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MODELING OLEFIN POLYMERIZATION USING MONTE CARLO SIMULATION: DETAILED COMONOMER DISTRIBUTION



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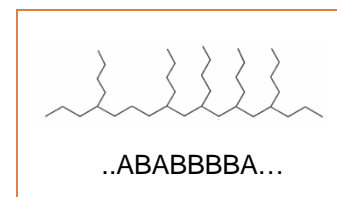
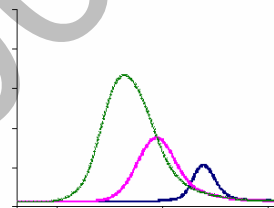
Outline



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Objectives

- To build a Monte Carlo model to describe the polymerization mechanisms of olefin copolymerization using single-site coordination catalyst
- To predict detailed polymer microstructure:
 - ▣ The complete chain length distribution;
 - ▣ Polydispersity;
 - ▣ Average comonomer (B) composition FB;
 - ▣ Comonomer composition distribution as function of chain length;
 - ▣ Monomer (A) and comonomer (B) segment length distribution as function of chain length;
 - ▣ Average triads distribution;
 - ▣ Triad distribution as a function of chain length;



Introduction

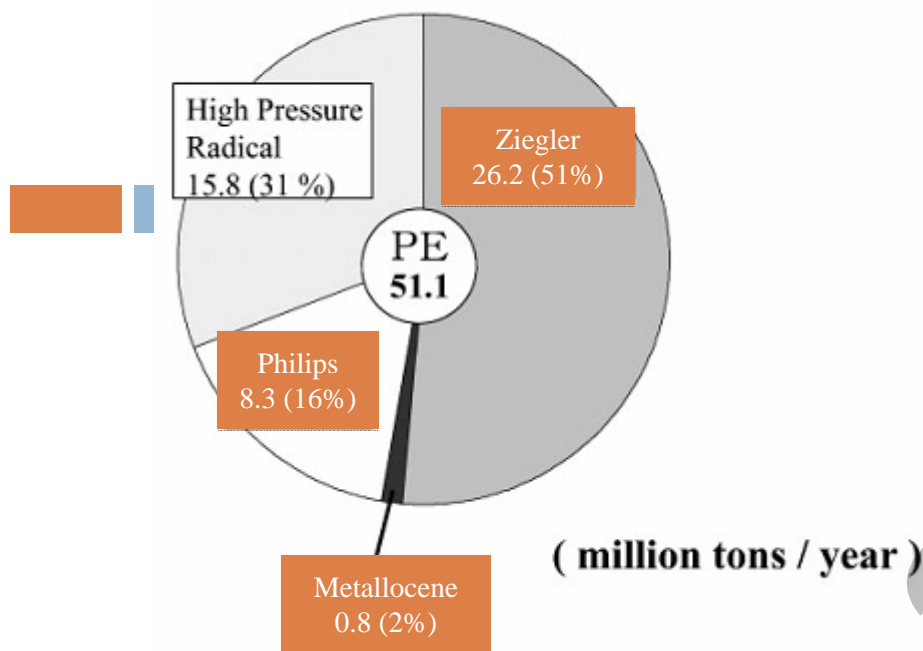
Stochastic Modeling & Monte Carlo Simulation

- Monte Carlo modeling is a convenient method of modeling polymer reaction mechanism and is widely used in polymerization systems (use of probabilities)
- It is useful specially when analytical solutions are not available



Polyolefins

- The basic properties of PE $(-\text{CH}_2\text{CH}_2-)_n$ are determined by the molecular structure
- Depends on degree of crystallinity, degree of polymerization, average molar mass and molar mass distribution



Catalysts for global PE production (Kashiwa, 2004)

Ziegler Titanium
Phillips Chromium

n Catalysts

n catalyst types available commercially in

Characteristics

Narrow molecular weight distribution

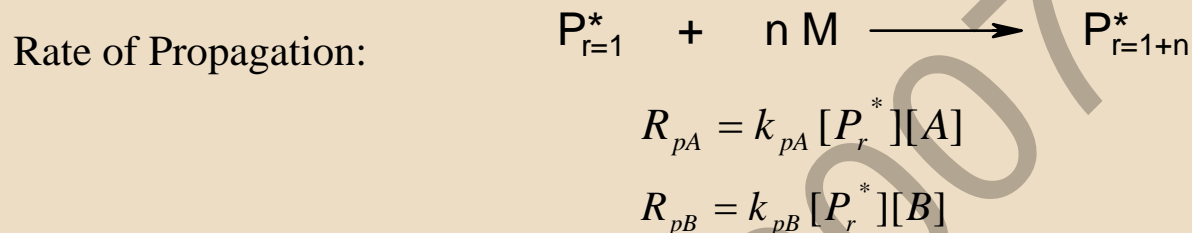
Co-catalyst required

Hydrogen as chain transfer agent

- Relatively narrow molecular weight distribution
- Aluminum alkyl co-catalyst required
- Hydrogen is used for molecular weight control

- Relatively broad molecular weight distribution
 - Co-catalyst not required
 - Hydrogen is not used for molecular weight control
-

Copolymer-Kinetic Equations-Probabilities Calculations



The probability of propagation:

$$p_p = \frac{(R_{pA} + R_{pB})}{(R_{pA} + R_{pB}) + (R_{tA} + R_{tB})} = \frac{(k_{pA} [P_r^*][A] + k_{pB} [P_r^*][B])}{(k_{pA} [P_r^*][A] + k_{pB} [P_r^*][B]) + (k_{tA} [P_r^*] + k_{tB} [P_r^*])}$$

Knowing that, the number average chain length:

$$r_n = \frac{R_p}{R_t} = \frac{(R_{pA} + R_{pB})}{(R_{tA} + R_{tB})}$$

Copolymer-Kinetic Equations-Probabilities Calculations

The probability of propagation is related to the number-average chain length:

$$P_p = \frac{R_p}{R_p + R_t} = \frac{1}{1 + \frac{R_t}{R_p}} = \frac{1}{1 + \frac{1}{r_n}} \approx 1 - \frac{1}{r_n}$$

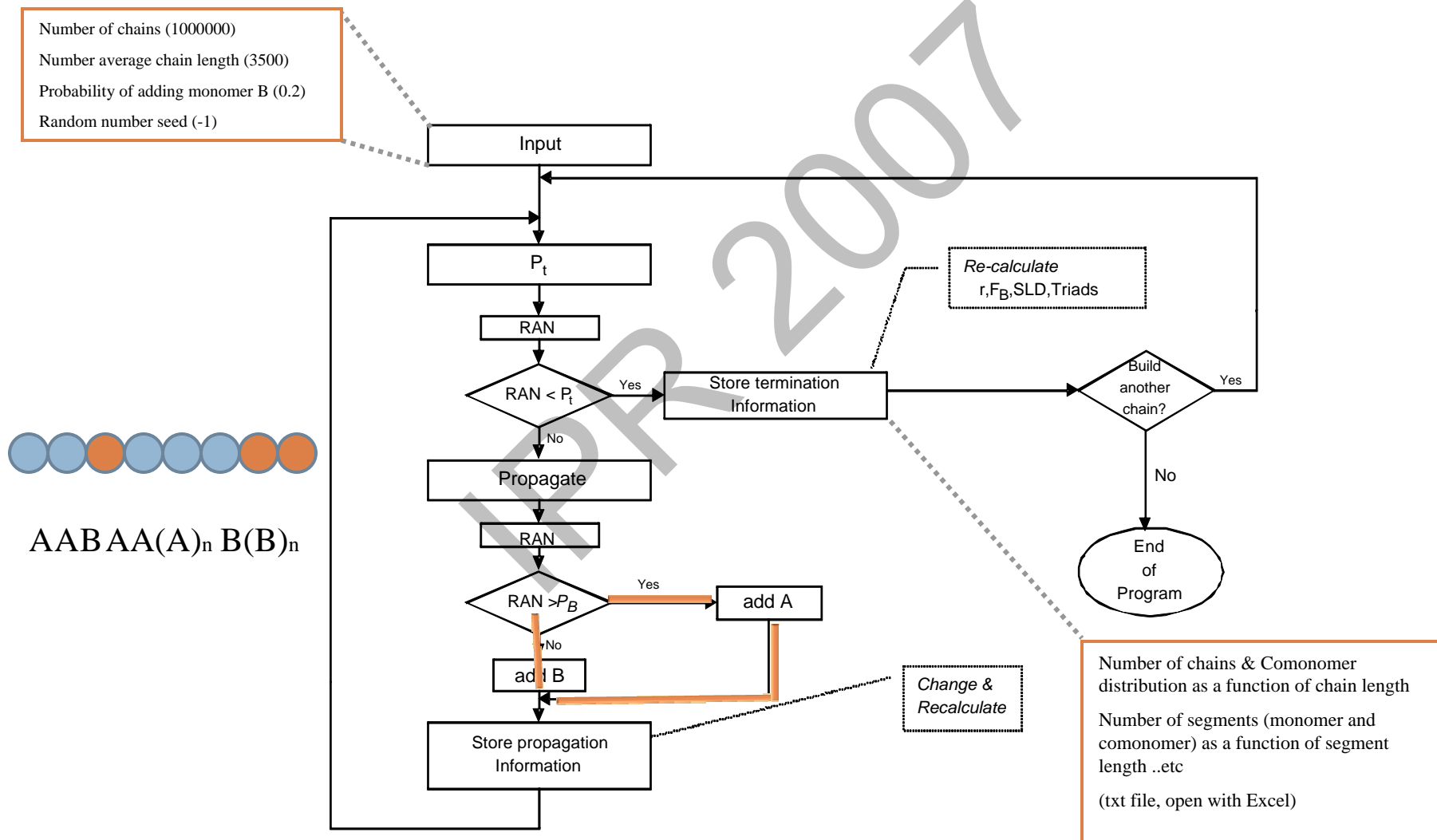
And, the probability of chain termination is related to the number-average chain length by:

$$P_t = \frac{R_t}{R_p + R_t} = \frac{(R_{tA} + R_{tB})}{(R_{pA} + R_{pB}) + (R_{tA} + R_{tB})} = \frac{k_{tA} [P_r^*] + k_{tB} [P_r^*]}{(k_{pA} [P_r^*] [A] + k_{pB} [P_r^*] [B]) + (k_{tA} [P_r^*] + k_{tB} [P_r^*])}$$

