

# A Paper Microfluidic Device for Monitoring Harmful Algal Blooms: A Small Solution to a Big Problem

**WATERLOO ENGINEERING**

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## Motivation and Purpose

- Harmful algal blooms (HABs):** excessive growth of algae in a body of water that produce toxins that are harmful to human and aquatic life
- The **Ontario Ministry of the Environment** estimates HABs in Lake Erie alone could cause up to **\$272 million in damages annually**



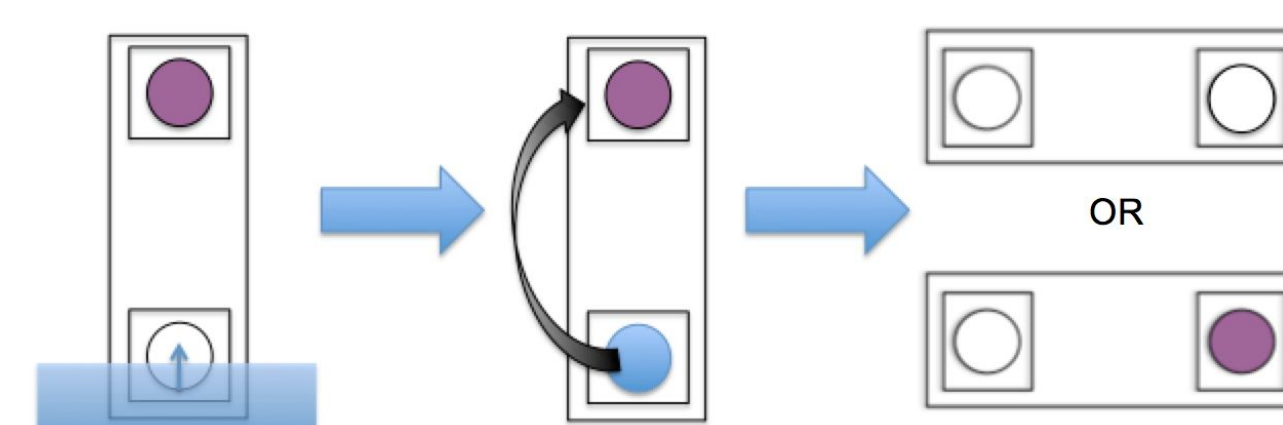
An aerial view and close up views of the algal blooms found in Lake Erie.

- Nitrogen species**, in the forms of **nitrite (NO<sub>2</sub>)** and **nitrate (NO<sub>3</sub>)**, have been identified as significant contributors to HAB formation
- Problem: Lack of public awareness and limited data collection**
- Solution: An easy to use, portable device** that can be distributed to **citizen volunteers** to raise public awareness and increase data collection to find areas most susceptible to algal bloom formation

## Design and Operation

### Nitrite sensor

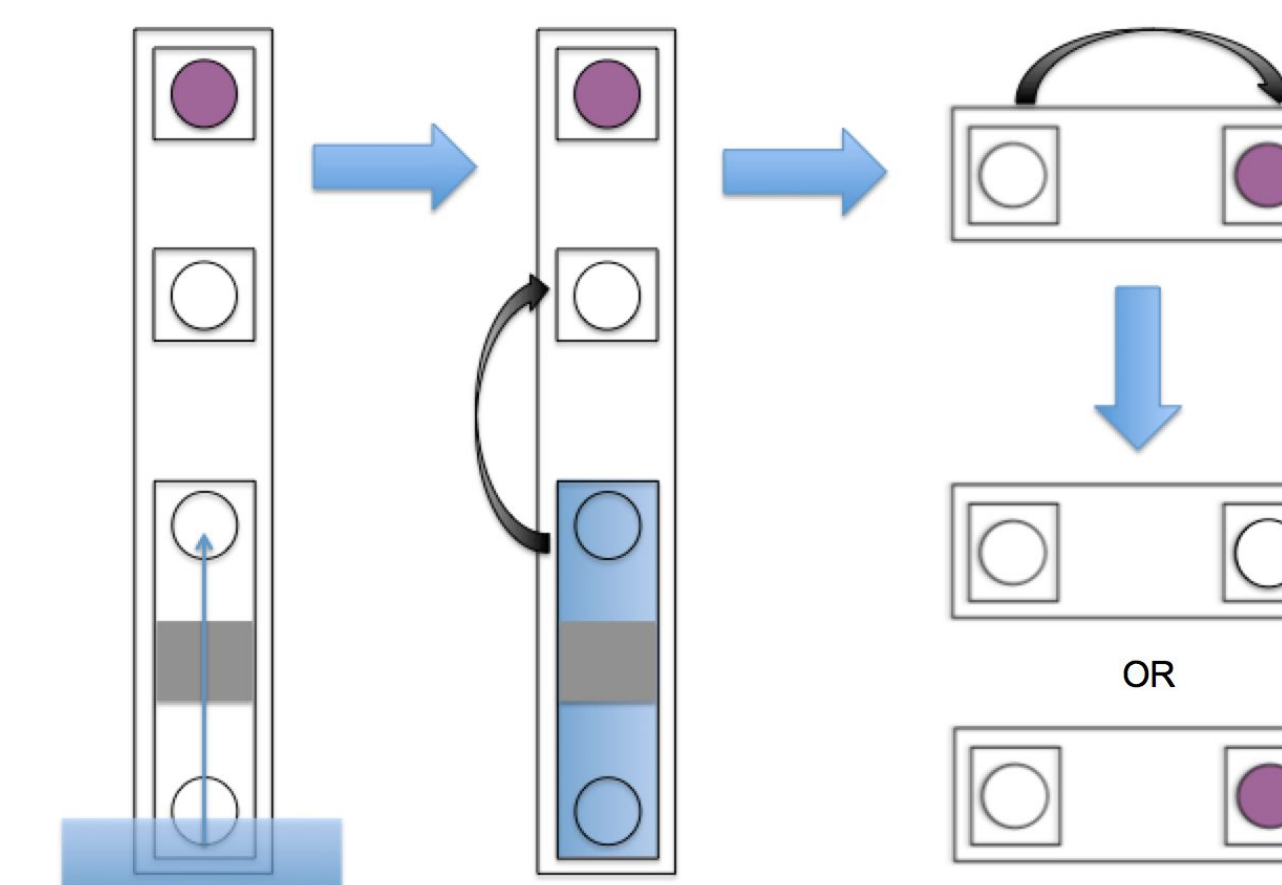
- 1,2-diaminoanthroquinone (DAQ)** is a dye that has been shown to **sense for nitrites** in solution
- DAQ is naturally **purple and turns clear** in the presence of nitrite
- An **acidic environment** is required for the reaction to take place
- DAQ is deposited onto the paper reaction well and a strong acid is deposited on the inlet



