

---

# SYNTHESIS OF ARBORESCENT POLYPEPTIDES FOR CONTROLLED DRUG DELIVERY APPLICATIONS

---

**Greg Whitton**, Mario Gauthier  
Department of Chemistry

IPR Symposium May 1<sup>st</sup>, 2009

# OUTLINE

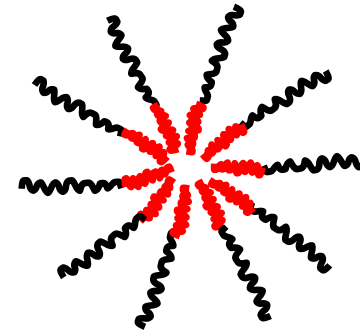
- Objectives
- Synthesis
- Results
- Future Work
- Acknowledgements

# PROJECT OBJECTIVES

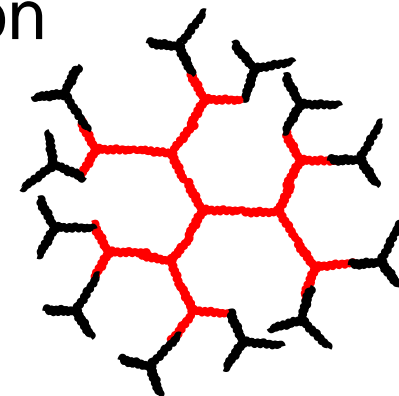
- Synthesize biocompatible branched polymers with narrow molecular weight distributions (MWD)
  - Unimolecular micelles
- Demonstrate the feasibility of the synthetic scheme
  - Grafting yield and coupling efficiency
- Demonstrate suitability for applications in microencapsulation (drug delivery)
  - Solubilization and release of hydrophobic probe molecules

# MICELLES

- Linear diblock copolymer micelles
  - Dynamic structure: Molecules in micelle interchange with free molecules
  - critical micelle concentrations (unstable at  $C < CMC$ )

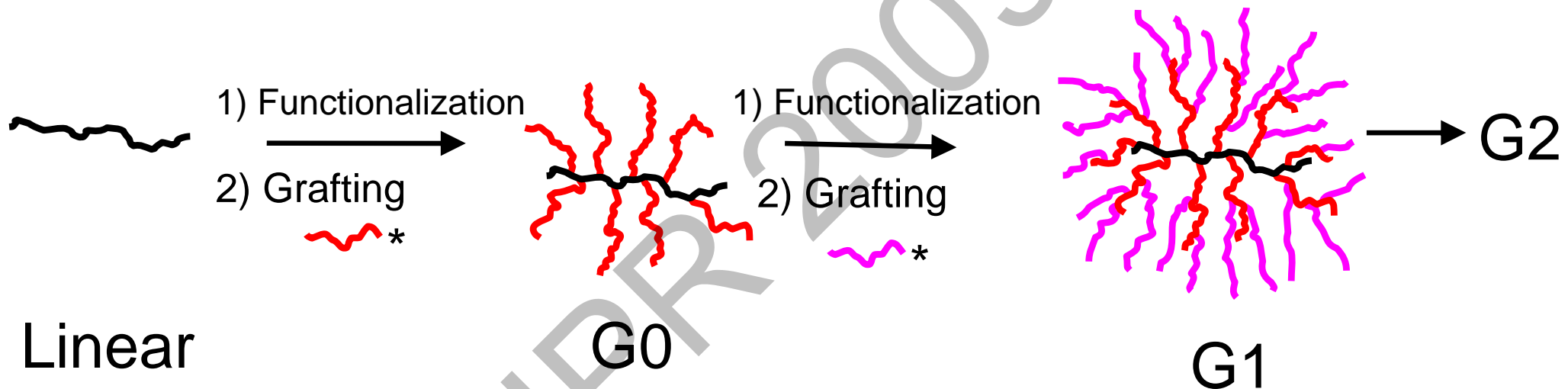


- Dendritic micelles
  - Narrow molecular weight distribution
  - low molecular weights



# ARBORESCENT MICELLES

- Branched structure obtained from **grafting onto** scheme



- Static structure (stable, no CMC)
- High molecular weights achieved in only a few grafting cycles
- Biocompatibility: Amino acid derivatives



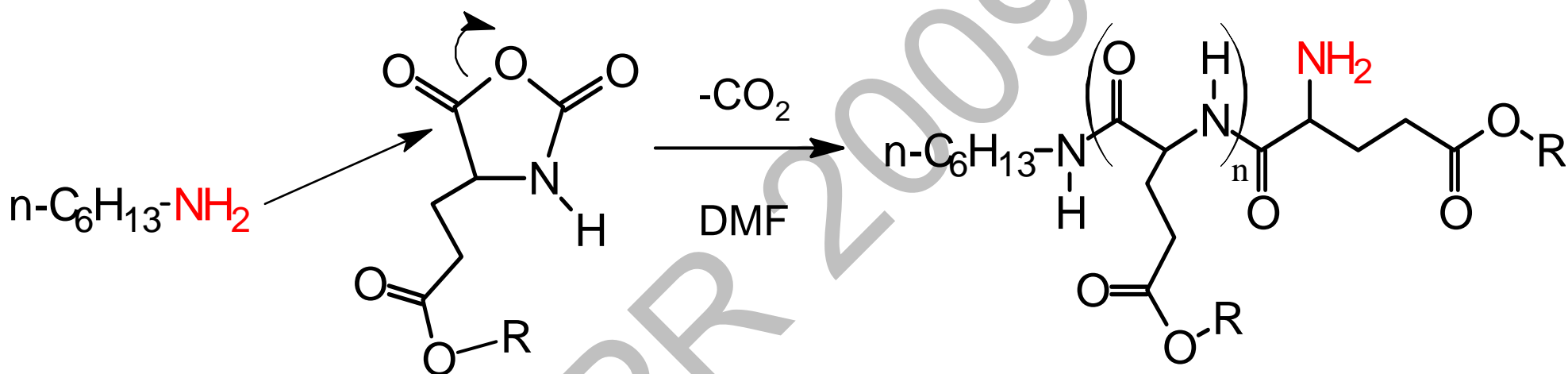
# SYNTHESIS

IPR 2009

# SYNTHETIC SCHEME

- Ring opening 'living' polymerization
  - $\gamma$ -benzyl or  $\gamma$ -t-butyl L-glutamate monomers
- Partial deprotection of poly( $\gamma$ -benzyl L-glutamate) (PBLG) substrate to generate randomly distributed grafting sites
- Grafting
  - Comb-branched polymer (Generation zero, G0)
  - Higher generations G1, G2 .....etc.
- Shell modification
  - Attach water-soluble components for micelle formation

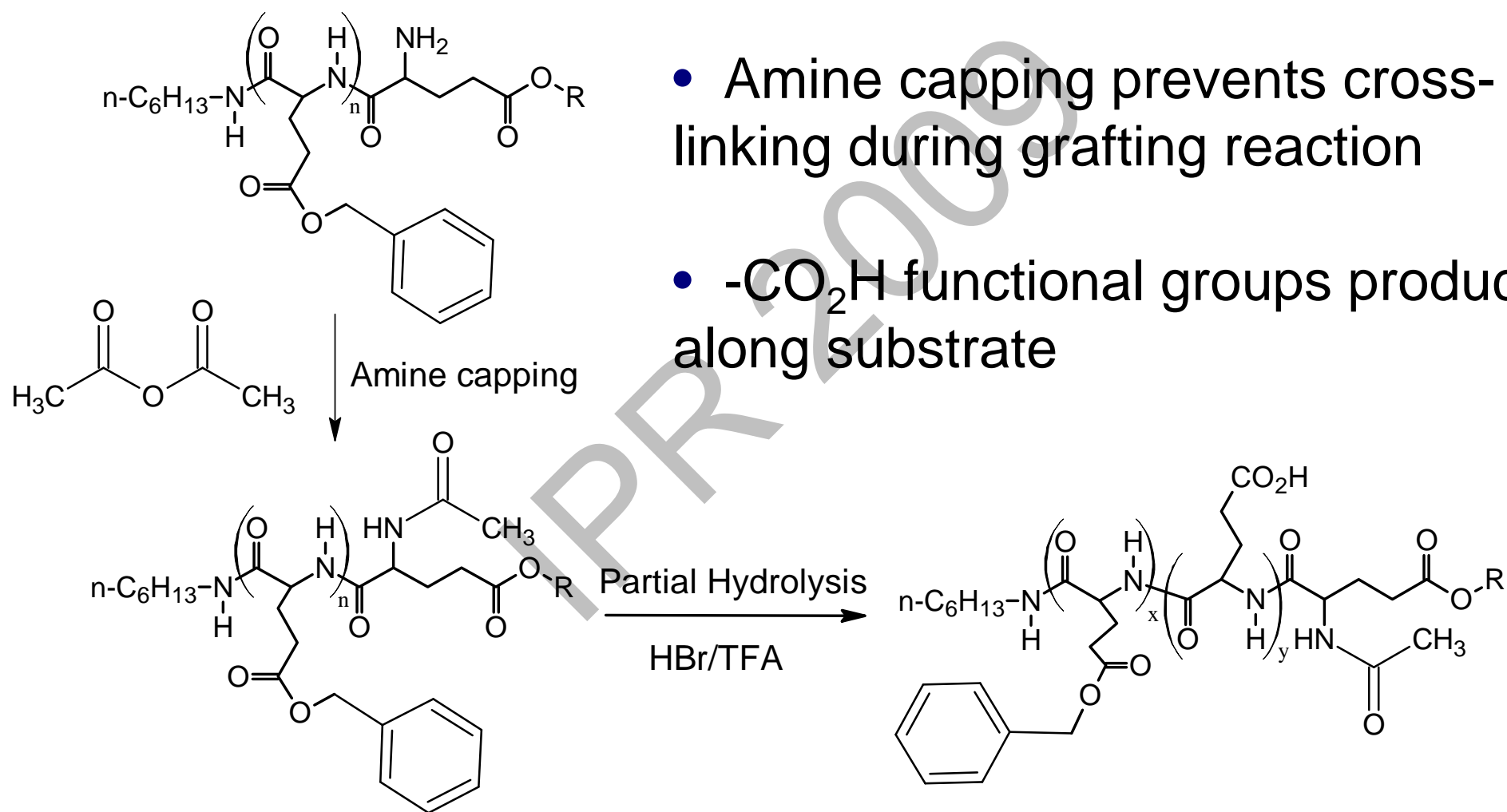
# SYNTHESIS OF SIDE CHAINS



**'living' primary amine end group**



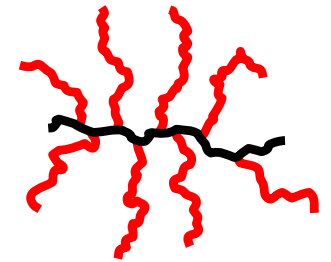
# SYNTHESIS OF SUBSTRATE



- Amine capping prevents cross-linking during grafting reaction
- $-\text{CO}_2\text{H}$  functional groups produced along substrate