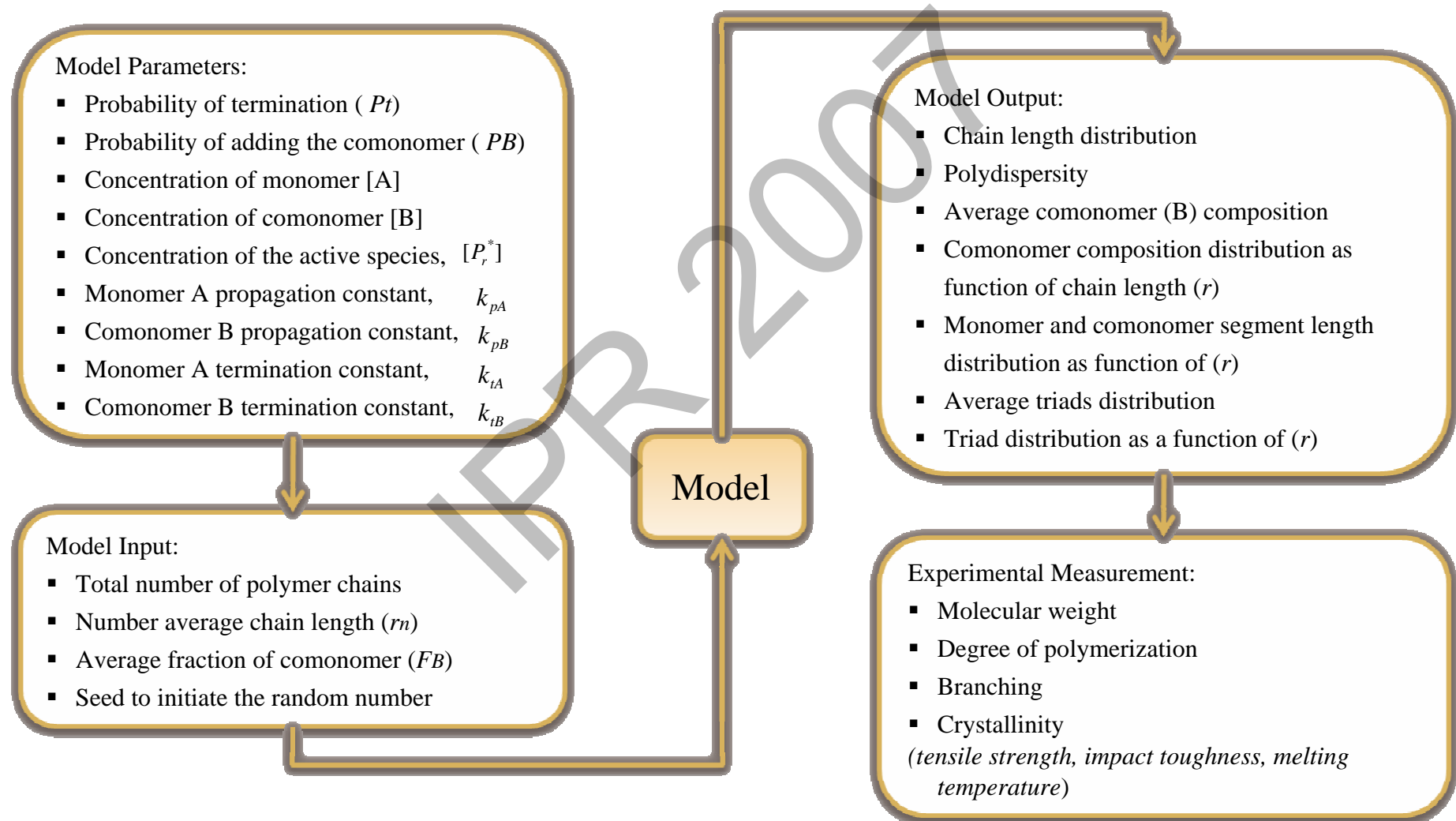
$$P_t = \frac{R_t}{R_p + R_t} = \frac{1}{\frac{R_p}{R_t} + 1} = \frac{1}{r_n + 1} \approx \frac{1}{r_n}$$



Copolymerization Schematic Flow Chart



Simulation Model Inputs-Outputs

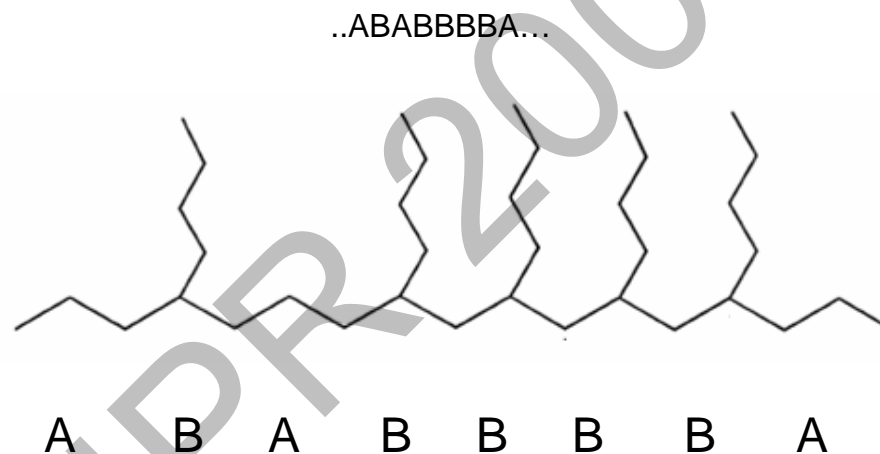


Triads Relative Intensities of ^{13}C -NMR Spectra

Nomenclature and ^{13}C -NMR Chemical Shift Assignments



Nuclear Magnetic Resonance spectroscopy is a very powerful technique for polymer characterization



A = ethylene
B = 1-hexene

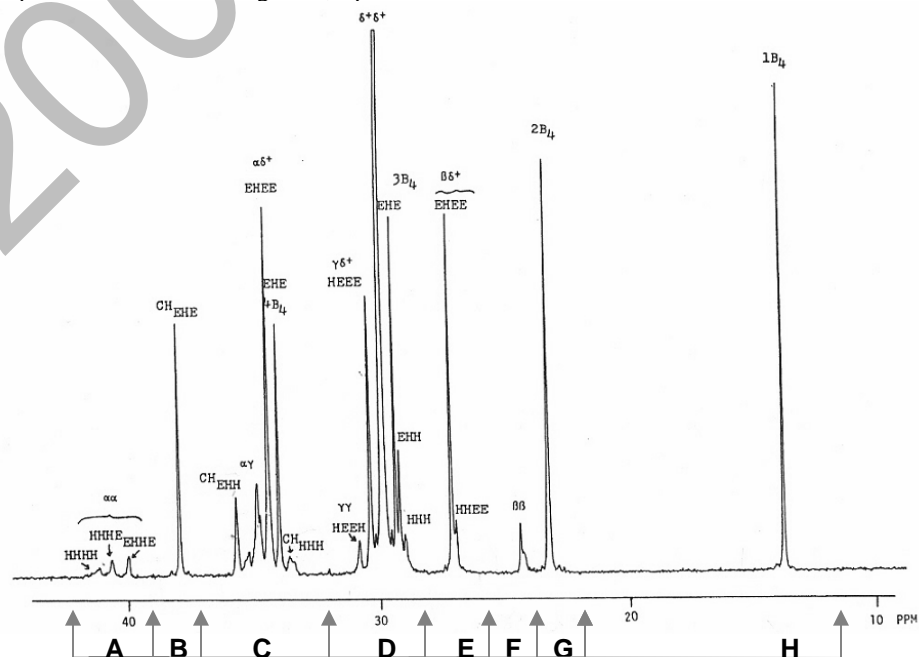
LLDPE = Linear Low Density Polyethylene

Tabulated Intensity Equations Chemical Shifts Assignments



Calculated intensities with respective chemical shift assignments ($r_n = 1008$ and $F_D = 5\%$)

Region	Range (ppm)		Contributing Carbons	
	from	to		
A	39.5	42	$\alpha\alpha$, Methylene	TA
B	38.1		(Methine) _{EHE}	TE
C	33	36	(Methine) _{EHH+HHE} , (Methine) _{HHH} , 4B ₄ , $\alpha\gamma$, $\alpha\delta$	TC
D	28.5	31	$\delta^+\delta^+$, 3B ₄ , $\gamma\gamma$, $\gamma\delta^+$	TI
E	26.5	27.5	$\beta\delta^+$	TE
F	24	25	$\beta\beta$	TF
G	23.4		2B ₄	TC
H	14.1		1B ₄	TH