- The device can be dipped in the sample water, collecting water at the inlet and then transferred to the reaction well, resulting in a **colour change**

### Nitrate sensor

- Addition of a longer inlet area with deposited **zinc dust to reduce nitrate to nitrite**
- Uses same sensing mechanism as nitrite



- The device can be dipped in the sample water, which passes through the zinc filled channel and then transferred to an acidic well and follows the same operational procedure as the nitrite sensor

## Evaluation of Our Device

To evaluate the usefulness of our device, we compared its capabilities to alternative, current solutions

|                                      | Easy to use? | Inexpensive? | On-site? | Rapid detection? |
|--------------------------------------|--------------|--------------|----------|------------------|
| Taking samples to lab to analyze     | ✗            | ✗            | ✗        | ✗                |
| MPC Buoy (current commercial sensor) | ✗            | ✗            | ✓        | ✓                |
| Electrochemical techniques           | ✓            | ✗            | ✓        | ✓                |
| <b>Our paper microfluidic sensor</b> | ✓            | ✓            | ✓        | ✓                |

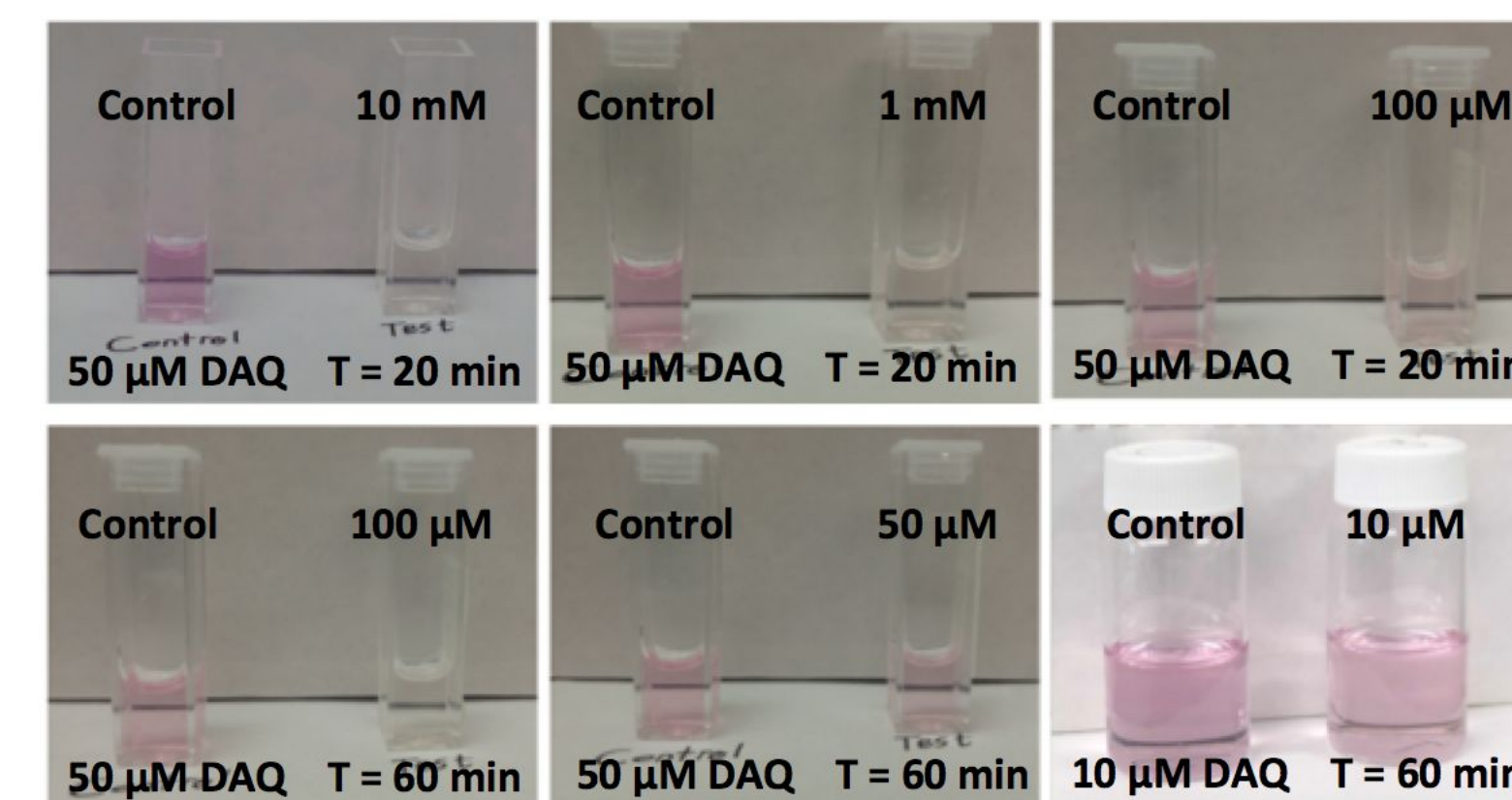
## Design Considerations

|                                       | Target                  | Ideal                | Optimal              |
|---------------------------------------|-------------------------|----------------------|----------------------|
| Limit of detection (NO <sub>2</sub> ) | 71 μM                   | 9 μM                 | 900 nM               |
| Limit of detection (NO <sub>3</sub> ) | 710 μM                  | 120 μM               | 2.4 μM               |
| Cost                                  | \$1.00                  | ~\$0.50              | < \$0.10             |
| Complexity of Use                     | Light training required | No training required | No training required |

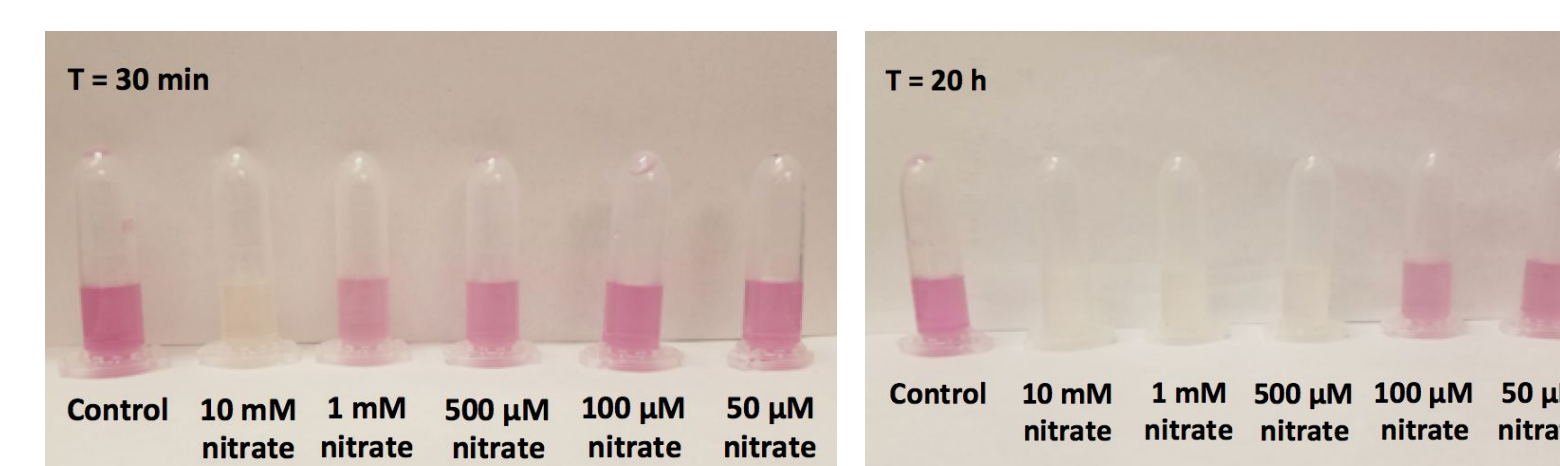
## Results

### In solution assay results

- Current **limit of detection for nitrite sensing** (by eye): 10 μM
  - Reaction time: 60 minutes



