NITRIC ACID

Hazard Description

Nitric acid (HNO₃) is a colorless liquid with yellow or red fumes and an acrid odor. Nitric acid is toxic and corrosive. Exposure to nitric acid can cause irritation to the eyes, skin, and mucous membrane; it can also cause delayed pulmonary edema, pneumonitis, bronchitis, and dental erosion. Nitric acid is a strong oxidizer. Nitric acid reacts violently with alcohols, alkalis, reducing agents, combustible materials, organic materials, metals, acids, cyanides, terpenes, charcoal, and acetone.

Prior to Use of Nitric Acid

- Complete a research specific laboratory risk assessment.
- Ensure that all required materials are available in case of spill, emergency and waste collection.

Handling

- Always work in a chemical fume hood.
- Keep away from combustibles.
- Avoid contact with metals! Nitric acid is extremely corrosive in the presence of aluminum, copper, and oxides and attacks all base metals.
- For small spills, absorb nitric acid with an inert dry material (universal or acid absorbent), place in an appropriate waste container, and wash remaining area with water.

Emergency Procedures

Always review SDS of purchased product for manufacture specific recommendations. Look at SDS for other modes of exposure.

Contacts	
Emergency: 911	
UW Special Constables: 519-888-4911 or ext. 22222	
	Poison Control: 1-800-268-9017
Whenever 911 is called, if possible, UW Special Constables should also be informed to make them aware of the	
emergency on campus and allow them to support as needed. Ask them to meet the paramedics and direct them to	
the incident location.	
Inhalation	Remove individual from contaminated area
	 Call 911 for transport to hospital
	 Corrosive substances may cause severe lung damage if inhaled
	 Perform CPR and artificial respiration if necessary
Skin Contact	Call 911 for transport to hospital
	 Remove contaminated clothing and quickly but gently wipe material off skin
	 Flush with water
Eye Contact	Call 911 for transport to hospital
	 Flush eyes using eyewash station for a minimum of 15 minutes



Storage

Storage Group OA – Oxidizing Acids

Hazardous Waste

- Must be disposed of as per the <u>University's Hazardous Waste Standard.</u>
- Do not mix nitric acid waste with any other waste streams, including other inorganic acids.
- Use secondary containers (Nalgene/ polypropylene tray or tub) to segregate from other waste and chemical containers.