# GRAFTING

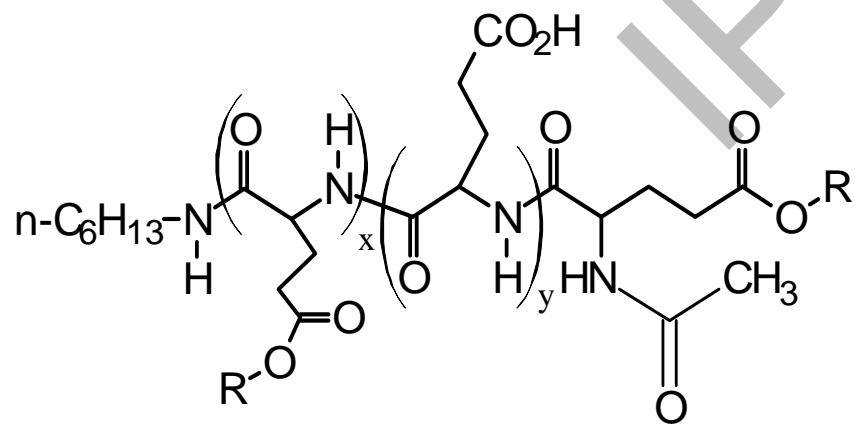
- Substrate coupling with side chains
  - Grafting sites activated by carbodiimide technique
    - DIC/HOBt
  - Stoichiometry varied to maximize grafting yield



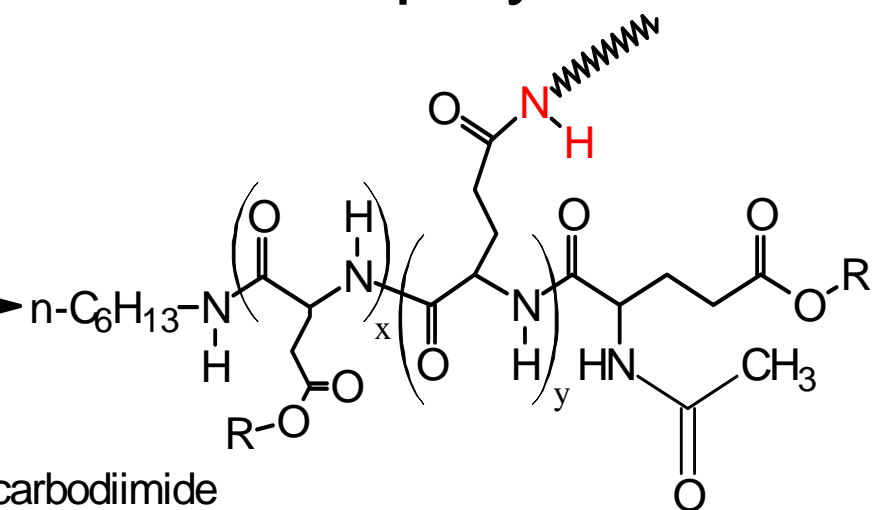
G0 polymer



+

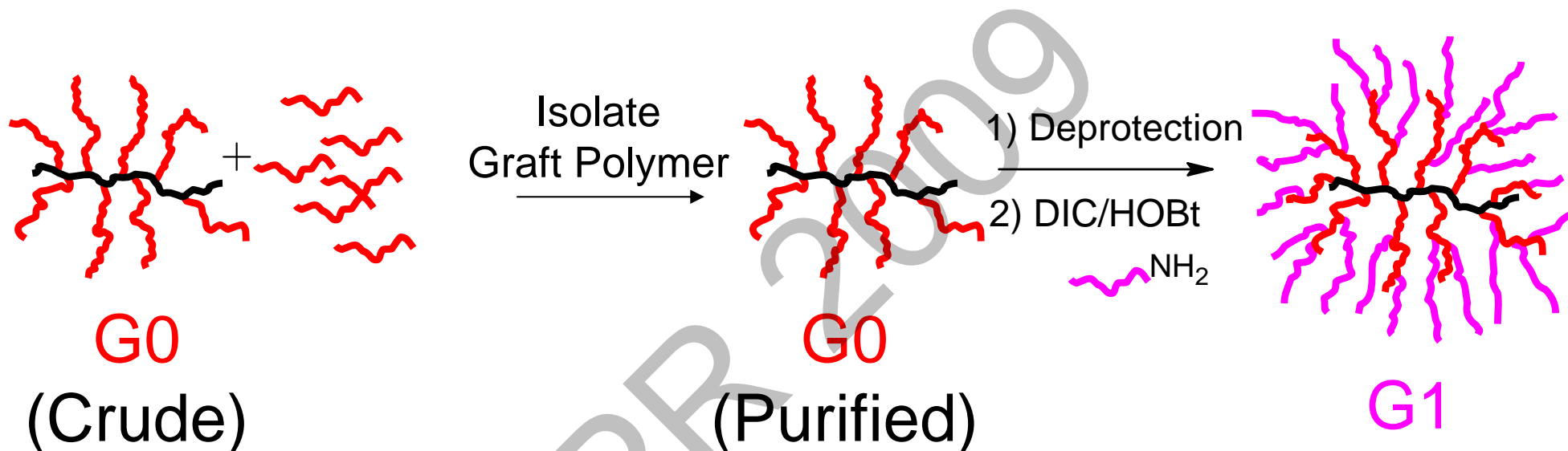


**SUBSTRATE**



DIC = 1,3-Diisopropylcarbodiimide  
HOBt = 1-Hydroxybenzotriazole

# GENERATIONS 1, 2,...

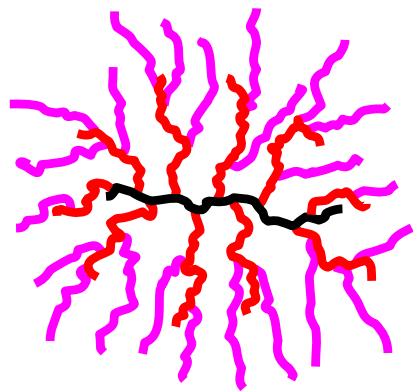


- Graft polymer isolated with preparative GPC column  
Fast separation of 60-80 mg crude product per injection

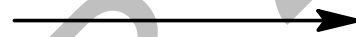
Analytical Column: 10 mm I.D.  
Prep Column: **22 mm I.D.**

# MICELLES: SHELL ADDITION

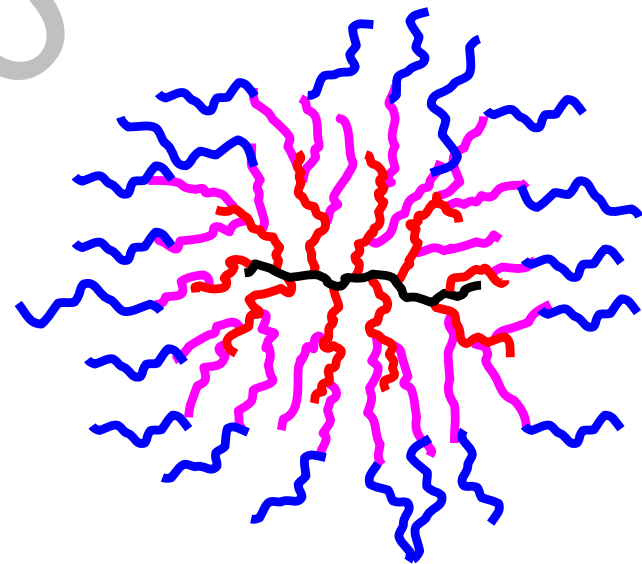
- Last grafting cycle: Hydrophilic side chains



1) Deprotection



2) DIC/HOBt



Water-soluble  
Unimolecular micelle

# CHARACTERIZATION

- Gel Permeation Chromatography (GPC)
  - Apparent and absolute molecular weights ( $M_n$ ,  $M_w$ )
  - PDI for linear and graft polymers
  - Grafting yield
- NMR Spectroscopy
  - Number-average degree of polymerization ( $DP_n$ )
    - Absolute  $M_n$
  - Deprotection level of substrate
  - Derivatization/ $^{19}\text{F}$  NMR to detect primary amine chain ends
- Static Light Scattering
  - Absolute  $M_w$ , PDI
  - Coupling efficiency
- UV and Fluorescence Spectroscopy
  - Monitoring of probe solubilization capacity and kinetics



# RESULTS

IPR 2009

# RING OPENING POLYMERIZATION

## Preparation of Poly( $\gamma$ -benzyl L-glutamate) Linear Side Chains

Sample #	% Yield	<sup>1</sup> H NMR		GPC <sup>a</sup>	
		DP <sub>n</sub>	M <sub>n</sub>	M <sub>n</sub> <sup>apparent</sup>	M <sub>w</sub> /M <sub>n</sub>
51	89	18.1	4,300	6,500	1.10
55	87	21.0	4,900	8,600	1.10
56	89	19.2	4,500	7,000	1.13
57	94	16.7	4,000	4,600	1.12
58	86	18.9	4,500	5,900	1.14
59	94	16.3	3,900	6,000	1.14

