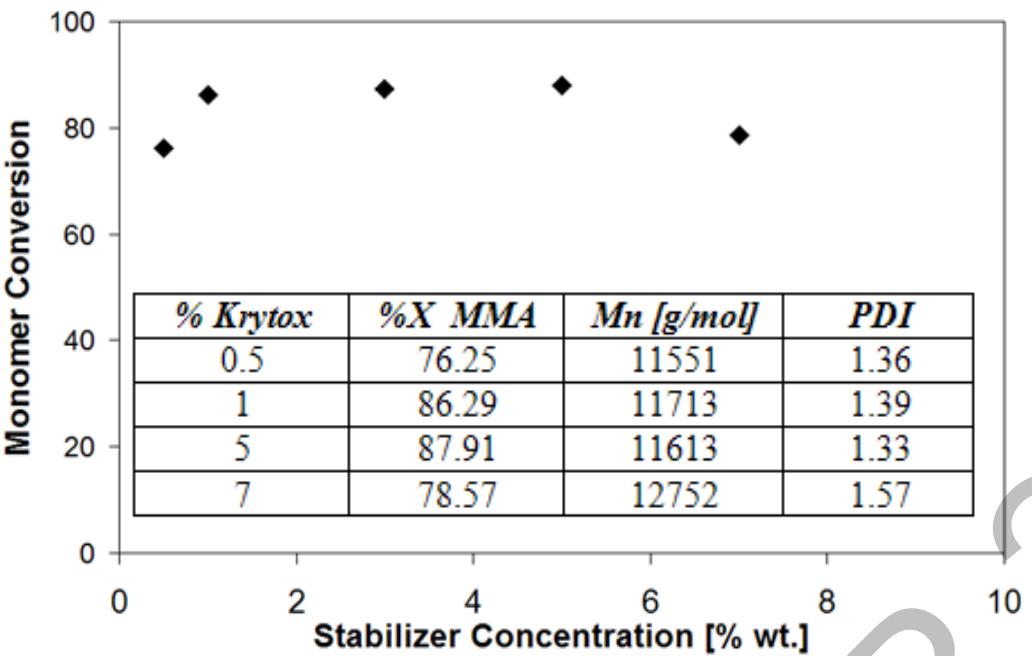
EXPERIMENTAL SYSTEM



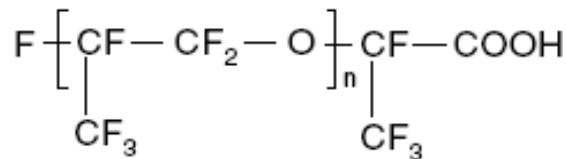
EXPERIMENTAL SYSTEM



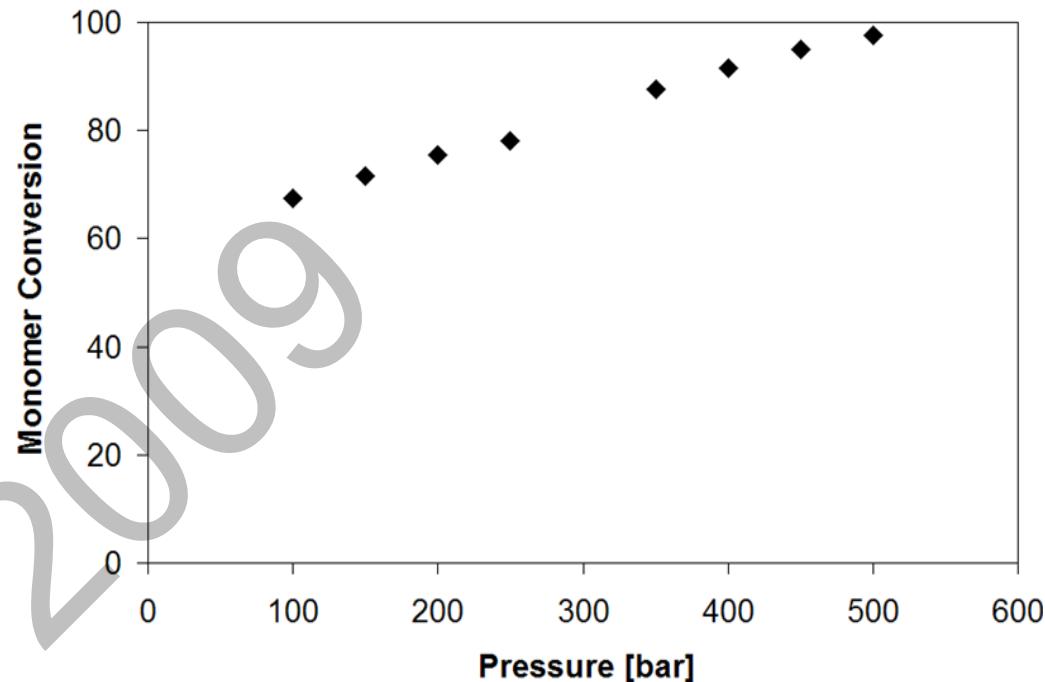