**Chain Length &
Segment Length
Distribution**

Triad Distribution

**Simulation
Results**

Case Study

**Triads Relative
Intensities of ^{13}C -NMR
Spectra**

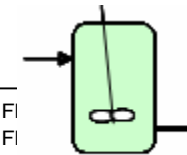
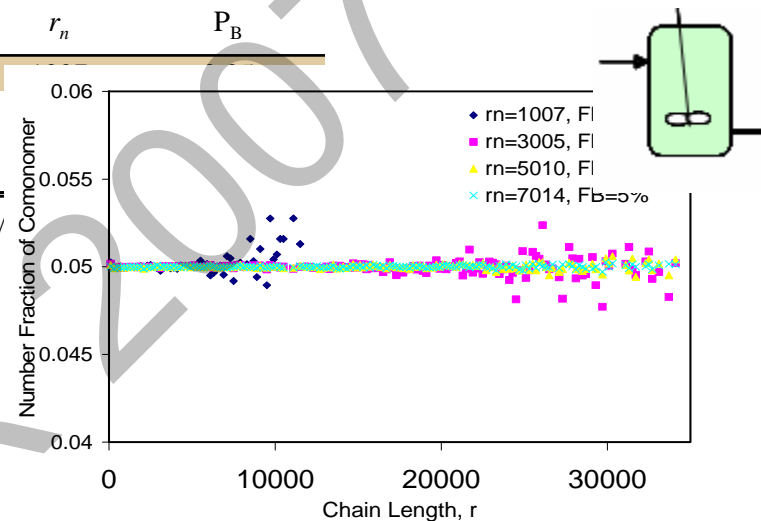
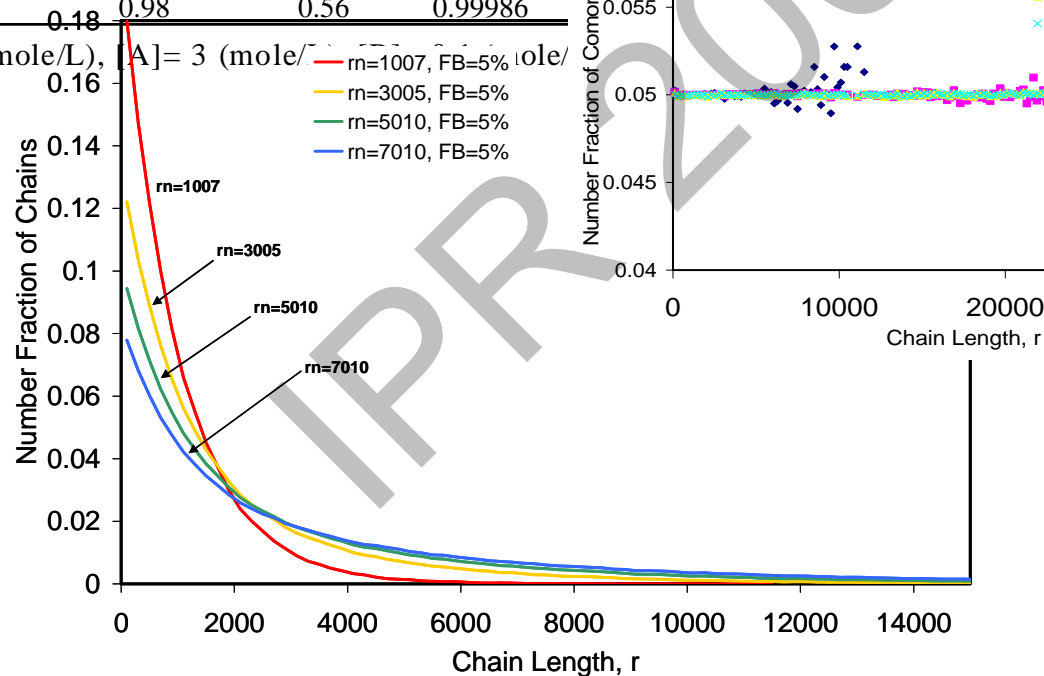
Number Fraction of Chains as a Function of Chain Length and Comonomer Distribution



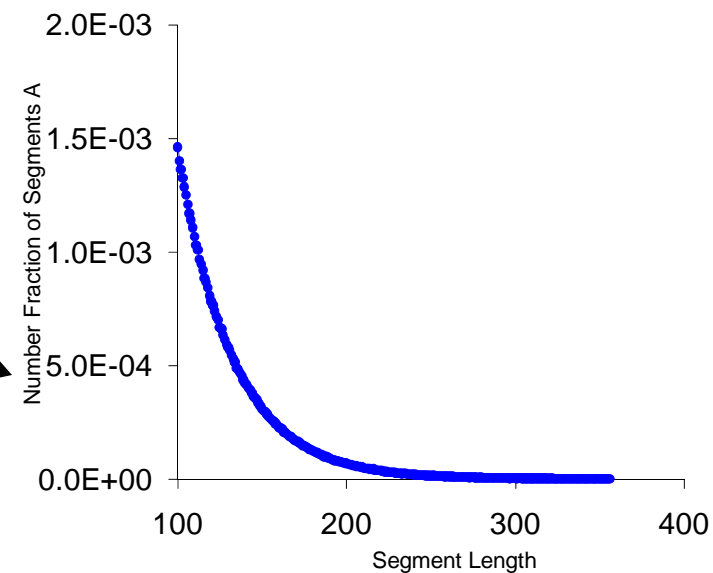
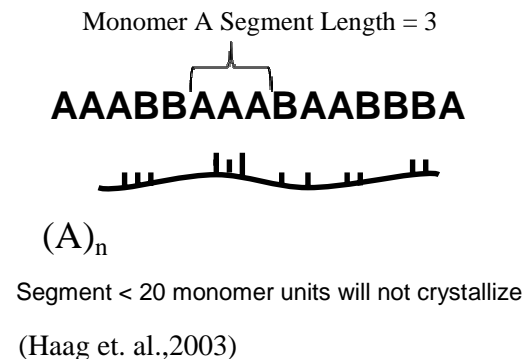
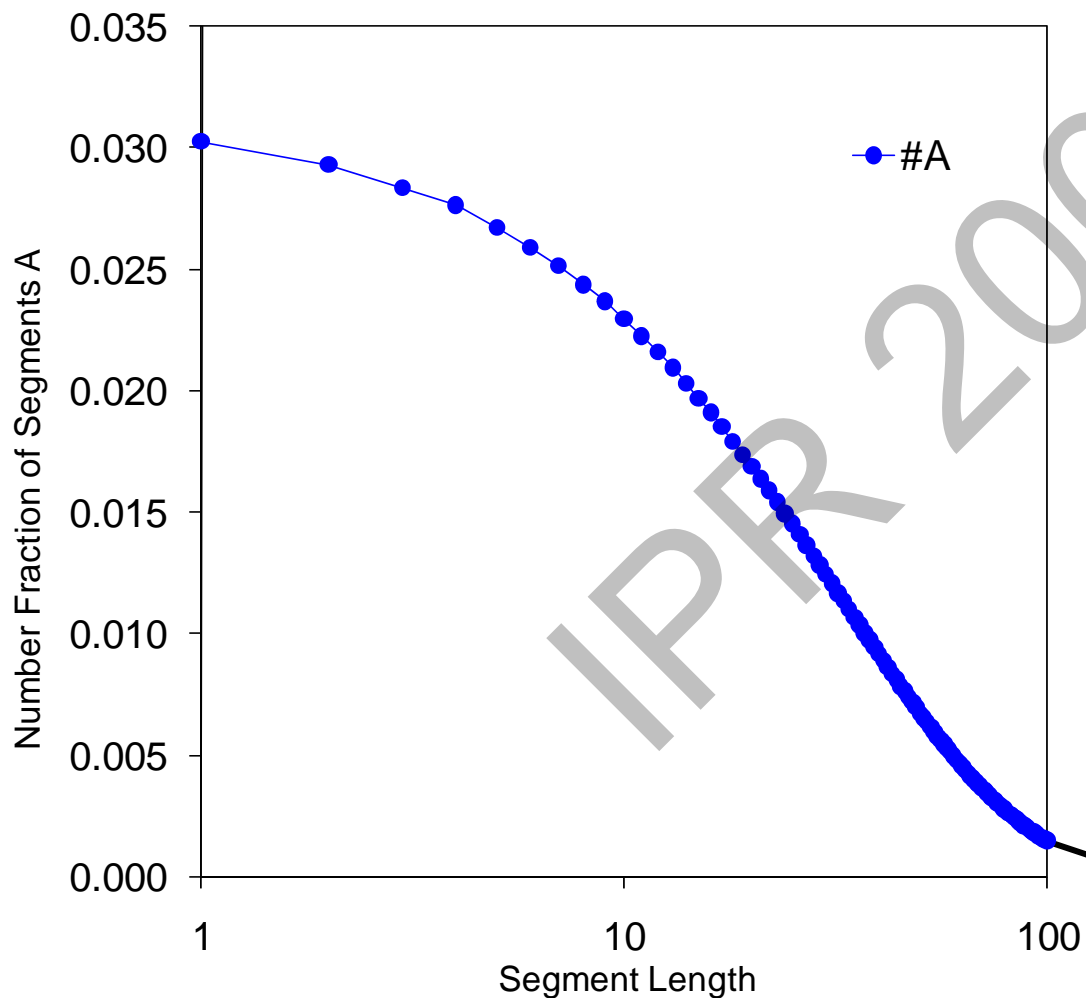
Four different simulations scenarios

k_{pA} (L/mole.s)	k_{pB} (L/mole.s)	k_{tA} (L/mole.s)	k_{tB} (L/mole.s)	Pp	r_n	P_B
1210	1910	2.70	1.10	0.99901		
2150	3395	1.35	0.91	0.99967		
3220	5080	1.23	0.80	0.99980		
3420	5340	0.98	0.56	0.99986		