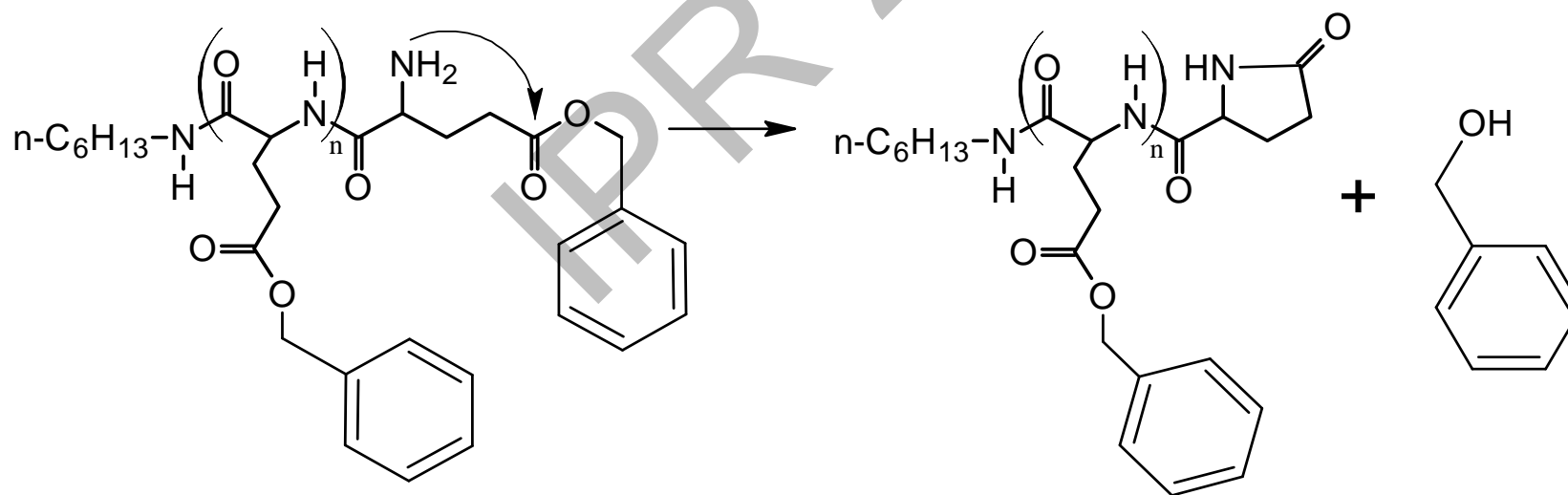
<sup>a</sup> DRI detector

→ PDI = 1.10-1.14, satisfactory

→ Chain functionality (-NH<sub>2</sub>) level determines coupling efficiency

# SIDE REACTIONS IN ROP

- Destroy 'living' primary amine
  - Broaden MWD
- ➔ Mostly end group cyclization
- Specific to poly( $\gamma$ -benzyl L-glutamate)



➔ Modify reaction conditions to reduce side reactions



# STRATEGIES TO AVOID SIDE REACTIONS

- Hydrochloride salt initiator

- Long reaction times (5-10 days)

W Vayaboury et al. *Macromol. Rapid Commun.* **2004**

- High vacuum techniques

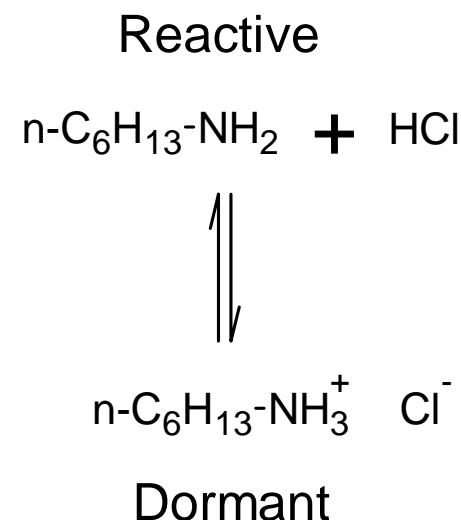
- Time-consuming

T Aliferis et al. *Biomacromolecules* **2004**.

- Lower polymerization temperature

- Moderate reaction times (2-3 days), easy setup

I Dimitrov et al. *Chem. Commun.* **2003**.

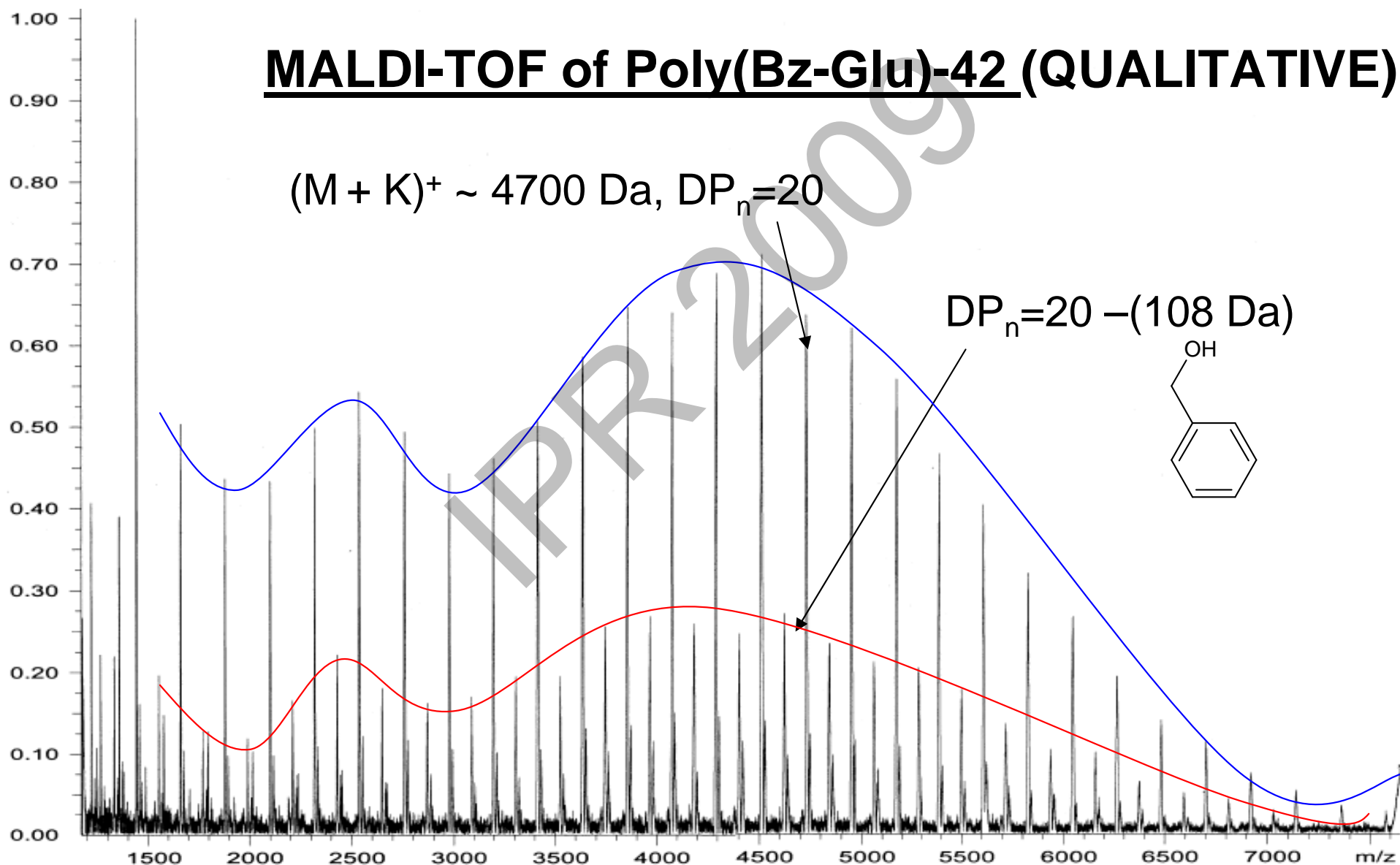
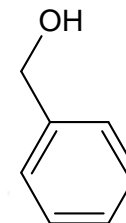


# DETECTION OF SIDE REACTIONS IN ROP

## MALDI-TOF of Poly(Bz-Glu)-42 (QUALITATIVE)

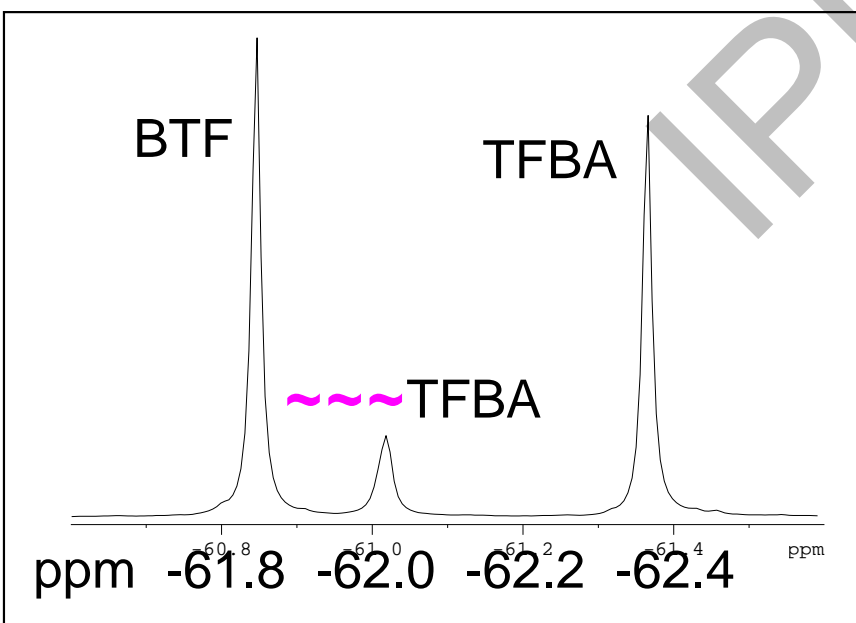
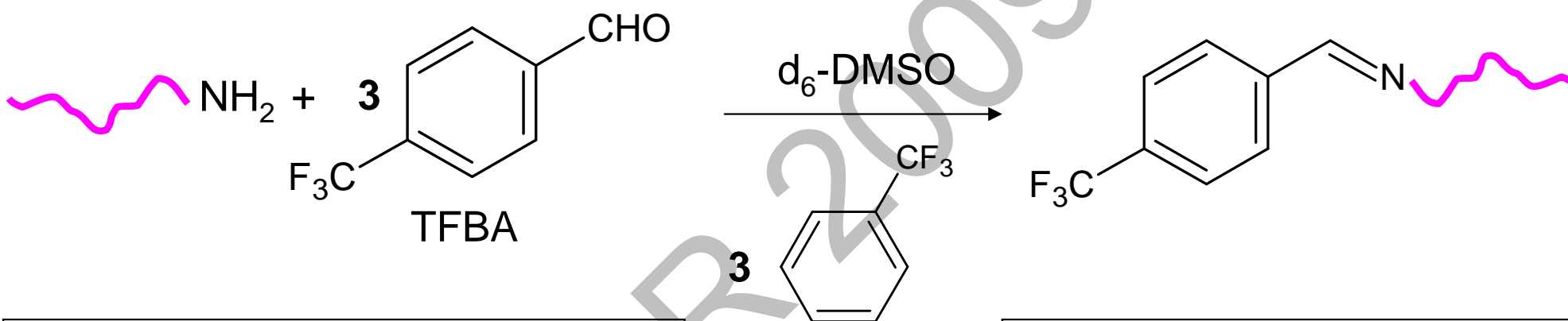
$(M + K)^+ \sim 4700 \text{ Da}$ ,  $DP_n = 20$