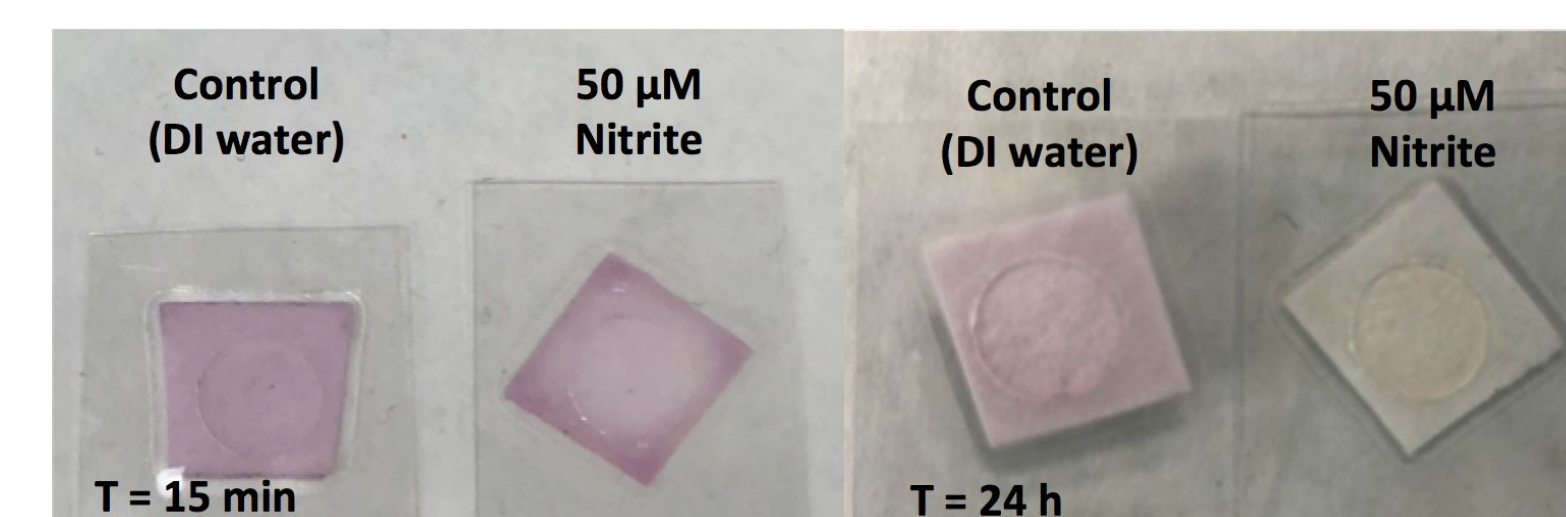
- Current **limit of detection for nitrate sensing** (by eye): 500 μM
  - Reaction time: 5 minutes for nitrate reduction; 20 hours to completion



- Nitrate conversion efficiency:** < 10% after 30 minutes; > 20% after 20 hours
- Cost of test:** > \$0.30 for a 1 mL test + shipping costs + lab technician labour

### Paper sensing results

- Current **limit of detection of the nitrite sensor** (by eye): 50 μM
  - Reaction time: < 15 minutes
  - Device mass: ~0.25 g



- Current **limit of detection of the nitrate sensor** (by eye): 1 M
  - Reaction time: 5 minutes for nitrate reduction, < 10 minutes for colour change
  - Device mass: ~0.4 g



- Cost of test including distribution:** < \$0.08

## Next Steps

- Improve **repeatability** and **reproducibility** of results by using **automation** for manufacturing
- Lower the **limit of detection** of sensors and demonstrate **selectivity** of the on-paper sensors
- Optimize **nitrate reduction** on paper
  - Investigate alternative **reducing agents** for nitrate to nitrite conversion
- Enable **quantitative analysis** using a portable **RGB sensor**
  - Optimize **calibration** of the device for consistent colour measurements
  - Increase sample pool to better **correlate colour change** of the sensor to **concentrations** of analyte
- Pivot from lamination to **wax printing** for sensor fabrication
- Develop similar paper sensors for other nutrients involved with algal bloom formation, such as **phosphates** and **calcium**

## Acknowledgements

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## Why Paper Microfluidics



**INEXPENSIVE**  
< \$0.10 per test



**COLOUR BASED**  
Easy to use and understand



**LIGHTWEIGHT**  
Portable, enables on-site use