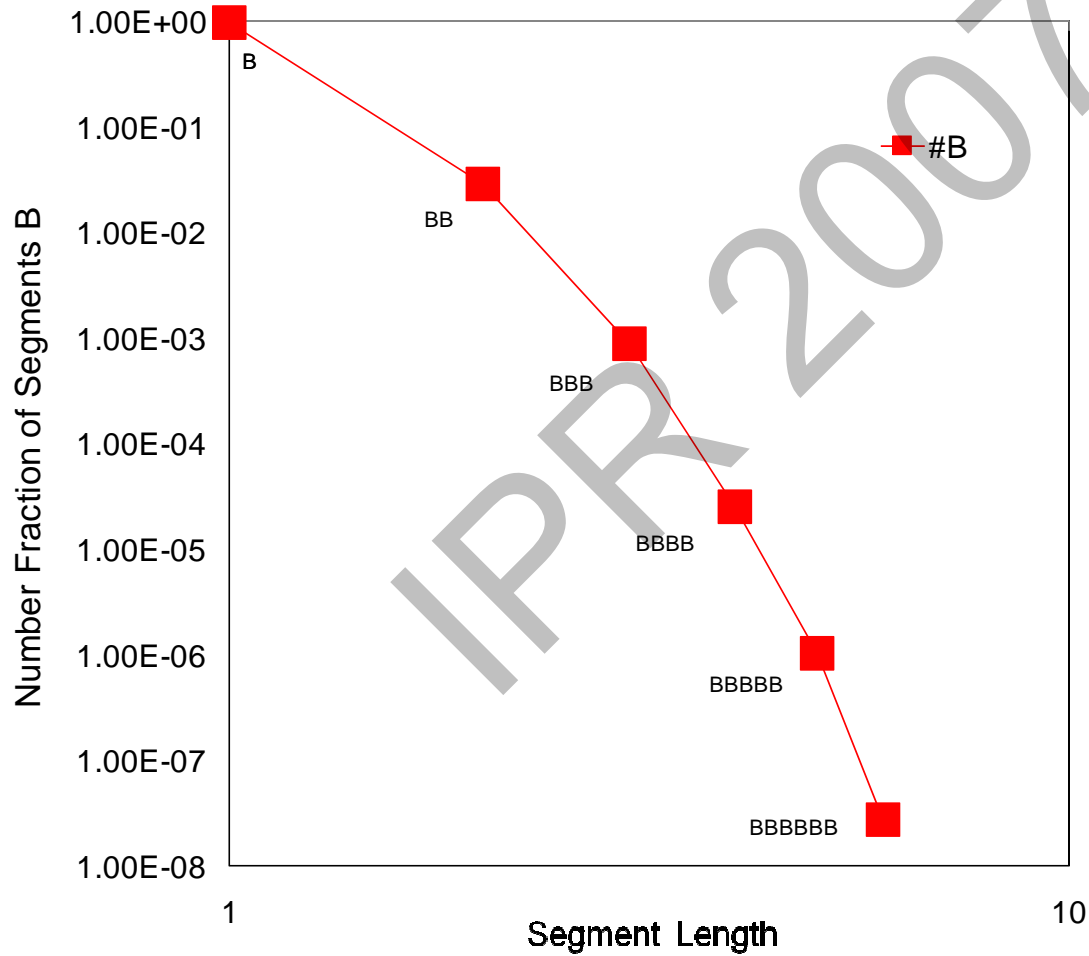
$[Pr^*] = 10 \times 10^{-6}$ (mole/L),
comonomer B



Fraction of Monomer Segment Length, A_n ($r_n=5004$, $F_B=3\%$)



Fraction of Comonomer Segment Length B_n ($r_n=5004, F_B=3\%$)

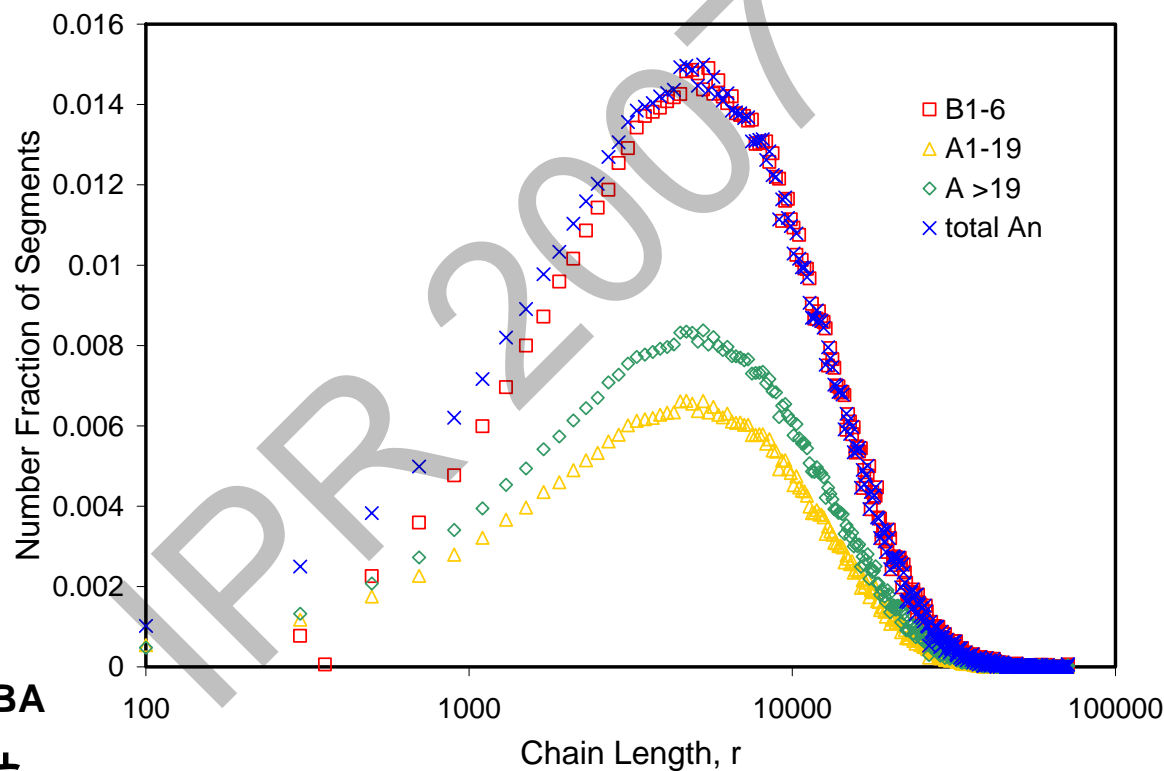


Monomer B Segment Length = 2

AAABBAABAABBBA



Distribution of Segments B_{1-6} , A_{1-19} , $A_{>19}$ and total A_n as a Function of Chain Length ($r_n=5004$, $F_B=3\%$)