$DP_n = 20 - (108 \text{ Da})$



# DETECTION OF SIDE REACTIONS IN ROP

- Using  $^{19}\text{F}$  NMR spectroscopy to quantify  $-\text{NH}_2$



	$\text{NH}_2$	TFBA	BTF
Before	1	3	3
After	1	2	3
Actual	1	2.52	3.20
TFBA	$f_{\text{NH}_2} = 2/2.25 = 79\%$		
BTF	$f_{\text{NH}_2} = 3/3.58 = 84\%$		

# DETECTION OF SIDE REACTIONS IN ROP

Sample #	% NH <sub>2</sub> groups	
	TFBA	BTF
Poly(Bz-Glu)-20	20	25
Poly(Bz-Glu)-30	58	66
Poly(Bz-Glu)-50	15	20
Poly(Bz-Glu)-58	74	79
Poly(Bz-Glu)-64	79	84
Poly(Bz-Glu)-65	40	43.7
Poly(Bz-Glu)-71	41	46
PEG1900-NH <sub>2</sub>	72	79

- ~5% Discrepancy between TFBA and BTF methods, but sample grading is mostly consistent
- **QUANTITATIVE** method

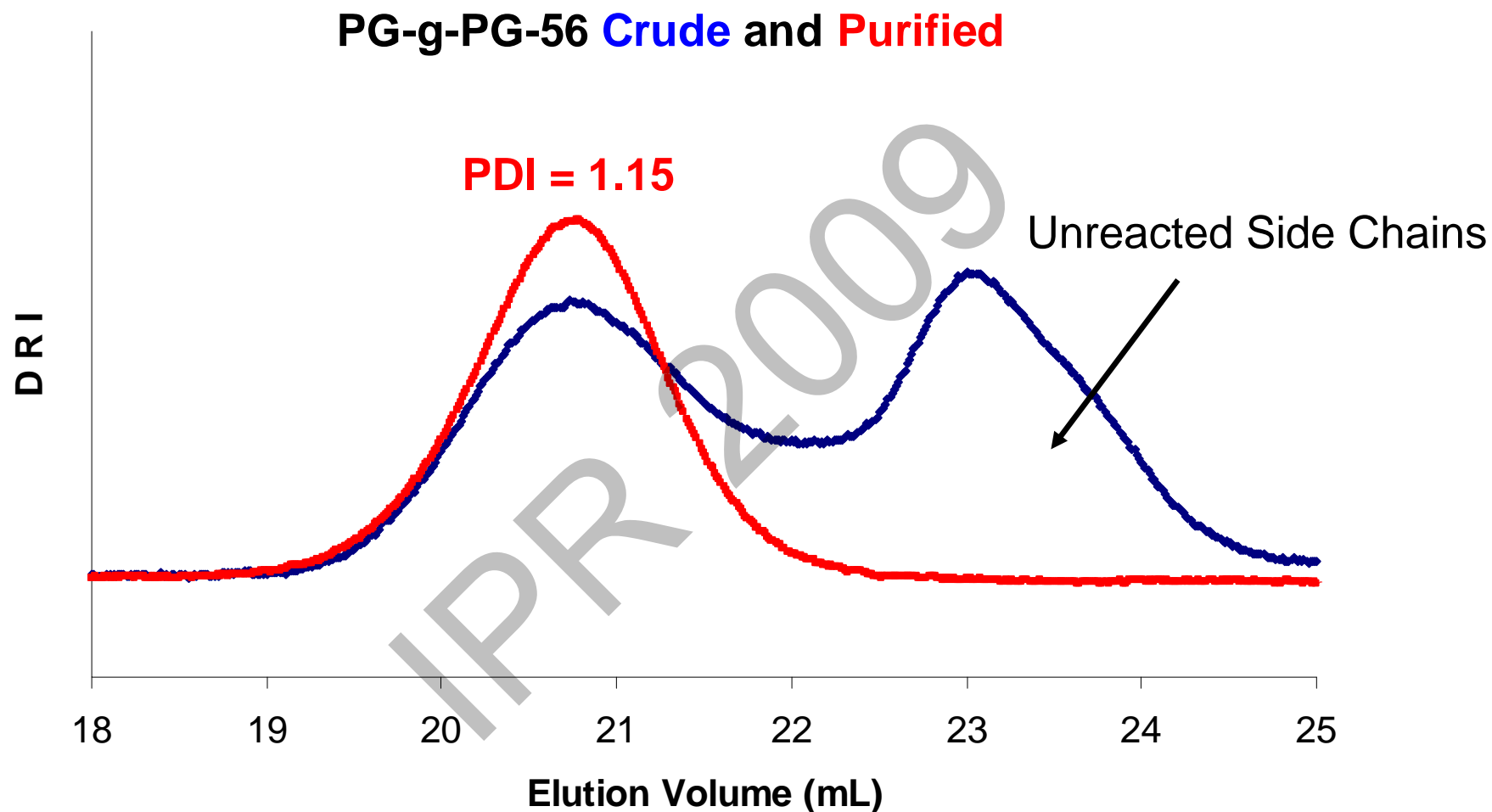
# DEPROTECTION OF SUBSTRATES

## Deprotection of Poly( $\gamma$ -benzyl L-glutamate) Substrates

Sample	Target Deprotection	Mole Ratio of HBr to Benzyl ester Units	% Deprotection	
			$^1\text{H}$ NMR	Titration
34	30	0.25:1	31	30
B	30	30:70	33	34
C	30	30:70	31	30
<b>G0-52</b>	30	0.3:1	32	--
<b>G0-53</b>	30	0.3:1	32	--
<b>G1-2</b>	20	0.2:1	16	--
<b>G1-4</b>	30	0.3:1	16	--
<b>G2-3</b>	30	0.3:2	26	--

- Good agreement between  $^1\text{H}$  NMR analysis and titration
- Titration difficult for arborescent substrates

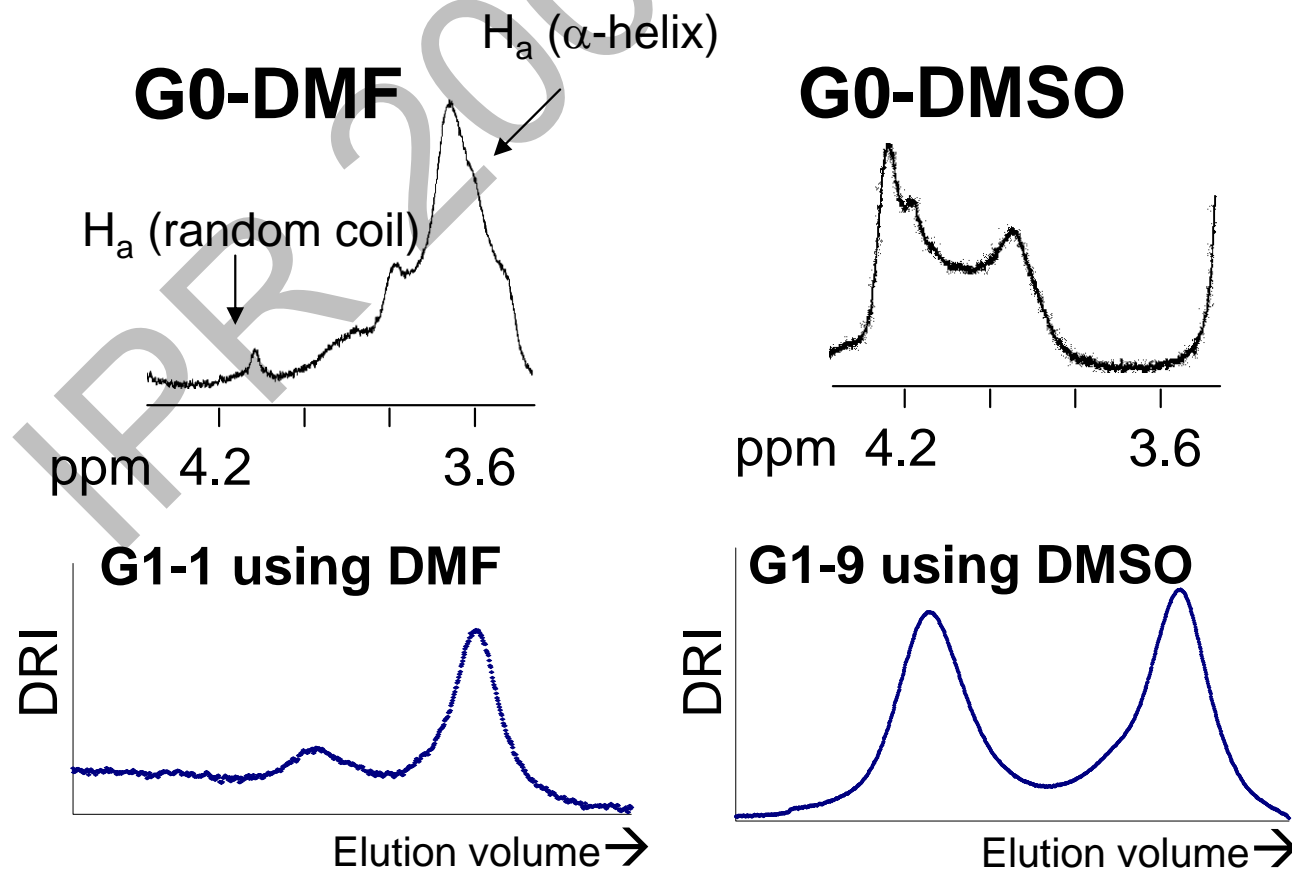
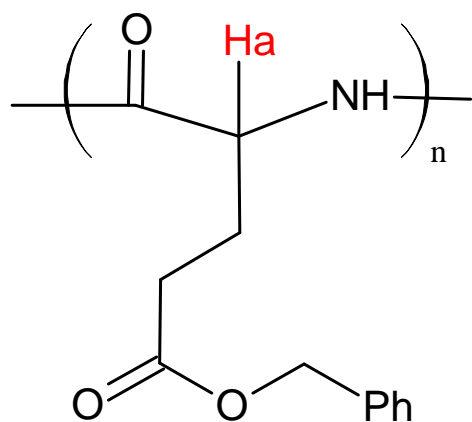
# GRAFTING REACTION: GPC ANALYSIS