MMA RAFT POLYMERIZATION IN scCO₂



S-Thiobenzoyl thioglycolic acid



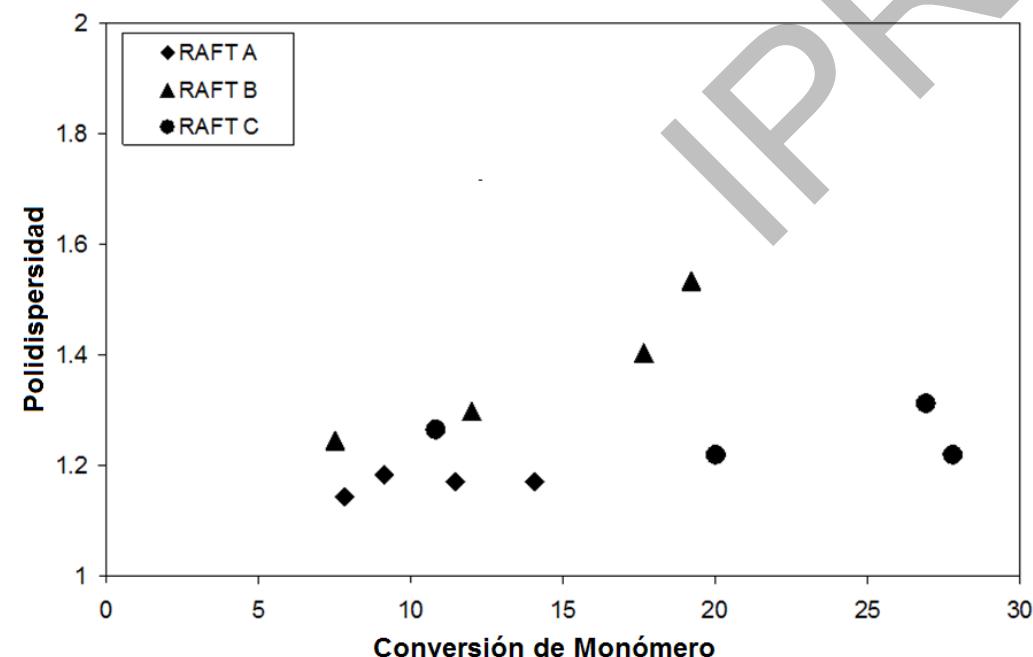
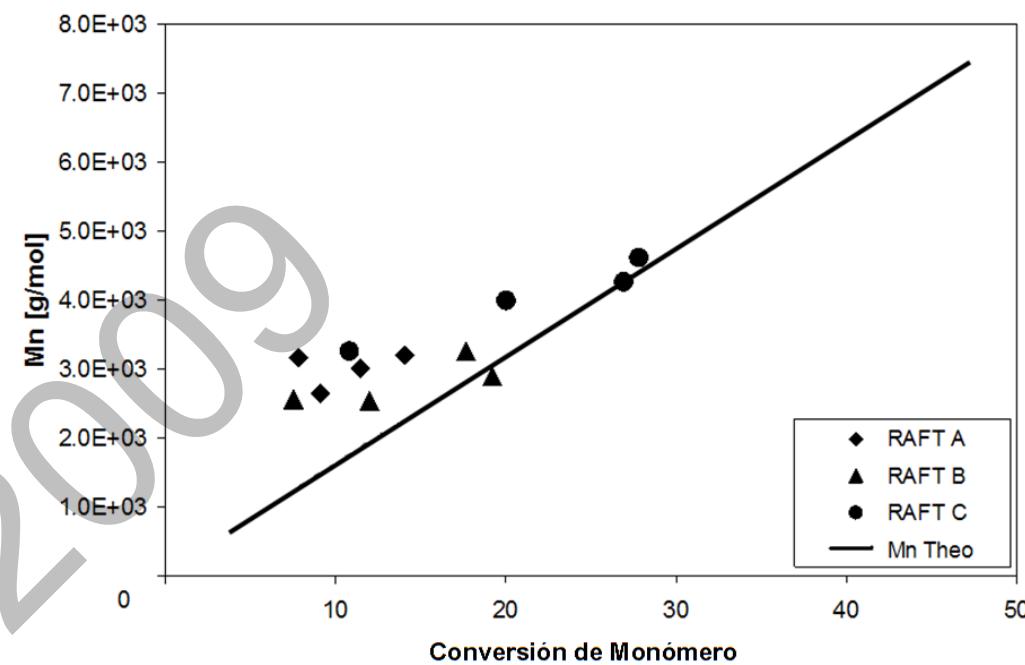