AAABBAABAABBBA



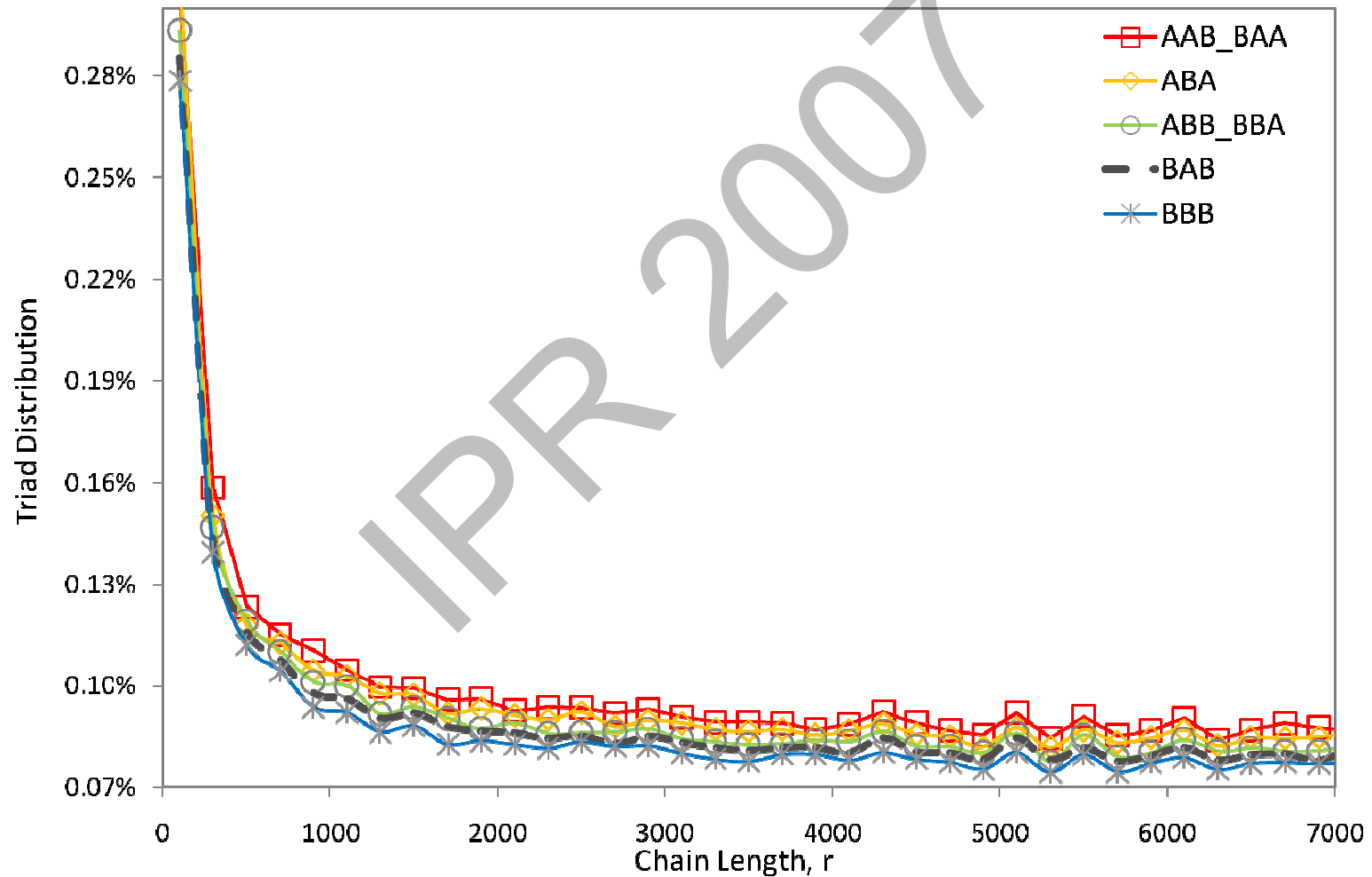
$(A)_n$

Segment < 20 monomer units will not crystallize

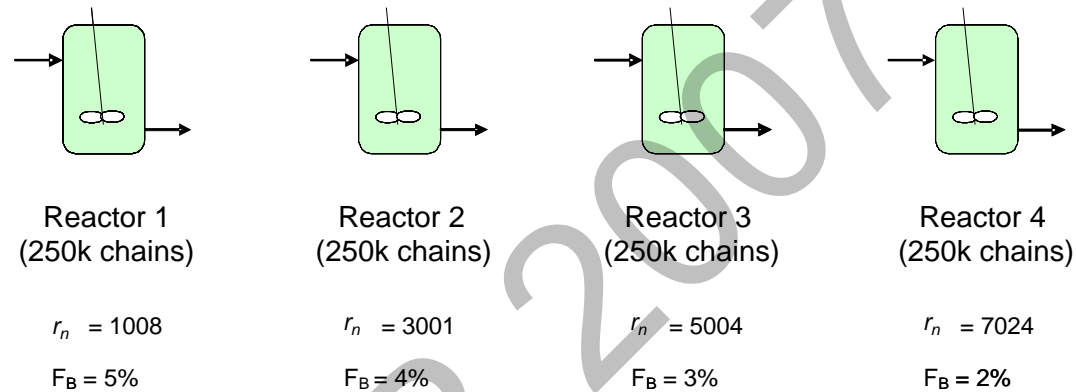
(Haag et. al.,2003)

Triad Distribution as a Function of Chain Length

($r_n=5004$, $F_B=3\%$)



Case Study



Kinetic parameters used in the case study (representing four different reactor conditions)

	k_{pA} (L/mole.s)	k_{pB} (L/mole.s)	k_{tA} (L/mole.s)	k_{tB} (L/mole.s)	P_t	P_p	r_n	P_B
R ₁	1210	1910	2.70	1.10	0.00099	0.99901	1008	0.05
R ₂	2160	2700	1.25	1.00	0.00033	0.99967	3001	0.04
R ₃	2200	2040	0.79	0.57	0.00020	0.99980	5004	0.03
R ₄	3900	2390	0.92	0.78	0.00014	0.99986	7024	0.02

$[Pr]^* = 10 \times 10^{-6}$ (mole/L), $[A] = 3$ (mole/L), $[B] = 0.1$ (mole/L)

Case Study



Effect of product mixing from reactor 1 to reactor 4 on the polymer parameters

	r_n	r_w	PDI	F_B
R ₁	1008	2001	1.98	5.0%
R ₁ +R ₂	2006	4994	2.49	4.5%
R ₁ +R ₂ +R ₃	3007	7777	2.59	4.0%
R ₁ +R ₂ +R ₃ +R ₄	4009	10513	2.62	3.5%

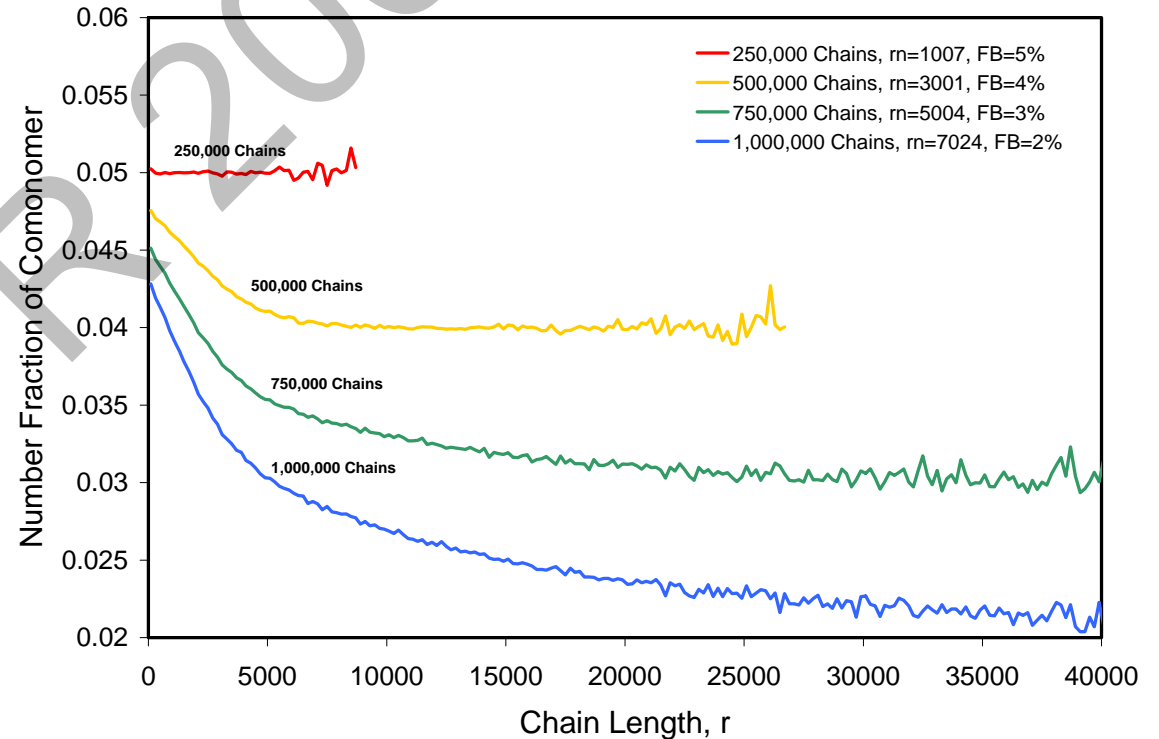
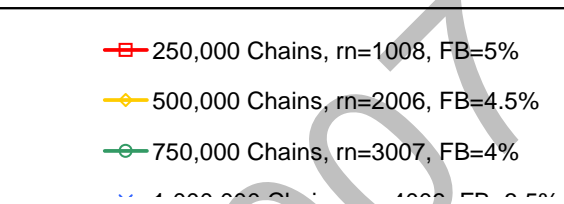
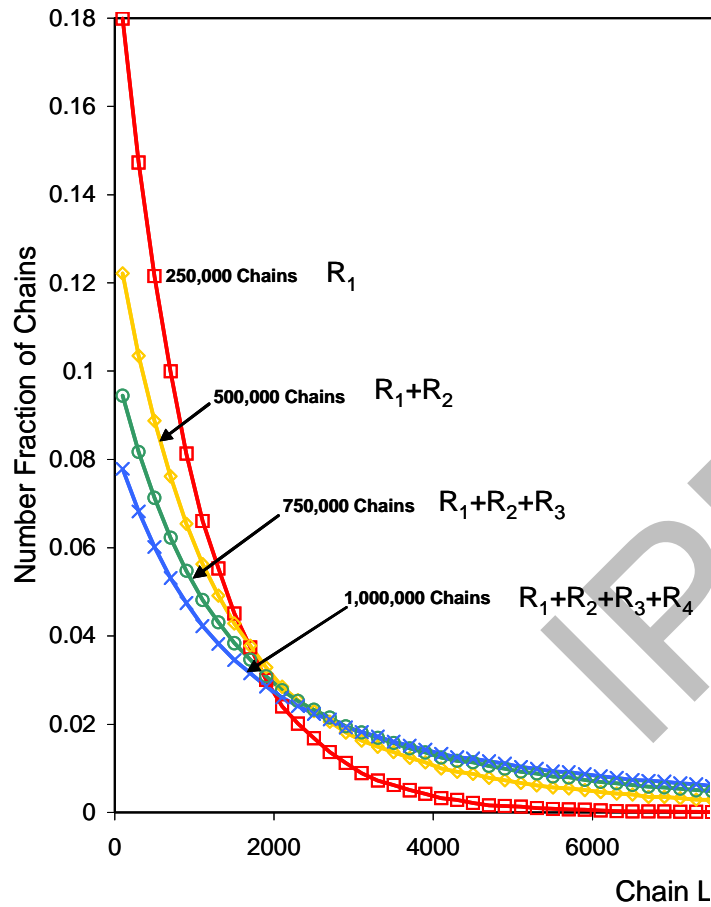
Kinetic parameters used in the case study after mixing (representing the products with $r_n = 1008$ to 4009 for mixed product from reactor 1 to reactor 4)