- Purification achieved after only one injection
- Narrow MWD maintained after purification

# HINDERED GRAFTING REACTION

- No reaction for grafting on G0 in DMF
  - Possible  $\alpha$ -helix conformation of G0 substrate and side chains
  - Run  $^1\text{H}$  NMR to confirm (500 MHz)



# GENERATIONS 1, 2,...

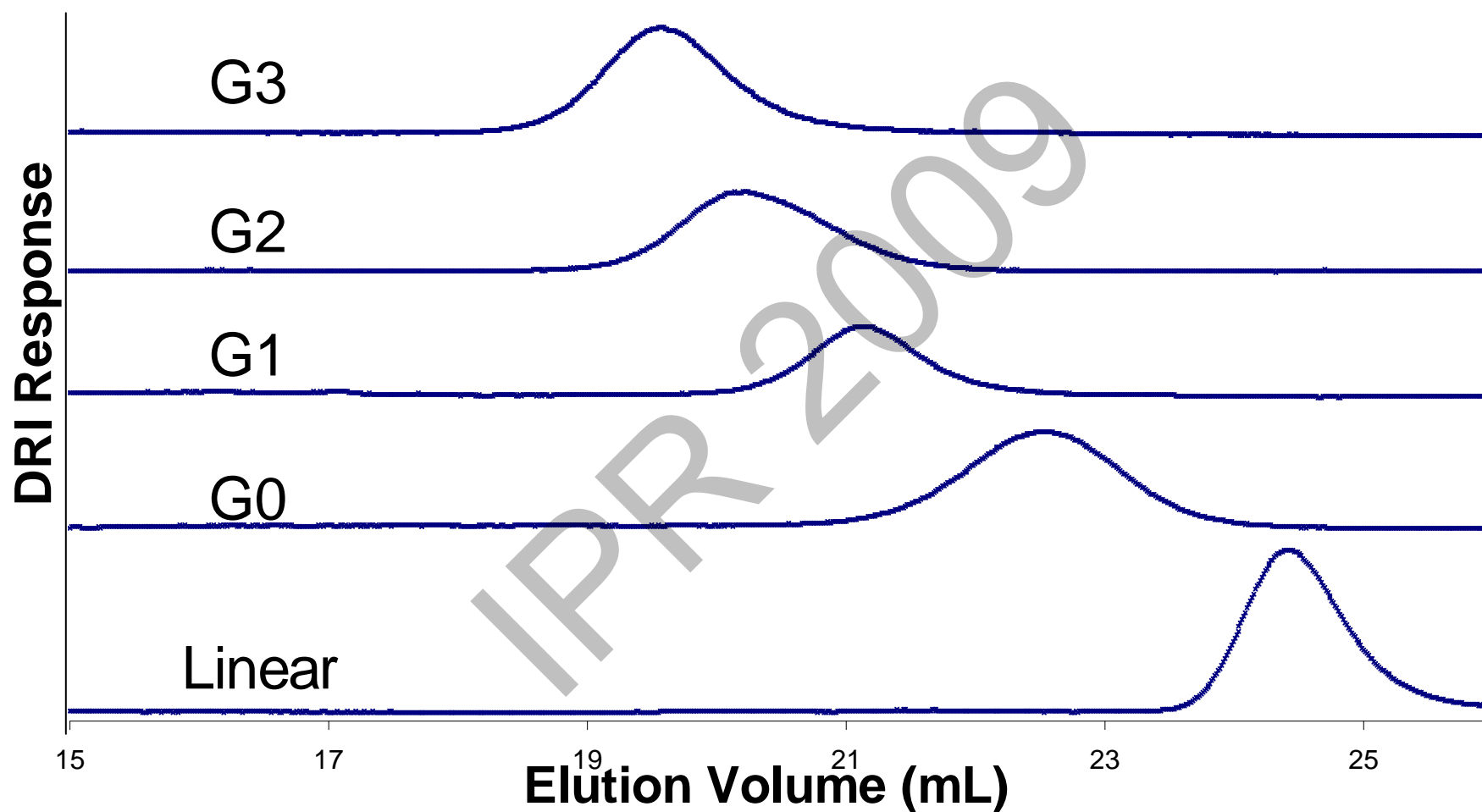
## Characteristics of Arborescent Poly( $\gamma$ -benzyl L-glutamate) for Successive Generations

Sample #	$M_n$ Side chain <sup>a</sup>	DRI	MALLS		Grafting Yield (%) <sup>b</sup>	Branching Functionality (%) <sup>c</sup>
		$M_n^{\text{app}}$	$M_n$	$M_w/M_n$		
G0	4,300	17,400	48,000	1.04	38	9
G1	4,000	39,300	133,000	1.06	63	21
G2	3,900	83,100	486,000	1.03	46	90
G3	3,900	134,000	1,060,000	1.03	32	147

<sup>a</sup>  $^1\text{H}$  NMR, <sup>b</sup> DRI detector, <sup>c</sup> MALLS Detector



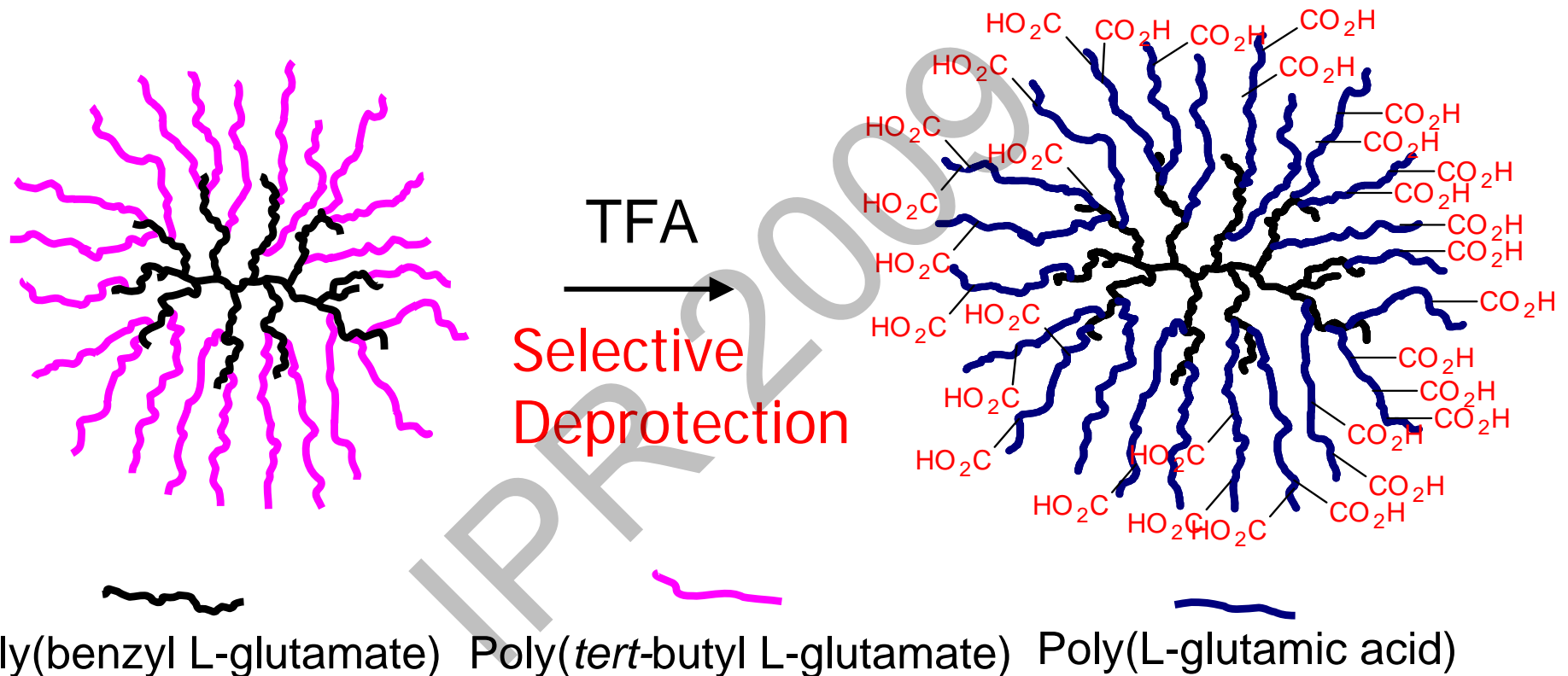
# GENERATIONS 1, 2,...



