



Mass-based

Light

Weight

K. M. E. Stewart and A. Penlidis Institute for Polymer Research, Department of Chemical Engineering, University of Waterloo



Stewart, K. M. E. and A. Penlidis. "Designing Polymeric Sensing Materials: What are we Doing Wrong?" Polymers for Advanced Technologies 130, 42 (2016) doi:10.1002/pat.3893. Stewart, K. M. E. and A. Penlidis. "Evaluation of Polymeric Nanocomposites for the Detection of Toxic Gas Analytes" Journal of Macromolecular Science: Pure and Applied Chemistry 53, 10 (2016) 610-618. Stewart, K. M. E. and A. Penlidis. "Sensor Array for Volatile Organic Compounds based on Doped Poly (2,5-dimethyl aniline)" Macromolecular Symposia 370 (2016) 120-127.

Application-driven Sensing Material Design for Volatile Organic Compound Sensors

Target Analyte: Ethanol

Small Molecule Length: 3.8 Å



Alcohol Functional Group

- Polar backbone and/or functional groups
- Ideally capable of hydrogen-bonding
- Chains able to pack closely together to reduce interstitial spaces and thus, able to filter out larger analytes
- Potential dopants must be able to coordinate with both the polymer and ethanol

Application Requirements: Reduce Drinking and Driving

- Highly sensitive and selective detection of ethanol
- Operating Temperature up to 60 °C
 - ***** Therefore, need a T_g above 60 °C
- **Type of sensor: Unknown (in principle)**
 - Therefore, no restrictions on conductivity
 - Best to choose some conductive and some non-conductive polymeric materials (flexibility for sensor selection)
- Common interferent: Acetone
 - Both acetone and ethanol are similar in size and polarity
 - But acetone is unable to hydrogen bond like ethanol

Polar Molecule Dipole Moment: 1.66 D

Capable of Hydrogenbonding

Potential Sensing Materials

Poly (ethylene oxide)

Poly (vinyl acetate) (PVAc)

> Polyamide (PA)

Poly (lactic acid) (PLA)

Poly (ethylene terephthalate (PETE)

Poly (vinyl alcohol) (PVA)

> Polyaniline (PANI)

Poly (methyl methacrylate) (PMMA)

Poly (acrylic acid) (PAA)

> Poly (vinyl pyrrolidone) (PVP)

Poly (2,6-dimethyl-1,4-phenylene oxide) (PPO)

> Polypyrrole (PPy)

Evaluation of Sensing Materials

Four potential polymeric sensing materials were chosen •••• Two conductive (PPO, PPy), two non-conductive (PMMA, PVP) ••• All polymers were highly sensitive and selective for ethanol ******





ass Transition mperature, T _g (°C) ¹	Hildebrand Solubility Parameter (MPa ^{1/2}) ¹	Structure	Functional Groups	
-43	19.9	↓°~~>↓	-O-	
30	18.21		COOR	
50	23.02		CNOR	
57	21	T of	COOR	
67	21.9	Į.	COOR x2	
85	21.7		ОН	
100	22.2		NH	
105	22.8		COOR	
106	19.2	о 0-н	СООН	
128	25.6		CONR	
215	19.6	-[~	-O-	
270	25.15	N H H	NH	