	k_{pA} (L/mole.s)	k_{pB} (L/mole.s)	k_{tA} (L/mole.s)	k_{tB} (L/mole.s)	P_t	P_p	r_n	P_B
R ₁	1210	1910	2.70	1.10	0.00099	0.99901	1008	0.05
R ₁ +R ₂	1100	1550	0.92	0.80	0.00050	0.99950	2006	0.045
R ₁ +R ₂ +R ₃	1308	1650	0.79	0.57	0.00033	0.99967	3007	0.04
R ₁ +R ₂ +R ₃ +R ₄	2220	2400	0.94	0.78	0.00025	0.99975	4009	0.035

$[Pr^*] = 10 \cdot 10^{-6}$ (mole/L), $[A] = 3$ (mole/L), $[B] = 0.1$ (mole/L)

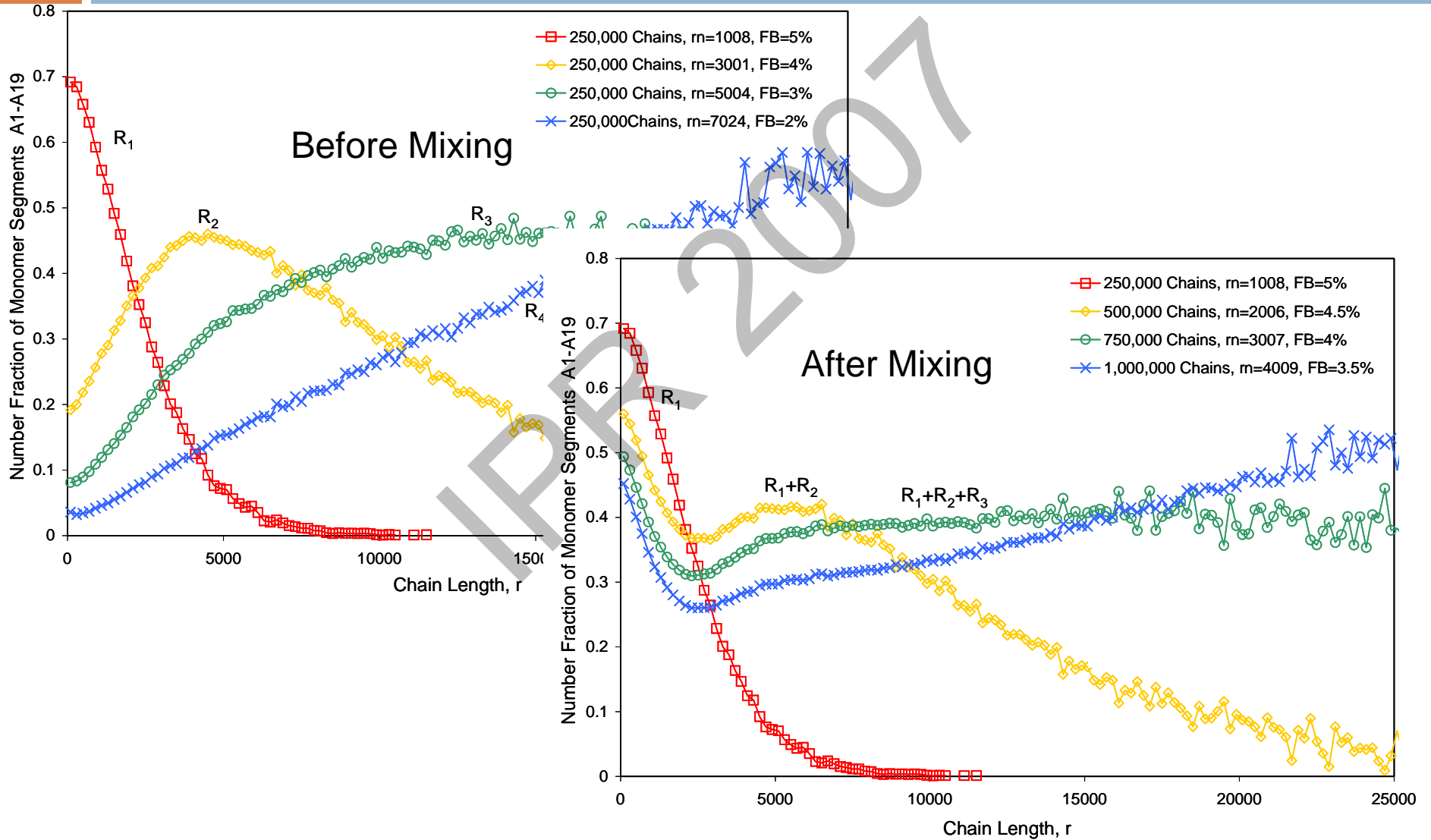
Case Study

From reactor 1 with 250,000 chains to reactor 4 with 1,000,000 chains



Case Study

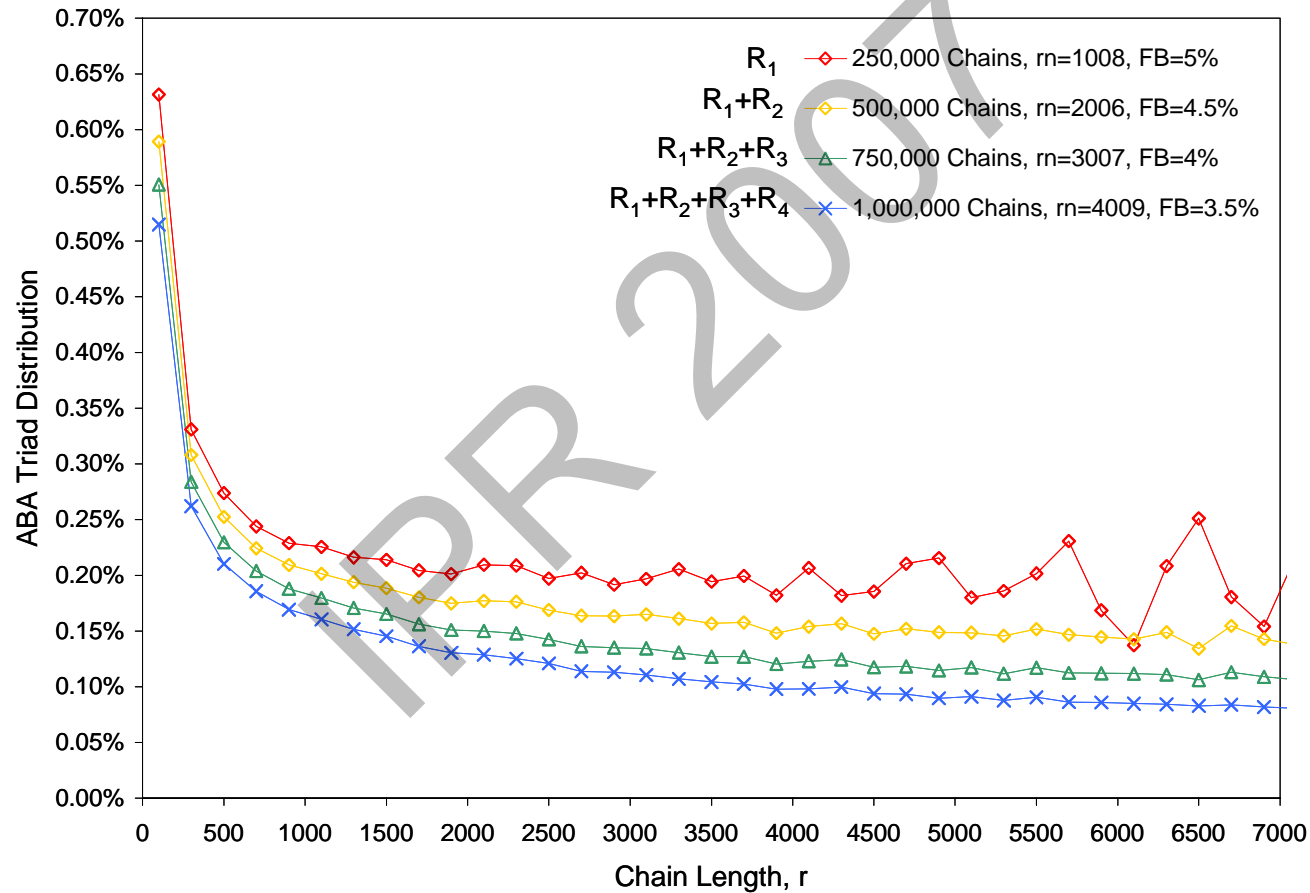
Number fraction of monomer segments for A_1 to A_{19} from reactor 1 with 250,000 chains to reactor 4 with 1,000,000 chains as function of chain length