- Narrow MWD maintained
- Graft polymers free of linear contaminant

# MICELLES: SHELL ADDITION

- Last grafting cycle: Poly( $\gamma$ -*tert*-butyl L-glutamate) side chains



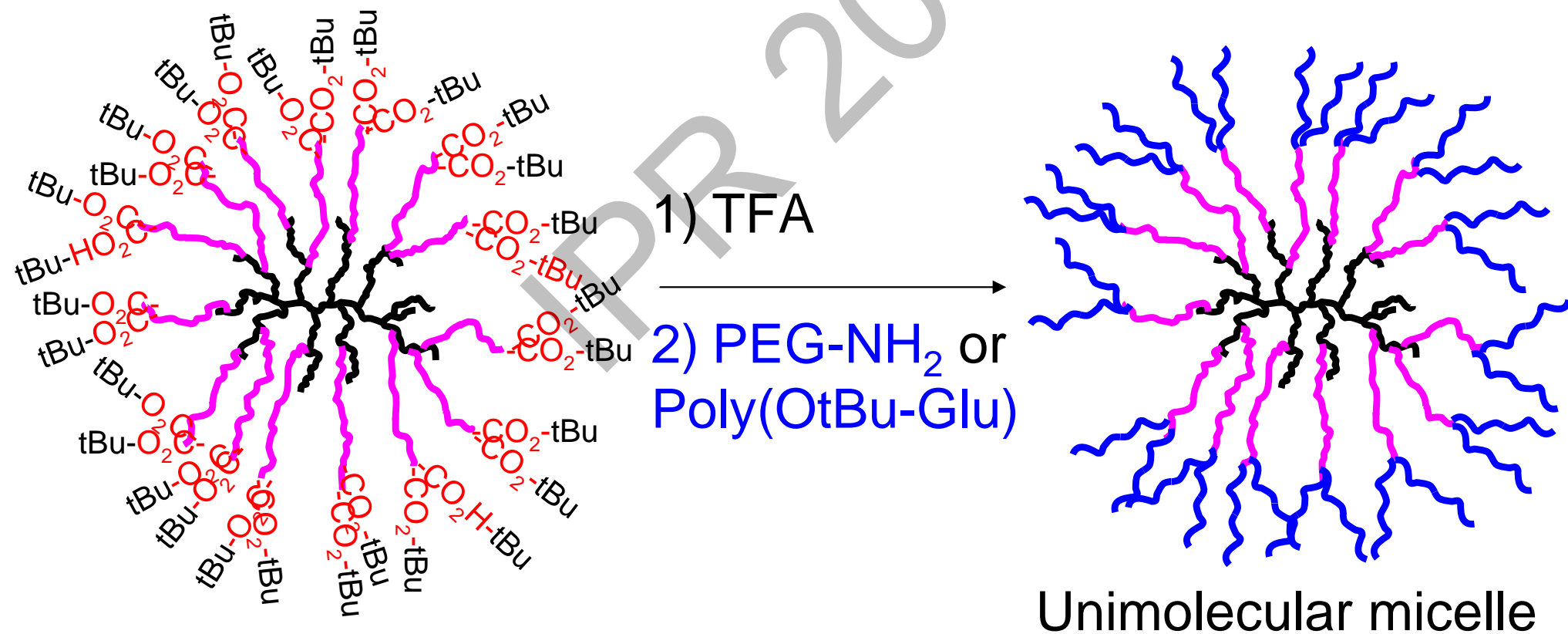
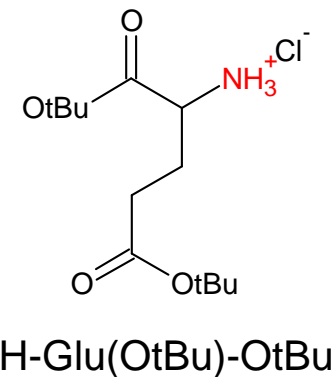
- Soluble in water and alkaline water solutions  
→ Gradual desolubilization by dialysis
- $-\text{CO}_2\text{H}$  groups buried in core (→ less hydrophobic)

# FUTURE WORK

- Modification to generate water-soluble micelles
  - Addition of longer or more PEG or Poly(OtBu-Glu) segments
  - Addition of hydrophilic segments at chain ends
- Solubilization studies with hydrophobic probes
  - UV-monitored
- Study of release kinetics
  - Fluorescence- and UV-monitored

# MICELLES: SHELL ADDITION 2

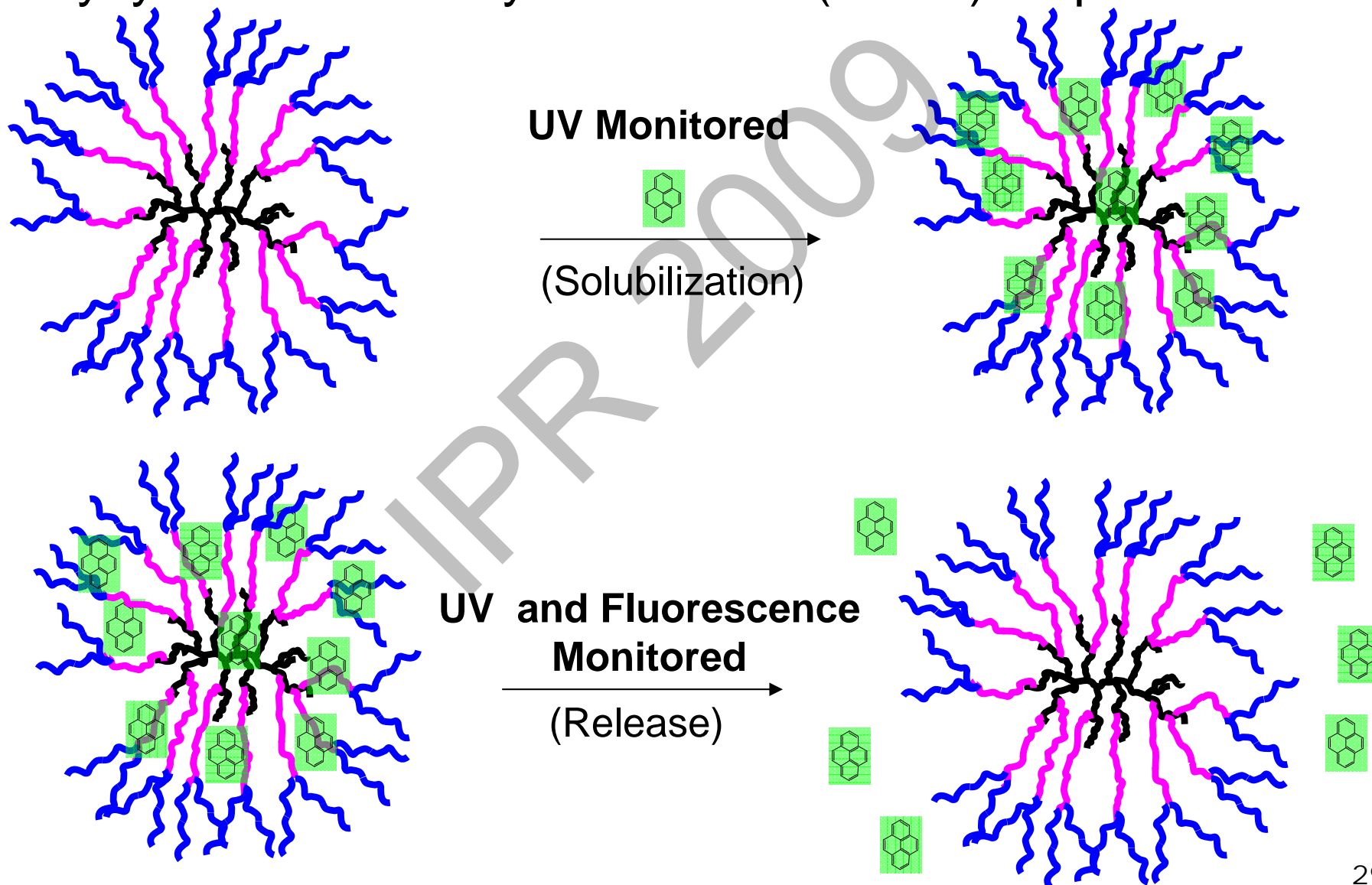
- Last grafting cycle: Side chains with H-Glu(OtBu)-OtBu initiator
- Remove protecting groups to functionalize ONLY chain ends (two  $-\text{CO}_2\text{H}$  per chain end)



Unimolecular micelle

# SOLUBILIZATION AND RELEASE STUDIES

- Polycyclic aromatic hydrocarbons (PAHs) as probes



# ACKNOWLEDGEMENTS

- Dr. Gauthier
- Dr. Duhamel
- Members of the Gauthier and Duhamel Groups

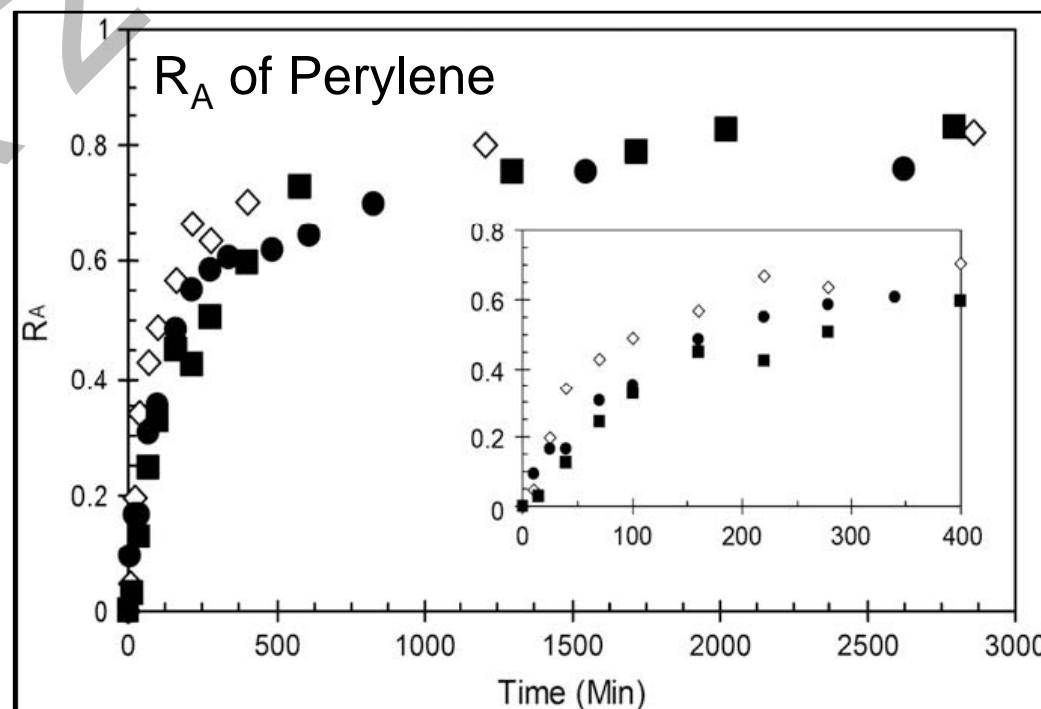
# RELEASE KINETICS

- Fluorescence Resonance Energy Transfer (FRET) experiments
  - Characterize release kinetics of hydrophobic probes from arborescent micelles

$$R_A = \frac{I_A(t) - I_A(0)}{I_A(\infty) - I_A(0)}$$

$R_A$  = Normalized experimental fluorescence intensity ratio (time-dependant)