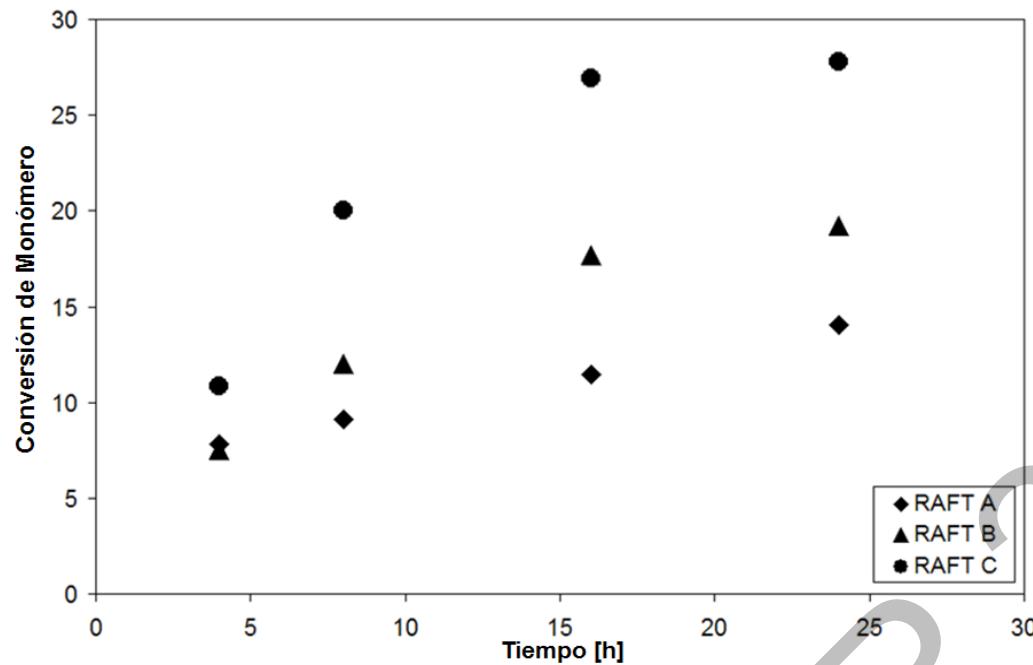
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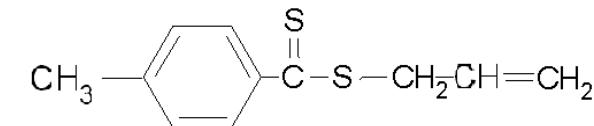
MMA RAFT POLYMERIZATION IN scCO₂