Case Study

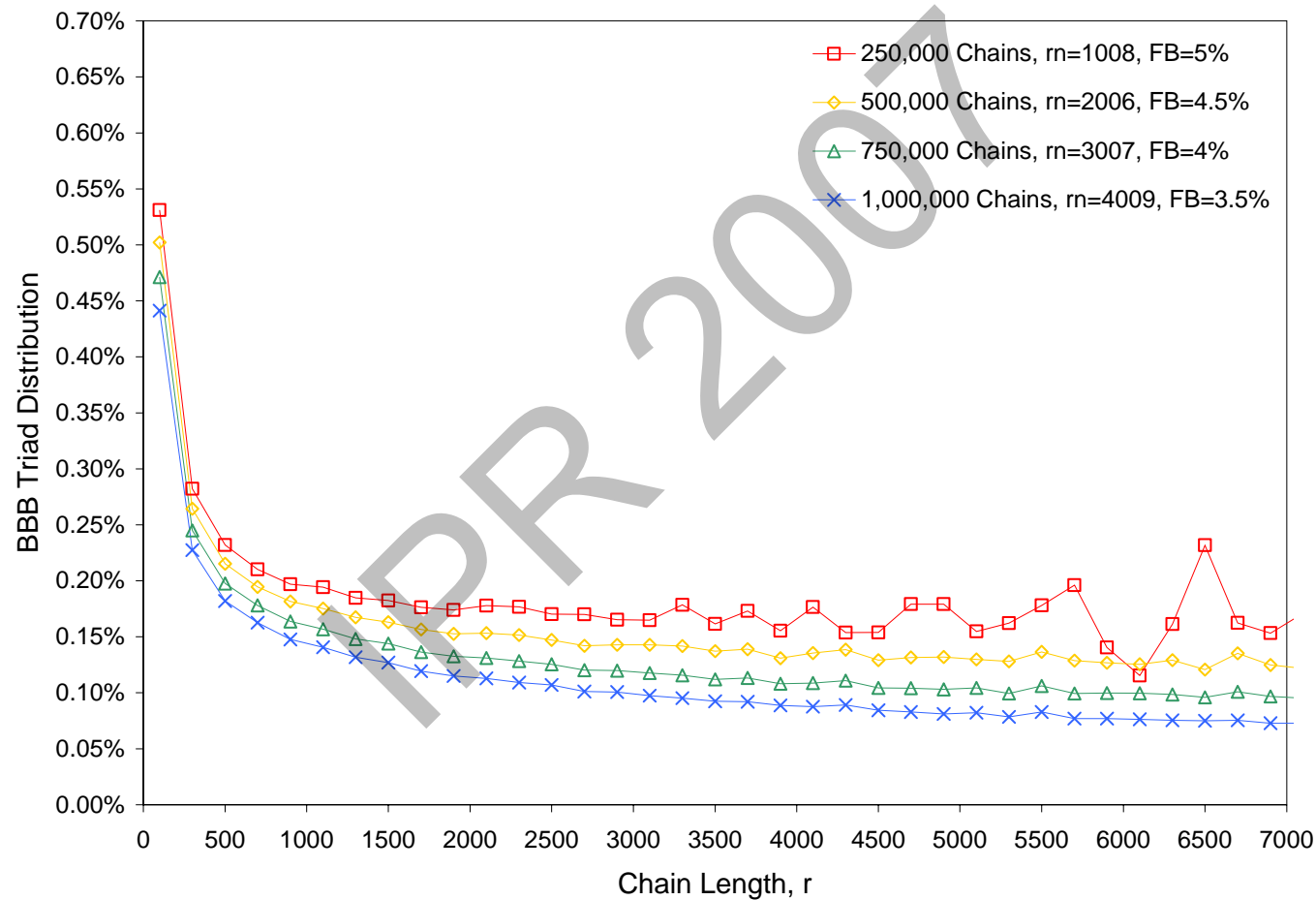


Triad Distribution



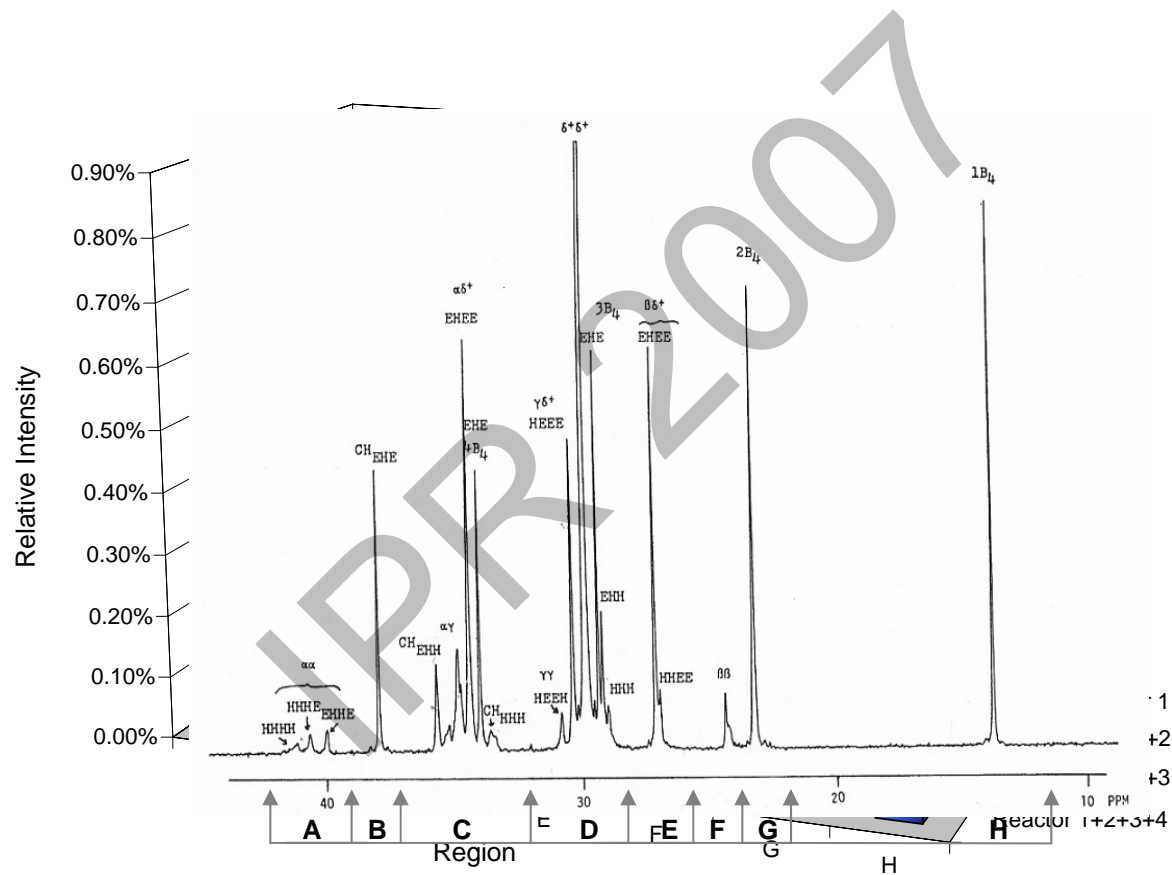
ABA triad distribution from reactor 1 with 250,000 chains to reactor 4 with 1,000,000 chains as function of chain length (varying r_n , F_B)