$I_A$  = emission intensity at start (0), given time (t), and at equilibrium ( $\infty$ )



# RELEASE KINETICS

- Fluorescence quenching experiments
  - Using **Pyrene** and ionic quencher thallium nitrate (**TINO<sub>3</sub>**)

**Stern-Volmer equation**

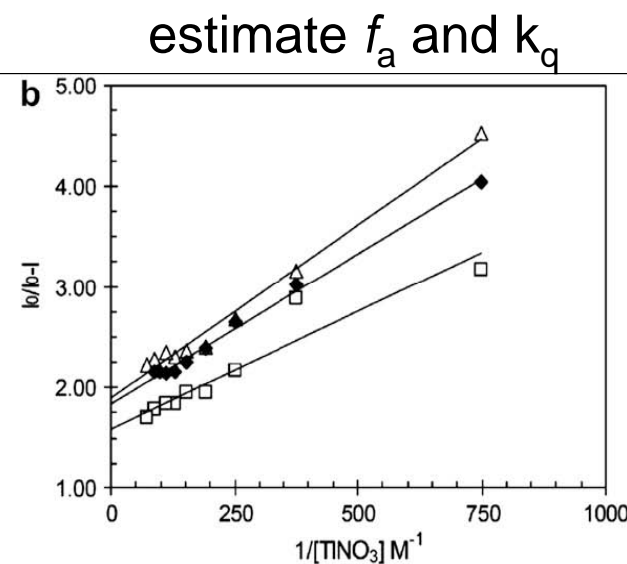
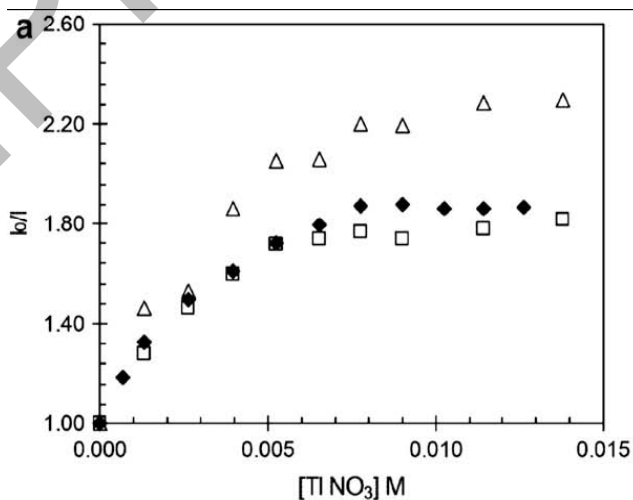
$$\frac{I_0}{I} = 1 + k_q \tau_0 [Q]$$