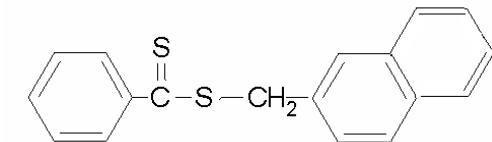
STY RAFT POLYMERIZATION IN ScCO₂



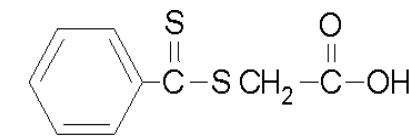
● *4-Methyl- allyl dithiobenzoate*



▲ *Methyl napthalene dithiobenzoate*



◆ *S-(Thiobenzoyl)thioglicolic acid*



Effect of Stabilizer Concentration and Controller Structure and Composition on Polymerization Rate and Molecular Weight Development in RAFT Polymerization of Styrene in Supercritical Carbon Dioxide. G.Jaramillo-Soto et. al.
Submitted to Polymer, under corrections.

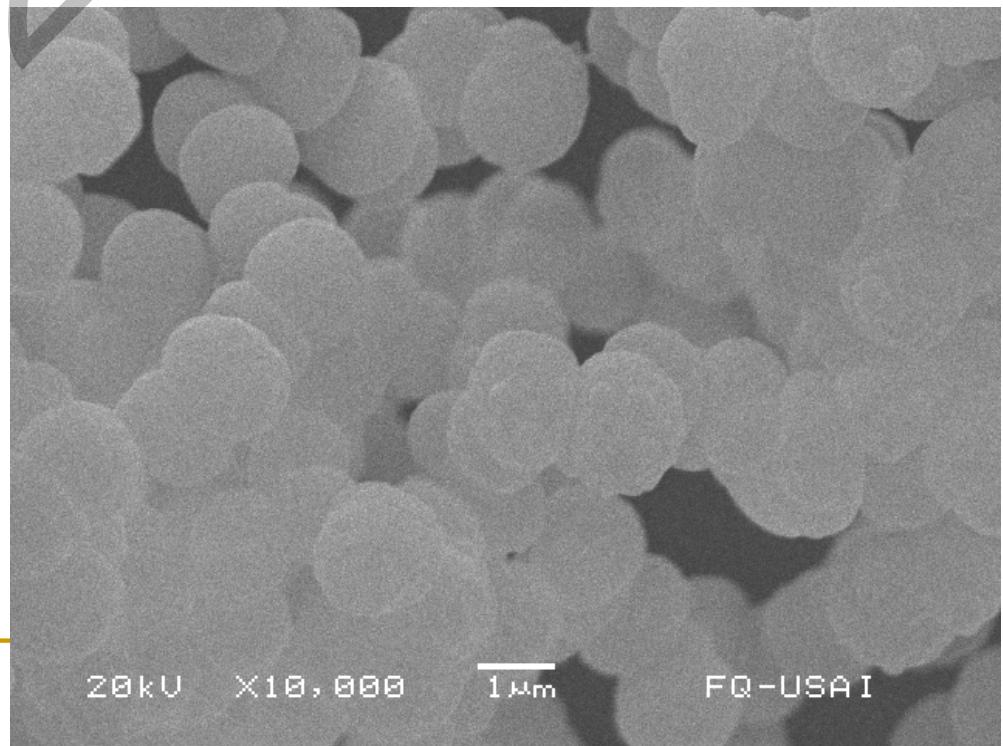
EXPERIMENTAL CONCLUSIONS

- There are too many experimental factors to study.
- Results strongly depends on the RAFT agent structure.
- S-(thiobenzoil) thioglycolic better results with STY than MMA.
- Stabilizer concentration importance.
- Model optimization.

PERSPECTIVES



IP 2009



20kV X10,000 1 μm

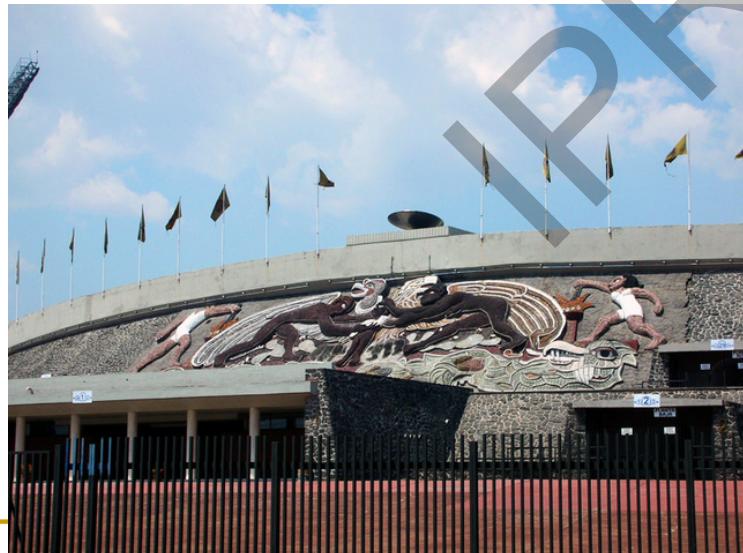
FQ-USAI

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THANK YOU