Case Study



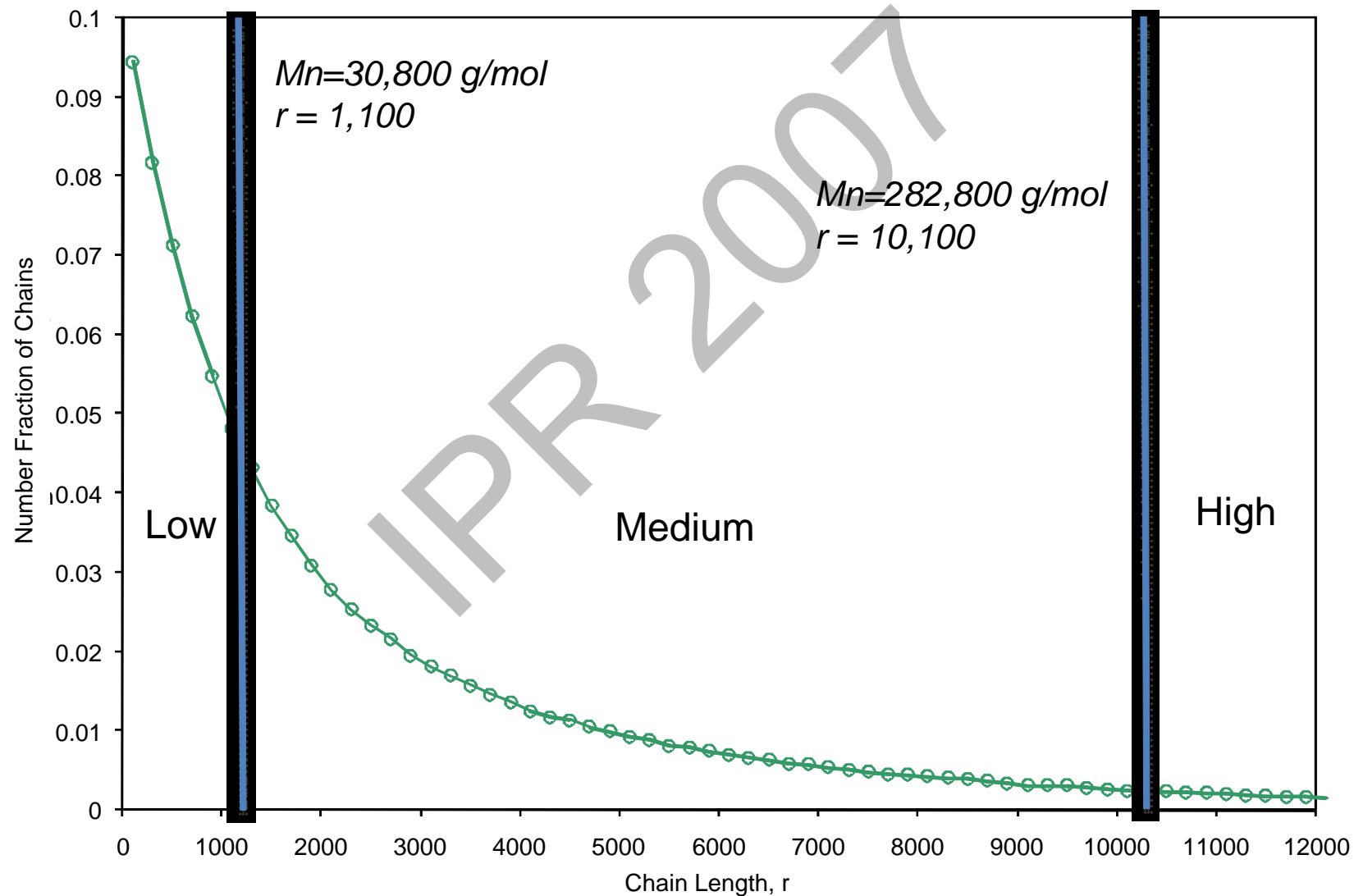
BBB triad distribution from reactor 1 with 250,000 chains to reactor 4 with 1,000,000 chains as function of chain length (varying r_n , F_B)

Triads Relative Intensities



The relative intensities of ^{13}C -NMR spectra for the branching regions

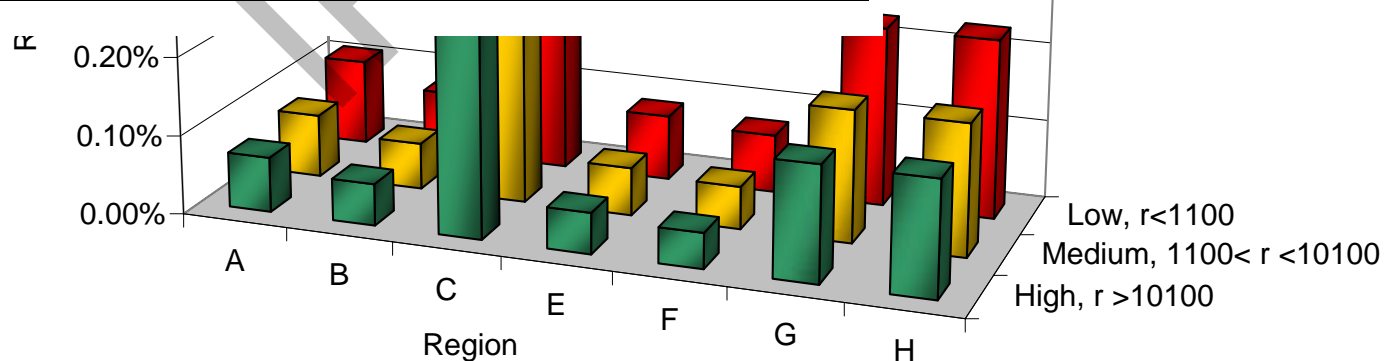
Fractionated Chain Length Population



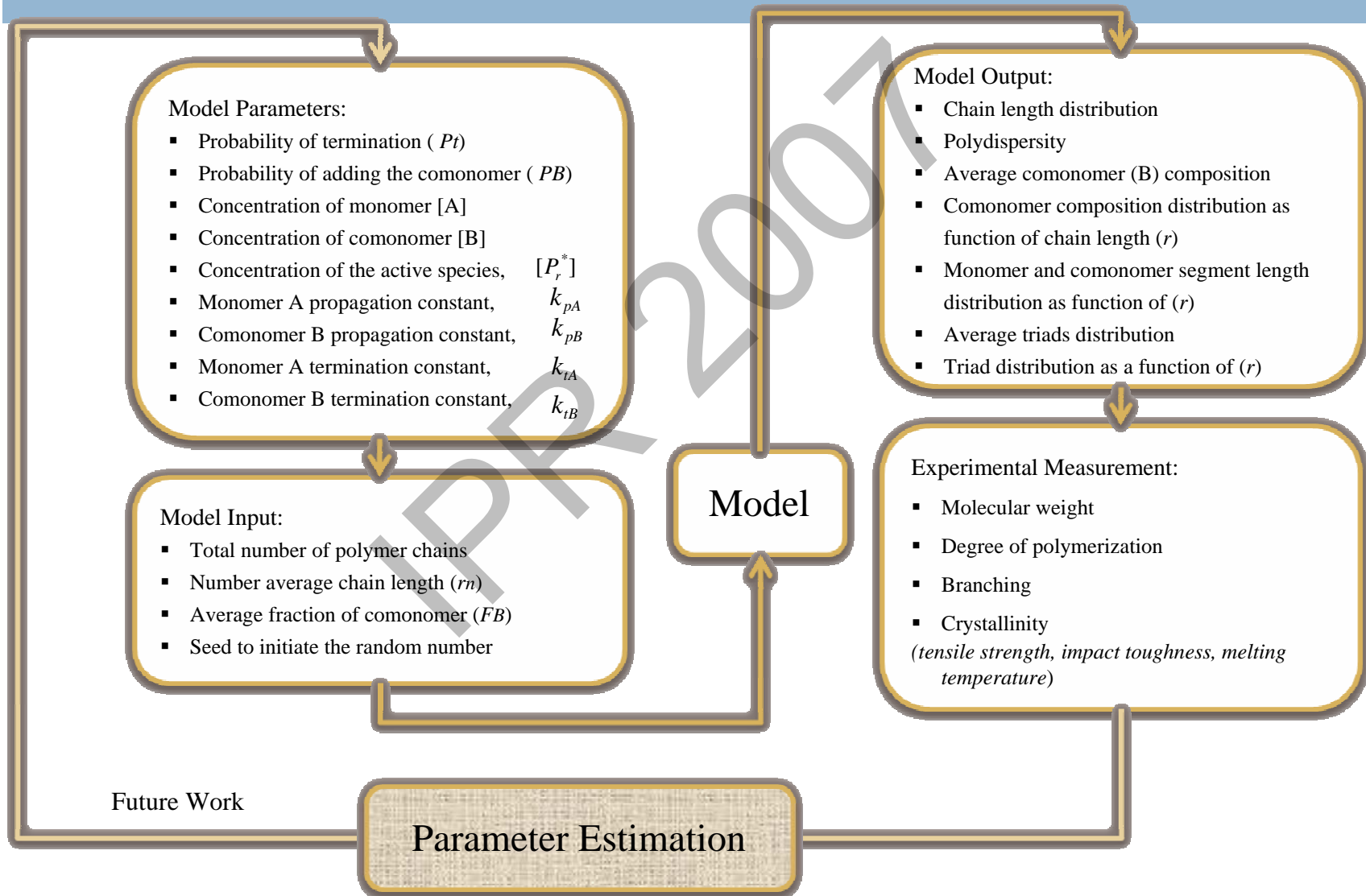
Triads Relative Intensities

Table-7 Fractionated population classes for the relative intensities with respective regions of reactor 4 after mixing the products from reactor 1 to 4 ($R_1+R_2+R_3+R_4$) according to chain length (r)

Region	Low $r < 1100$	Medium $1100 < r < 10100$	High $r > 10100$
A	0.109%	0.080%	0.070%
B	0.080%	0.059%	0.053%
C	0.605%	0.446%	0.381%
D	98.595%	98.964%	99.107%
E	0.083%	0.061%	0.052%
F	0.074%	0.054%	0.045%
G	0.227%	0.167%	0.146%
H	0.227%	0.167%	0.146%



Future Research Work



Conclusions

- ❧ The comprehensive Monte Carlo model which is developed and tested in this research work was able to describe the detailed comonomer distribution in the copolymerization of ethylene- α -olefins.
- ❧ For copolymerization, the model was able to predict chain length and comonomer sequence distribution with great detail.
- ❧ In the case of copolymerization the fraction of monomer B incorporated in the polymer chain was not dependant on the size of the chain, which is in agreement with the polymerization mechanism.
- ❧ The input information for running the model can be obtained from experimental polymer analysis or through the reaction kinetics.

Conclusions

- ☞ The input information is used to calculate model probabilities that are then used to determine each event of the polymerization mechanism.
- ☞ The model demonstrated great ability in predicting the detailed segment length distribution as a function of chain length, as well as the relative intensity for the peaks of the ^{13}C -NMR.
- ☞ This is a powerful tool to explore the chemical composition of the polymer in more detail.
- ☞ Knowing the segment lengths and triads distribution as a function of chain length is an advantage that allows us to study the polymerization mechanism and the properties of the polymer.