**Modified Stern-Volmer equation**

$$\frac{I_0}{I_0 - I} = \frac{1}{f_a} + \frac{1}{f_a k_q \tau_0 [Q]}$$

$f_a$  = fraction of pyrene accessible to quencher

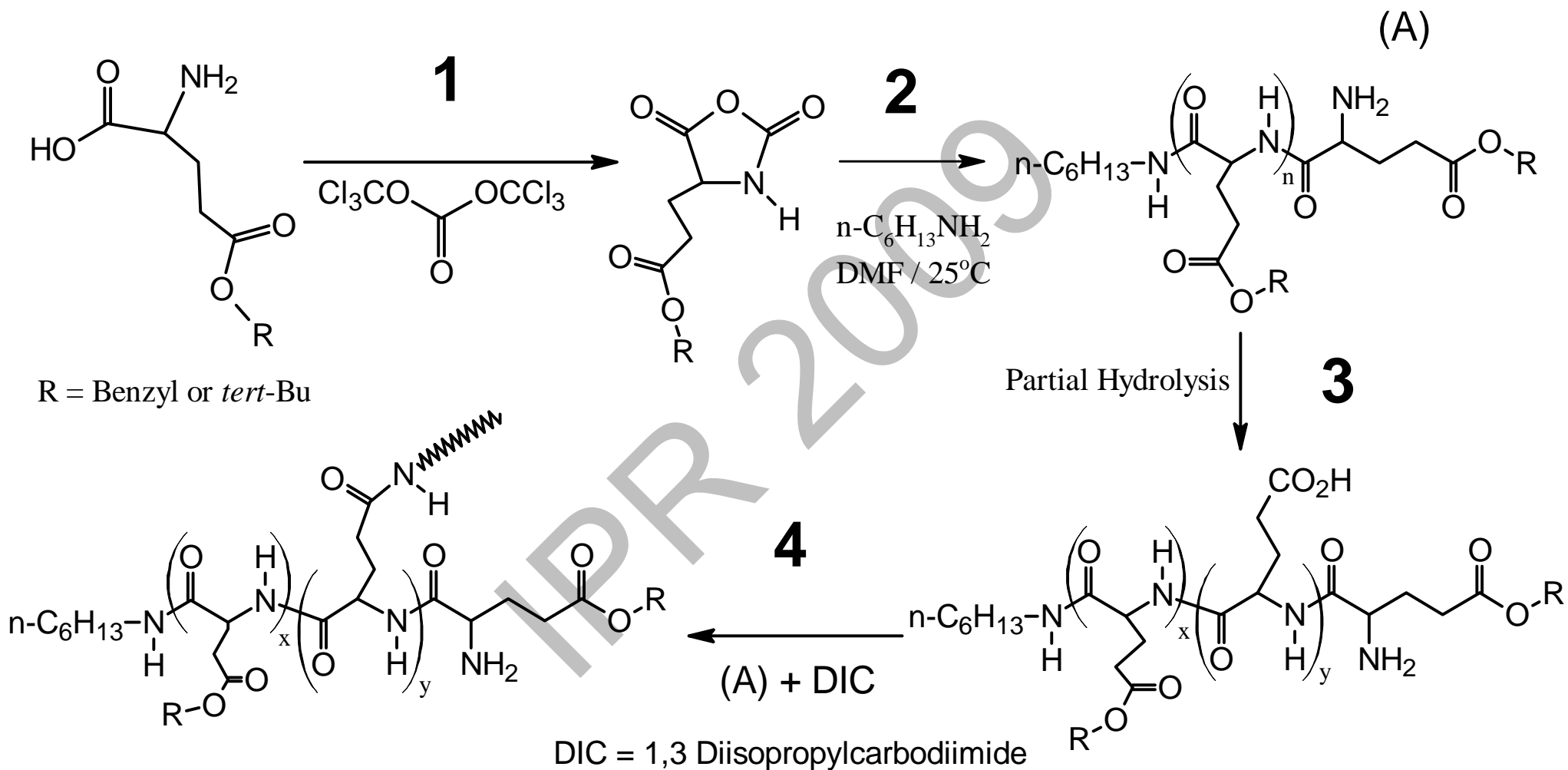
$k_q$  = bimolecular quenching constant



estimate  $f_a$  and  $k_q$

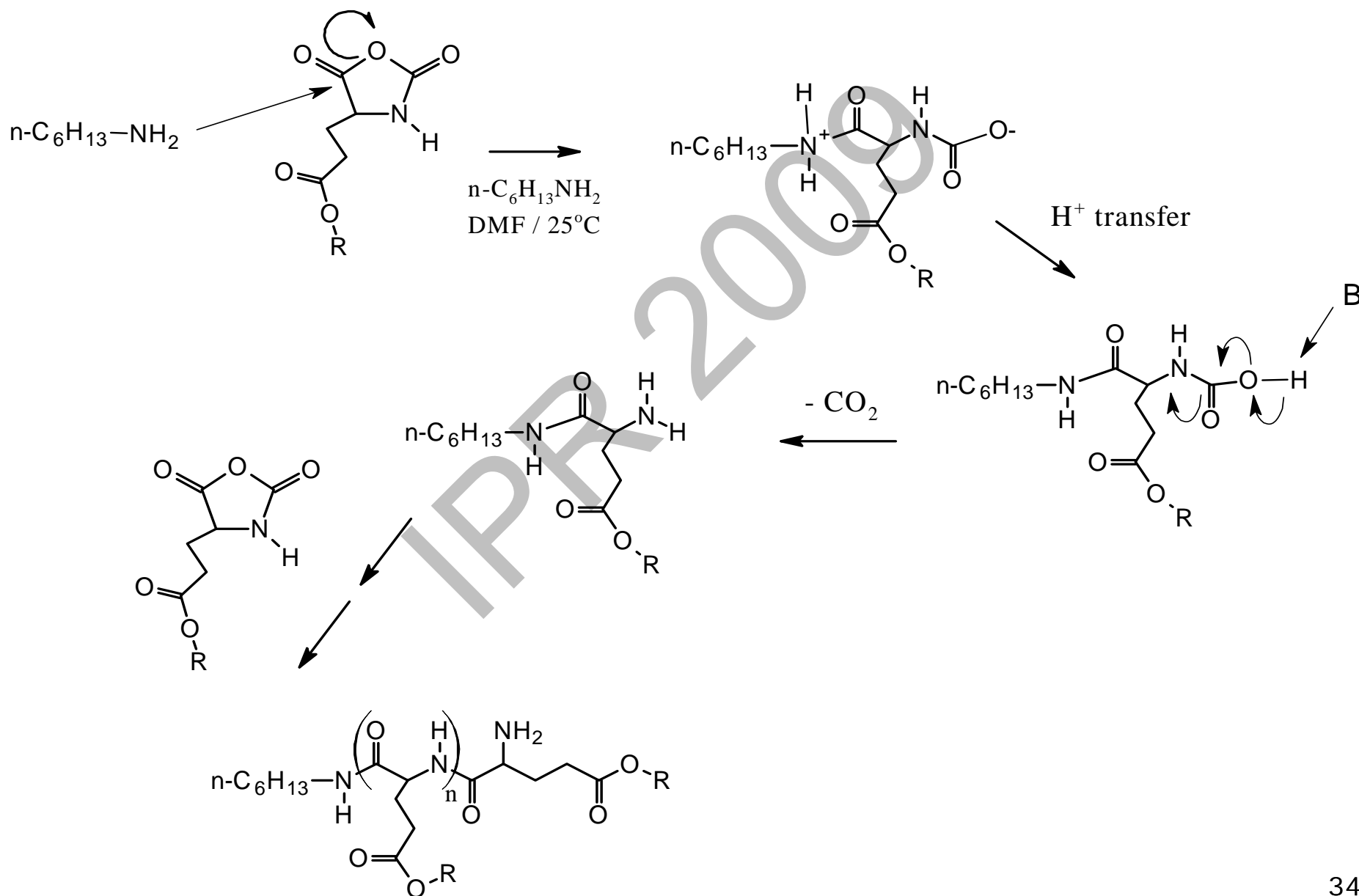


# Overall Synthetic Scheme

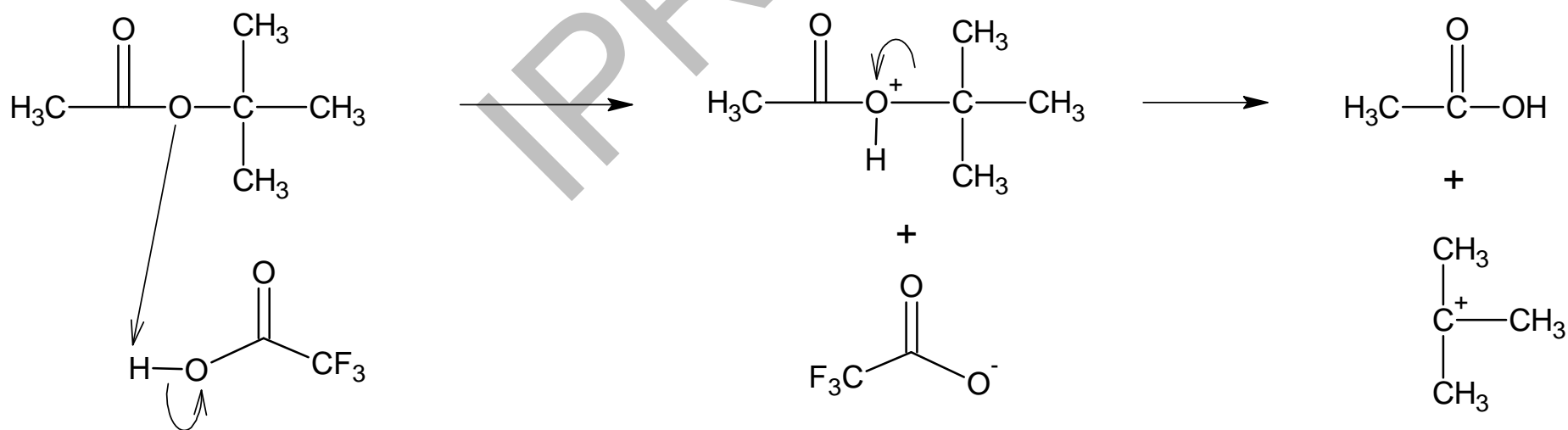
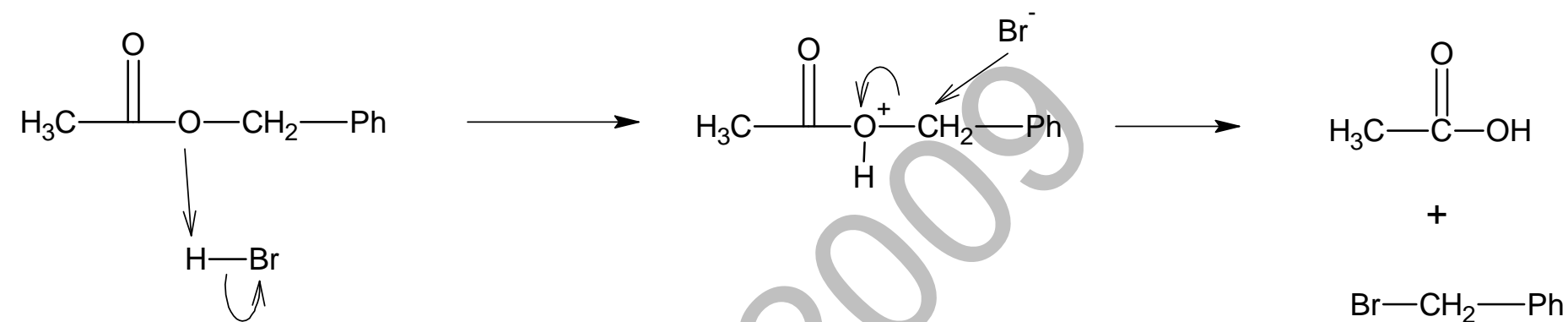


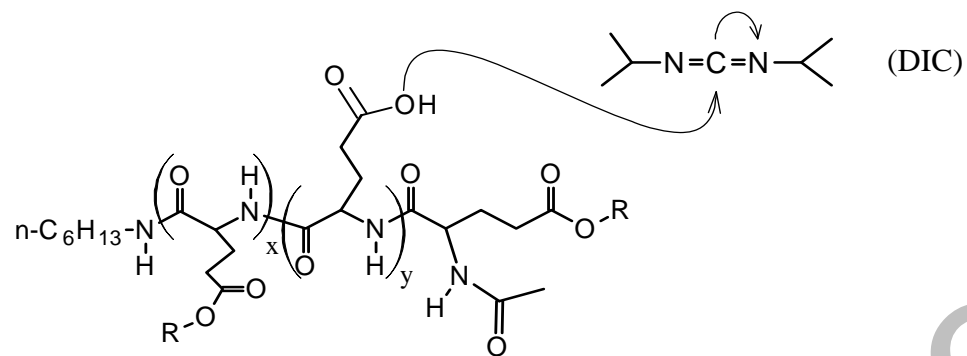
- Explain each step briefly so audience can get an idea of what is going on
- mention more details on synthesis later

# ROP MECHANISM

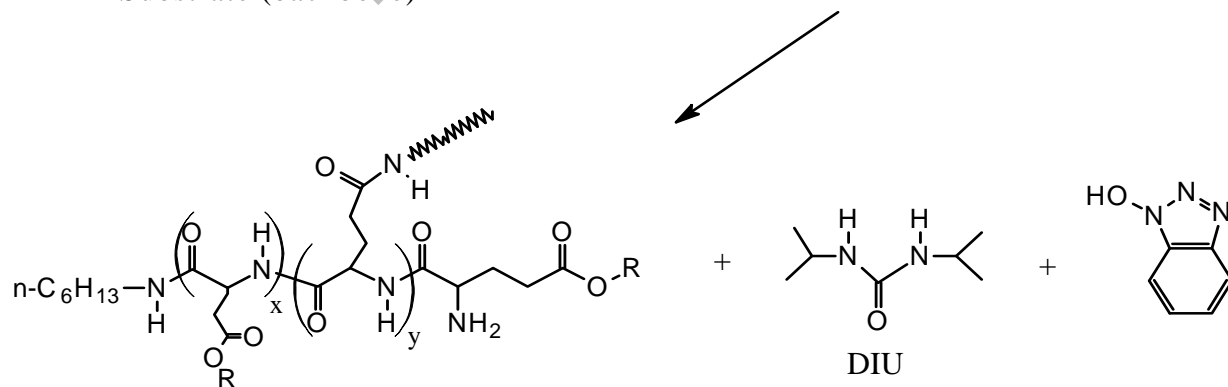
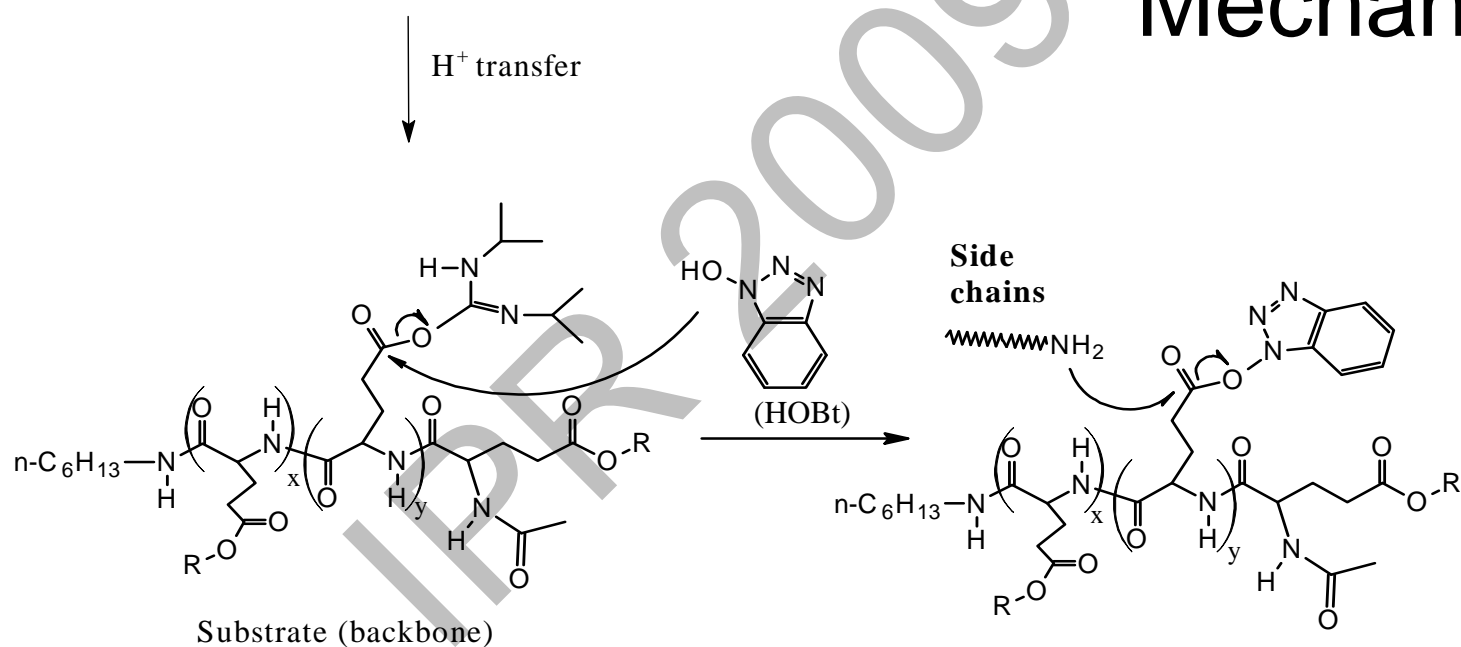


# Deprotection Mechanism





# Grafting Mechanism



# Grafting Side Reaction

